



OCCASION

This publication has been made available to the public on the occasion of the 50th anniversary of the United Nations Industrial Development Organisation.



DISCLAIMER

This document has been produced without formal United Nations editing. The designations employed and the presentation of the material in this document do not imply the expression of any opinion whatsoever on the part of the Secretariat of the United Nations Industrial Development Organization (UNIDO) concerning the legal status of any country, territory, city or area or of its authorities, or concerning the delimitation of its frontiers or boundaries, or its economic system or degree of development. Designations such as "developed", "industrialized" and "developing" are intended for statistical convenience and do not necessarily express a judgment about the stage reached by a particular country or area in the development process. Mention of firm names or commercial products does not constitute an endorsement by UNIDO.

FAIR USE POLICY

Any part of this publication may be quoted and referenced for educational and research purposes without additional permission from UNIDO. However, those who make use of quoting and referencing this publication are requested to follow the Fair Use Policy of giving due credit to UNIDO.

CONTACT

Please contact <u>publications@unido.org</u> for further information concerning UNIDO publications.

For more information about UNIDO, please visit us at www.unido.org

FINAL REPORT

PROJECT NUMBER: DP/GRE/ 83/005

PROJECT: Techno-Economic Investigation of Froduction

of Deironed Refractory Bauxite

SUECONTRACT: 84/108

AUTHORS: Dr. S. Riederauer, A. Geszti, L. Stocker and

J. Tobler

Table of contents

Chapt	ers an	d Sections	⊥a√e ∵
1.	ixecu	tive Summary	3
2.	Backg	Lound and history of the project	7
	2.1.	lntroduction	7
	2.2.	history of activities connected with the project	9
3.	Evalu	ation of test results	34
4.		et appraisal and production cam for refractory grade bauxites	42
	4.1.	Refractory Grade bauxites	‡2
	4.2.	oses and applications	44
	4.5.	rrices	40
	4.4.	Greek products; possible markets	4 ¹ / ₂
	4.5.	Production program, possible revenues	51
5.	Iron	oxide	54
	5.1.	Introduction	54
	5.2.	The market	59
	5.3.	Revenue from iron oxide sales	Şι
6.	Proce	sses and plant	تر
	ú.1.	aterial balance	59
	6.2.	Processes	69
	6.3.	Equipment list	8 6
	6.4.	Location criteria, utility requirements	97
	6.5.	Layout	104

7. Joss	estimates 1	107
7.1.	Investment cost	lud
7.2.	Cost of operation	119
8. Doone	omic and financial evaluation	129
કે . 1.	Profitability	129
ე.2.	Sensitivity	132
8.3.	Pinancial evaluation	130
8.4.	Sational economic evaluation	144
List of ref	Cerences	1 4 5
Annex I	Test report by Prof. 2. Remecz	

44

1. Executive summary

Technical and economic aspects of beneficiating Greek baux.tes with the objective of producing a high alumina, low iron bauxite grade to be marketed domestically as well as abroad, have been investigated under a subcontract from UNIDO, Vienna for UNDP. New York and the Greek Government.

Current local demand for the refractory grade bauxite product was found to be around 12 thousand tons/yr.

This may increase significantly to enable supply of refractory products to the Soviet Union, and to European and Middle East countries. Current European demand for refractory grade bauxite is in the range of 300,000 tons. Greek plant should be able to sell upwards of 40,000 tons calcined bauxite and its total output of the iron oxide by-product. The market for iron oxide pigments in Europe is in the 100,000 ton range.

Ambient temperature leaching by hydrochloric acid, the most advanced technology currently available for the purpose, has been demonstrated to be suitable to remove over 90 % of the iron in the bauxite feed. Raw material selection is, however, a key factor in the technical and the economic viability of the project, as various alumina and iron oxide bearing minerals have been shown to react differently to leaching by HCl. Also, the refractory industry has some strict requirements on the physical properties of the 'white' calcined bauxite, and these properties are largely letermined by the minerological structure of the raw material.

Although the availability of excellent raw material in Greece has been demonstrated, the government agency promoting the project in Greece had difficulties in obtaining sufficient quantities of these grades from the private sector companies operating the state owned mines. Consequently the pilot scale tests could be run only on a lower grade. A calcined refractory bauxite comparable to the Chinese bauxites and a red iron oxide pigment comparable to Bayferrox 645 were produced and tested in Minnova's Pécs, Hungary and Andritz's Vienna, Austria pilot plant.

For 40,000 ton output of calcined bauxite about 68,000 tons of the material tested in the pilot plant have to be processed, and some 21,000 tons hydrochloric acid have to be added to compensate for chloride loss mainly due to the extremely high Ca+Mg content of this feed. Hain equipment needed include leaching/washing towers, spray roasting /Andritz-Ruthrer/ type acid regeneration equipment and tunnel kiln for calcination. The plant consumes some 11.5 thousand tons fuel oil, 6 million kWh power and 1.6 million m³ process water. Site area required is 40 acres. Most suitable location is close to an alumina plant.

Based on laboratory data estimates appear in the Report for a 40,000 ton output plant on the gray/yellow bauxite of the 3rd horizon in the Parnassus-Kiona-region. In this case 15 % less bauxite, 60 % less hydrochloric acid, 25 % less power, 35 % less fuel oil and 40 % less water are required, and 40,000 tons top grade refractory material comparable to the Guyanese RASC grade can be produced. In this case, however,

about 45 % less iron oxide pigment is gained.

Total investment outlay for a 40,000 ton plant is 3,400 /3,000/* million Drs. Process equipment would cost 2,100 /1,850/ million, know-how and rights to the technologies used cost about 210 /185/ million. The working capital requirement is estimated to about 190 /165/ million Drs.

Manpower requirements for a 40,000 ton plant operating for 8000 hours a year can be covered by 9 university graduates, 23 other engineering and administrative staff and 50 skilled and semi-skilled workers. A plant is erected and installed in about two years. Production start-up takes 6 months, running-in another six.

Total production cost is calculated to be about 1,000 /850/million Drs /excluding depreciation/. Bauxite, HCl and fuel oil account for 31 /42/, 13 /7/ and 28 /21/ per cent of the cost of operation, respectively. Fuel oil price used in the report is about twice the current Western European one.

The operating cost before depreciation is Ers 24,400 /21,175/per ton of calcined bauxite. The product comparable to the Chinese refractory grades could fetch between USD-114 and 155 /Drs 15,400 to 18,200/per ton. The high grade raw material could yield a calcined bauxite matching the market standard Guyanese RASC grade and sell for an ex-works price of Drs 22,400 to 26,600. Iron oxide pigment sold at Drs 28 per kg solids could add about 50 % /15 %/ to the sales revenue of the plant.

^{*}Figures in brackets apply to use of high grade feed.

Profitability of a plant using medium grade feed is not satisfactory. The pay-back period for a 40,000 ton plant is 21 years.

The alternative plant using high grade bauxite feed would benefit from its more favourable minerological and chemical composition through much lower hydrochloric acid losses, lower material flows and higher price for the calcined bauxite, etc. The internal rate of return on the owners' equity is calculated as 10.5 % when no outside loan is used and as 14 % when foreign exchange component of the investment outlay is covered by a Forex loan with 7 % interest.

The project's implementation could result in import substitution, create a new value added export item and provide work for 70 to 100 persons in the country.

For a full bankable feasibility study, there would be required

- i/ a specification of the correct feed material;
- ii/ further large scale tests on the finally specified feed material in order to determine consumption figures and other data:
- iii/ market studies on both the local and foreign markets;
 iv/ the selection of the site.

2. Background and history of the project

2.1. Introduction

Dissolution with hydrochloric acid, with all its advantages, has been known for a long time in hydrometallurgy. Metals are dissolved out of their ores usually faster and with a better efficiency than by the more usually applied sulphuric acid, but hydrochloric acid's higher price and the difficulties in treating the chloride solutions forming as wastes, also the serious corrosion of the equipment have prevented its propagation in practice.

Lately the industrial development has permitted the production of corrosion resistant equipment, including the apparatus of the Ruthner firm for decomposing chloride solutions with a process of spraying calcination. The metal chloride solution is sprinkled in a reactor at 300-900 °C where the following reactions occur:

The solution drops evaporate, the metal chlorides are decomposed and solid metal oxides and gases containing hydrochloric acid are the results. The hydrochloric acid gas is absorbed forming concentrated hydrochloric acid that can be used again for leaching. This process is closed with

respect to the aggressive acid reagent and in comparison with the former processes it does not pollute the environment.

In Hungary the Mecsek Ore Mining Enterprise and the Tatabánya Coal Mines have long been working with hydrometallurgical processes. Utilizing the Ruthner process of acid regeneration, a new process was developed for lowering the iron content of bauxites with hydrochloric acid, known as TATABANYA Process /1/. Some of the features of the TATABANYA Process:

- selective dissolution /90-95 % of the iron can be removed from the bauxite, while only 0,5-7 % of Al_2O_3 is dissolved/;
- relatively coarse grained feed can be used;
- the phases are separated without use of filter:
- very good acid utilization.

The TATABANYA Process had been tested for use on hauxite of different origin /China, Malaysia, Yugoslavia, Hungary, etc./ and type /diasporic, gibbsitic, boemitic, etc./. It was found that diasporic bauxites are best suited for this treatment.

After years of detailed investigation the construction of the first plant to operate on the TATABANYA Processes was started in 1985. The plant located in Pécs, Hungary, has a design capacity of 10.000 tons/year refractory grade bauxite and some 3.000 tons pigment grade iron oxide and is due to start production in 1987.

This report covers the techno-economic aspects of the possible use of the TATABANYA Process in Greece.

2.2. <u>Mistory of activities connected with the project</u> 2.2.1. <u>Initial tests</u>

The first Creek bauxite samples on which the TATABALTA Process was used were received back in 1981/82 through the good services of Mineral Resources and Mine Development 3.A.

These samples were taken from the so-called "3rd bauxite horizon" of the Farnassus-Kiona-Helikon area that holds at least 3 million tons of economically recoverable deposits /2/. The different bauxite grades had been tested in 1981/83 in our Pécs, Hungary, laboratory. The tests included laboratory leaching tests on a few hundred kilo material and subsequent calcination of the samples. The technical results were extremely encouraging. Tables 2/1 and 2/2 present data for sample 3 /a gray-yellow colour lumpy material/ received in 1982. These data have been confirmed in tests by Ideal Refractories, Athens. A first estimate of the economic feasibility made in 1983 indicated that it was worthwhile to pursue the matter.

After UNIDO/UNDP agreed to support further work on the subject, MINMOVA was subcontracted to prepare this technoecomomic feasibility study on the production of deironed refractory bauxite in Greece. Concurrently Dr. E. Mack, expert on mineral resources evaluation & mine development,

with several years of experience in the creek bauxite mining was retained to assist in selecting the suitable bauxite samples for use in the study and to assess the marketing possibilities for the products. Dr. L. Mack produced this report in December 1984 containing many useful data and recommendations, some of which will be used in this report /2/.

Table 2/1
Comparative data on the chemical composition

Composition	1	2
Al ₂ 0 ₃	69.0	91.82
Fe ₂ 0 ₃	9.7	0.86
SiO ₂	2.8	3.85
TiO ₂	2.9	3.28
CaO	1.1	0.21
MgO		
Na ₂ O	0.13	0.13
L.O.I.	14.2	-
	99.83	100.15

^{1 =} Bauxite sample 3/82

Table 2/2
Characteristic data of calcined product of Sample 3/82
from laboratory test made in 1982.

parameter	value
corundum	85 %
mullite	3 %
glass	8 %
water absorbtion	3 %
bulk density	3.45 gr/cm ³

^{2 =} Calcined product from Sample 3/82

2.2.2. Let through some prior to the pilot-scale test

MillOVA Legar work under the contract with UNIDO in movember 1984 by paying visits to mining operations in the said larmassus-hiona-Iti-Helikon area. Two visits have been arranged by HILIC and Ir. L. Mack was invited to participate in them. luring the visits it was made clear to the representatives of the mine operators that we were going to do large scale testing of the came material that was tested earlier, and we asked for their co-operation. We actually saw yellow and tray lauxites deposited in the Itea harbour of one of the operators and received verbal confirmation of the availability of such material in our discussion with the other operator. Dr. ... Hack did not join us for the visit to Bauxite Parmasse Mining Company /EMFC/ hat he participated in the meeting with Hellenic Eauxite mining Company of Helikon /Hb.H/.

In our subsequent meeting with representatives of HIMIC we were told that the grades we worked with previously were not available and we were asked to base our study on another grade which we then agreed to select from what the mine operators promised to make available to HIMIC.

Eack in Hungary, chemical data on eight grades, as well as minerological data on five of them were reviewed.

Leaching and calcination tests were performed in early

1935 on four samples of 50 kg each representing 2 grades

each of the two mine operators. It should be noted that samples received from MEMH were blends of different materials and one of the samples furnished by RMPC satisfied the criteria appearing in Column 1, Table 2/3, also to be found in Appendix 3 of /2/. Laboratory tests confirmed that the various grades show individual behaviour when leached and that the three grain size fractions 0.315 to 1.1 to 5 and 1 to 15 also behave differently. This latter difference was found to be more marked for the sample material received from HEMH.

The 1 to 5 mm grain size fraction of sample D, a grey/
/red material, could be leached to have a 2.46 % Fe₂O₃
content when calcined. Its 0.315 to 1 mm fraction had
a residual iron content of 1.86 % after calcination.
It was, therefore, agreed that -4mm material will be
used in the pilot-scale test. Although sample A performed
similarly, its price was about 50 % higher. /HIMIC information/.

2.2.3. Pilot-sca a leaching

Sample material for the pilot-scale tests was delivered by HIMIC in June 1985 in two shipments of 20 tons and 40 tons, respectively, and it was then made clear to us that it was not possible for HIMIC to provide more. Results of the chemical analysis performed in our laboratory on homogenized samples taken from the two shipments are shown in Table 2/3. Some 10 % of the material received was found to be +4 mm, another 11.2 % of the material was -1 mm. The minerological composition of this material is given in Table 2/4.

. Table 2/3
Chemical composition of different bauxite grades
/all data in w. %/

Component	1	2	3
^{Al} 2 ⁰ 3	min. 62	68.35	56.7 /!/
Fe ₂ 0 ₃	max. 18	13.01	18.6 /!/
SiO ₂	max. 3.5	1.22	2.1
TiO ₂	/max. 3.0/	2.75	2.4
CaO	max. 1.3	0.69	3.0 /!/
MgO	max. 0.2	0.10	2.0 /!/
Na ₂ O+K ₂ O	max. 0.2	0.37	0.1
L.O.I.	/max.15.0/	13.12	14.7

^{1 =} Specification for "guaranteed quality" by UNIDO expert /2/

^{2 =} Established by MINNOVA for blend D

^{3 =} Established by MINNOVA to be the average composition of the 60-ton feed for the pilot-scale test.

Material received by the first shipment was fed into the pilot plant column without delay to compensate for the time spent for material selection. Further quantities were fed into the system subsequently. In view of the large -1 mm fraction, flow problems were anticipated that were thought to result in lower leaching rate, thus aggrevating the product quality problem arising from the presence of +4 mm material. Figure 2/1 shows the pilotscale leaching plant, demonstrated to representatives of UNIDO and HIMIC in July, 1985. Pilot-scale leaching of the 60 t sample, described in detail in the Interim Report /3/, was completed by the end of October, 1985. The first 8 tons produced were calcined at the Budapest facility of Magnezite Works, Budapest. Approximately 6 tons of calcined and some 29 tons of uncalcined, de-ironed bauxite were collected in December, 1985 by HIMIC and shipped to Greece. The leached, uncalcined product was analized chemically, as well as for its minerological composition by both MINNOVA and Société Hellenique de Surveillance S.A., Athens. The results appear in Tables 2/5 and 2/6.

2.2.4. <u>Filot-scale calcination</u>

In the calcination tests bricks were pressed from the deironed bauxite with 6 % moisture, using pressures of 800 to 1.200 bar; drying and calcination were carried out in a tunnel kiln with gas firing, /width: 2.2 m, length: 144 m/. The green bricks were dried for 4 days at 60 to 80 °C. The tunnel kiln had a maximum temperature of 1.660 °C, •the height of the stacks was 1.3 m. The material was fired

1-6 Leaching/washing columns

7-11 Feed tanks of HC1

12-16 Collecting tank for iron chloride liquor

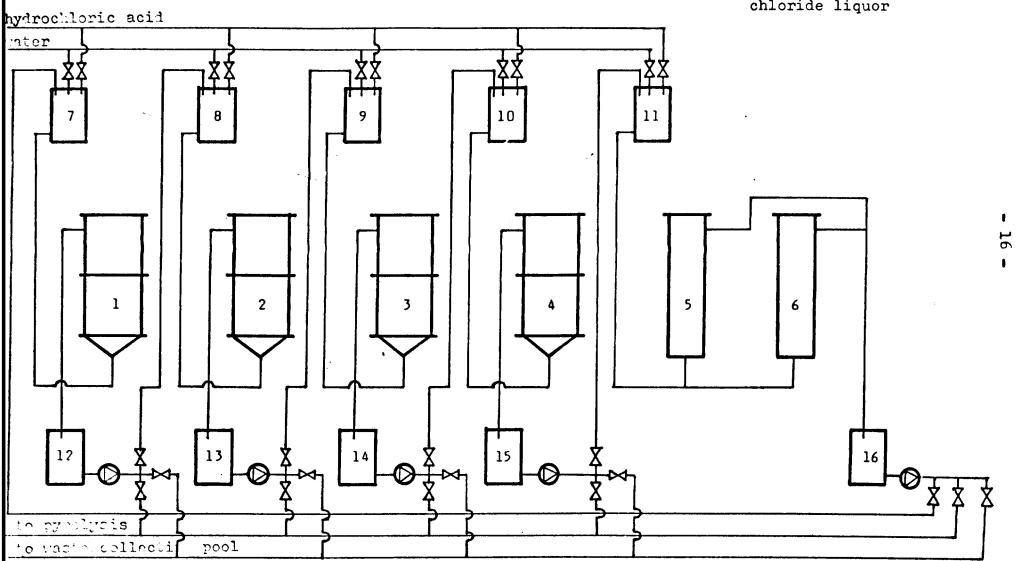


Figure 2/1: Multi-stage countercurrent system of the pilot plant

Table 2/4
Minerological composition
/all data in w. %/

Component	1	2
Diaspore	58 .7	64.0
Boehmite	7.5	-
Hematite	7.0	4.1
Goethite	13.0	18.3
-uartz .	1.5	-
Caolinite	1.0	6.6
Rutile	0.6	0.7
Anatase	1.8	3. 5
Calcite	4.2	-
Corundum	-	2.0
Other minerals and H ₂ O	4.7	1.0
	100.6	100.2

^{1 =} Average value, established by the Metal Research Institue, Budapest, for samples taken from the two bauxite shimpents subsequently processed in MINNOVA pilot plant.

^{2 =} Typical composition of diaspore-type Greek bauxite as it is given by the UNIDO expert in this Technical report /2/.

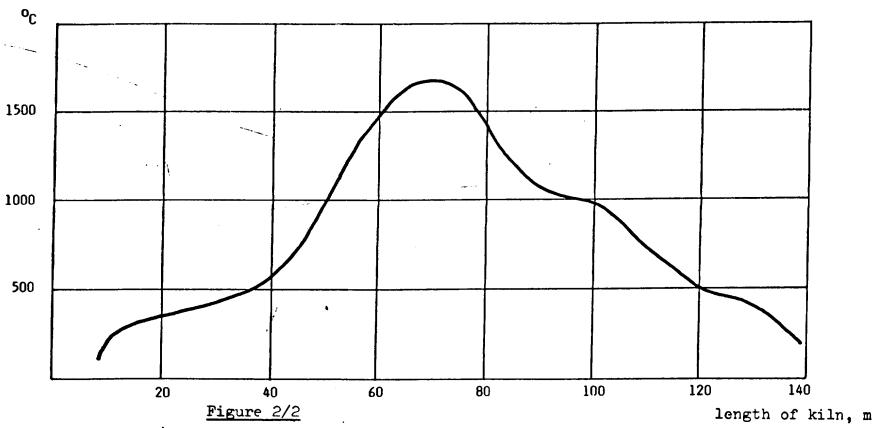
at SK 29 to 30 for 4 hours and spent altogether 100 hours in the kiln. Figure 2/2 shows the temperature distribution of the kiln.

Several samples have been taken from this material. These were subsequently evaluated by the refractory makers MIM, Budapest, and Ideal Refractories, Athens, as well as Surveillance, Athens. Chemical analysis data are presented in Table 2/7.

Minerological analyses of calcined, refractory material made from the pilot-scale test product were also performed both by MIN and Ideal Refractories. These results are given in <u>Table 2/8</u>. Respective results by Surveillance, if any, did not reach us in time to be included here. The calcined, refractory grade bauxite from the pilot-scale test was measured in the laboratories of MIM, Budapest for its physical properties. Results are presented in <u>Table 2/9</u>.

Different compositions have then been made using the dried material from the pilot-scale test, clay, alumina and other ingredients, to adjust the properties of the raw material. Tests with these compositions, data from which are presented in Table 2/9, confirmed that all relevant parameters of the refractory material made from such compositions were acceptable. Microscopic investigations presented in Figures 2/3 and 2/4 confirmed that the relative inhomogenity of calcined material was significantly reduced in the product.

Temperature



Temperature profile of tunnel kiln

Table 2/5 Chemical analysis data /all data in w. %/

Component	1	2	3
A1 ₂ 0 ₃	77.3	75.5	56.7
· Fe ₂ 0 ₃	1.65	1.96	18.6
SiO ₂	2.14	4.75	2.1
TiO ₂	2.81	3.22	2.4
CaO	0.57	0.07	3.0
MgO	x	0.10	2.0
Na ₂ 0+K ₂ 0	x	0.18	0.1
L.O.I.	14.62	14.46	14.7

1 - MINNOVA analysis

f of the de-ironed product
from the pilot-scale test

2 = Surveillance analysis

3 = MINNOVA analysis of the feed of the pilot scale test

Table 2/6
Minerological composition data
/all data in w. %/

Component	in bauxite feed	in de-ironed product
Diaspore	58 .7	77.4
Boehmite	7.5	12.6
Hematite	7.0	1.5
Goethite	13.0	<u>.</u>
_s uartz	1.5	1.5
Caolinite	1.0	1.5
Rutile	0.6	0.7
 Anatase	1.8	2.1
Calcite	4.2	-

Table 2/7

Chemical composition of the product from the pilot-scale test /calcined/.

/All data in w. %/

Component	1	2	3	4
Al ₂ 0 ₃	88.37	89.51	88.0	min. 88
Fe ₂ 03	1.98	2.53	2.28	max. 1.7
SiO ₂	3. 58	3.37	5.53	max. 6.5
TiO2	3.4 9	3.68	3.75	max. 3.5
CaO	0.68	0.10	0.08	max. 0.2
MgO	x	0.20	0.12	max. 0.2
K ₂ 0+Na ₂ 0	0.20	0.12	0.21	max. 0.2

^{1 =} Average from repeated analyses by MIM.

^{2 =} As found by Surveillance /4/.

^{3 =} Calculated from Column 2 of Table 2/5

^{4 =} Preliminary target specification, confirmed by UNIDO expert /2/.

Table 2/8

. Minerological analysis data

/all data in w. %/

Components	1	2	3	4	5	6
Corundum	60	~80	60	7 5	7 5	80-90
Mullite	15	~10	10	0	0	max. 5
T ial ite	10	~ 5	10	10	10	max. 6
Glass phase	15	> 5	20	10-15	15	max. 8

- 1 = As found by MIM, Budapest; /no additive/
- 2 = As found by HIMIC; /no additive/
- 3 = As found by Ideal Refractories /Athens/; /no additive/
- 4 = As found by Ideal Refractories; /sample prepared with catalitic agent/
- 5 = As found by Ideal Refractories; /sample prepared with catalitic agent and ground and sized/
- 6 = Preliminary target specification confirmed by UNIDO expert /2/.

Table 2/9

Comparative data on physical properties chemical and minerological composition

Froperties	1	2
Eulk density /gr/cm ³ /	3.25	3 .3 8
water absorbtion /%/	4.8	2.8
porosity	15.4	9.5
Chemical composition /in w. %/		
Al ₂ 0 ₃	88.37	88.58
Fe ₂ 0 ₃	1.98	1.41
Phases /in w. %/		
Corundum	60 ,	80
Mullite	15	10
Tiolite	10	5
Amorphous	15	5

calcined beneficiated bauxite /as is/

^{2 =} Composition containing 75 % beneficiated bauxite



Figure 2/3
Structure of the calcined refractory grade bauxite
./Magnification factor: 42/

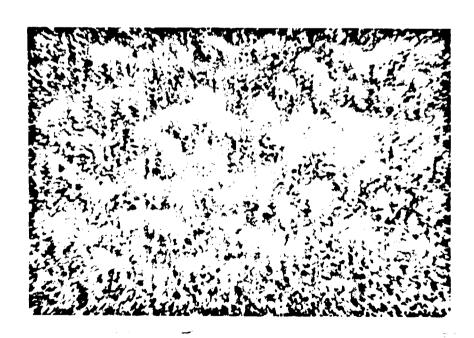


Figure 2/4

Structure of the material made from a composition containing 75 % dried, uncalcined refractory grade bauxite from the pilot scale test and 25 % alumina, ground, fired at 1.630 °C and screened to -5 mm./Magnification factor: 42/

A brick made from composition named G-2 was subjected to a slag corrosion test according to the method described in Hungarian Standard 5917-83. Slag composition characteristic for Siemens-Martin furnaces with a basicity of 2.1 was used in test conducted at 1.580 Celsius. Penetration was similar to that established for other high-alumina refractory products.

2.2.5. Regeneration/Iron oxide

In the mean time two batches of 5 m³ each spent liquor had been processed at Vienna facility of Maschinenfabrik Andritz, Ruthner Division. The chlorine content of the spent liquor had been successfully converted to hydrochloric acid and between 200 kg and 660 kg iron oxide /as shown in Table 2/10/ were collected for each batch in the reactor and in the cyclone of the regeneration plant, respectively. A discussion of the process was held with representatives of UNIDO and HIMIC. HIMIC was subsequently given 50 kg samples of the two reactor oxide grades produced. A detailed description of the test and its results are to be found in the Interim Report.

A sample of the reactor oxide produced from the unpurified liquor has been treated in MINNOVA laboratory in Tatabénya to free it from water soluble chloride residues that did not dissociate at pyrolisis temperature /CaCl₂ and others/. For this purpose lime milk was used and the resulting product was washed repeatedly with water.

Table 2/10

Chemical composition of iron oxide samples

/average values in %/

Component	1	2	3	4
Fe ₂ 0 ₃	97	94.8	70.3	68.3
FeCl ₃	2.6	6.8	-	-
CaCl ₂	-	-	17.1	16.6
sio ₂	0.4	0.4	-	-
Al ₂ 0 ₃	-	-	3. 8	3. 9
MgO	-	-	8.2	7.6
other chlorides	-	-	0.6	3. 6

- 1 = Reactor oxide from the purified liquor
- 2 = Cyclone oxide from the purified liquor
- 3 = Reactor oxide from the unpurified liquor
- 4 = Cyclone oxide from the unpurified liquor

A one-liter sample of a suspension containing 37.4 % solids, appropriately conditioned, was examined in the laboratory of a major paint maker for its suitability as a pigment for use in /a/ concrete products, /b/ in dispersion paints and /c/ in alhyd-type paints. The examinations were conducted according to the Hungarian Standards 10962 / "Man-made iron oxide red pigment"/, 21191 / "Pigments based on iron oxide"/ and 21189 / "Figments for the cement & concrete industry/. The results also were compared to the properties to those of the industry standard products of Bayer AG, Leverkusen. Samples of pigmented concrete and of paints formulated were prepared, a set of which is being attached to HIMIC's master copy of this report. The results of the examinations are shown in Table 2/11.

2.2.6. Additional tests on possibilities to improve refractory properties

In May 1986 the test results then available have been discussed in Athens, where representatives of UNDP, UNIDO, HIMIC, Ideal Refractories and MINNOVA participated. The general feeling was that the raw material used for the pilot-scale test was not optimal and as a consequence the properties of beneficiated product were not up to the highest standard. Upon HIMIC's special request, MINNOVA agreed to perform additional laboratory tests on two more bauxite samples. MINNOVA representatives at the meeting asked that the type of bauxite called "grey//yellow" by the UNIDO expert and described on page 18 of his report be provided by HIMIC for these tests.

Table 2/11

Properties of iron oxide extracted from 60 t sample in the pilot plant beneficiation test /after treatment/

Properties	1	2	3
Iron oxide content /%/	-	82	86-88
SiO ₂ + Al ₂ O ₃	-	4.5	6
Sieve residue /%/	0.3	0.025	0.05
on 45/um mesh	0.3	0.025	0.05
on 125 um mesh	×	0.005	¥
pH of the suspension	6 - 8	6	6
water soluble salts	1.5	1.2	0.4
organic content	none permitted	none	none
"blooming"	none permitted	none	none
alkali resistance	no colour change	no colour change	no colour change
colouring power	į	•	·
/ratio of pigment to cement for same effect/	2-5	4.5	3
compatibility with acrilate dispersion	-	no	yes
compatibility with alhyde resin		yes	yes

^{1 =} Values specified in relevant Hungarian Standard

^{2 =} Sample

^{3 =} Bayferrox brand product

The samples - 35 kg each - were received early July, 1986. The sample 457/1 had a pale yellow/brown colour, sample 457/2 had a fuller hue of the same colour. Both samples turned out to be blends of different grades. The samples lost their yellow hue when washed in water. The chemical composition of the two samples is shown in Table 2/12, the minerological one in Tables 2/13 and 2/14.

Both samples were then leached in laboratory glass columns for 96 hours at 80 °C by using a hydrochloric acid concentration of 438 gr/l in order to establish the limiting residual iron oxide content. The results are shown in <u>Tables</u> 2/12, 2/13 and 2/14.

Calcination of the samples was deemed unnecessary as properties superior to those of the beneficiated product from the pilot-scale test could not be expected.

In another attempt to improve the properties, beneficia+ed Greek bauxite from the leaching/washing stage was subjected to magnetic separation in high intensity magnetic separator. Field strength values between 10.000 and 15.000 gauss have been used and the material was passed through the system repeatedly in order to get a first indication of the technical viability of such a treatment of the sample. Table 2/15 shows the results. It should be noted that magnetic separation tests done on Guyana refractory grade bauxite in the United States, as well as similar investigations conducted in Brazil have been reported in the literature /4/.

Lable 2/12
Chemical composition of samples 457/1 and 457/2 before /1/
and after /2/ leaching

Component	457/1		457/2	
	1	2	1	2
Al ₂ O ₃	6 7.7 5	78.2 5	63.55	78.2
JiG ₂	1.49	0.97	2 .2 3	0.99
l'e ₂ 0 ₃	9.09	2.78	15.52	2.65
1102	3.18	2. 96	2.95	2.95
CaC	1.15	0.07	1.26	-
kiro O	×	0.02	×	0.01
K ₂ 0 + La ₂ 0	×	0.02	¥	0.02
30 ₋	×	0.13	¥	0.13
L.C.1.	14.78	14.79	13. 95	15.06

Table 2/13

Minerological composition of 457/1, before /1/ and after /2/ leaching test

/all data in w. %/

Component	1	2
Diaspore	78	84
Boehmite	6	10
Anatase	. 2	2
Rutile	1	1
Hematite	8	, 3
Goethite	4	-
Calcite	1	-

Table 2/14

Minerological composition of 457/2, before /1/ and after /2/

leaching test

/all data in w. %/

Component	'n	2
Diaspore	56	83
B oehmite	, 20	11
Anatase	2	2
R utile	1	1
Hematite	14	3
Goethite /	5	, -
Calcite	2	•

Table 2/15
Results of a laboratory magnetic separation test
/All data in w. %/

Material	1	2	3
Initial sample paramagnetic fraction diamagnetic fraction	100	1.8	100
	6.87	2.7	10.5
	93.13	1.73	89.5
Diamagnetic fraction from the first run paramagnetic fr. diamagnetic fr.	3.82	2.5	5.61
	89.3	1.6	83.88

^{1 =} Relative amount of fraction /initial sample = 100 %/

 $^{2 =} Fe_2O_3$ content of fraction

^{3 =} Relative amount of $Fe_2^0_3/Fe_2^0_3$ content of initial sample = 100 %/

3. Evaluation of the test results

Most hydrometallurgical processes are made cumbersome and costly by the following technological stages:

- the grinding of the raw materials
- the provision of the higher than ambient temperature required for the chemical reactions
- the mixing and the separation by filtration of the fluid and solid phases
- the evaporation of the so arising relatively dilute solutions and its power consumption.

The technology jointly developed by Tatabánya Coal Mines, Mecsek Ore Mines, both of Hungary and Andritz-Ruthner Industrie-anlagen AG, of Austria, for reducing the iron content of bauxites by hydrochlorid acid treatment is a hydrometallurgical process without the above stages. At ambient temperature, namely, concentrated hydrochloric acid does not attack the aluminium minerals and it dissolves iron from bauxite selectively. By keeping the flow of the solution at a slow rate the mixing of the fluid and solid phases and the production of a slurry can be prevented while the proper permeability of the bauxite bed can be insured by the correct determination of the upper and lower particle sizes of the bauxite. To ensure a possible maximum iron concentration in the iron chloride solution arising during digestion, a countercurrent leaching process has been developed.

To meet the above criteria the maximum and minimum particle sizes, the rate of solvent flow, the number of the acid treating units connected in countercurrent and the duration of the acid treatment best suited for the process must be determined separately for each type of bauxite.

The efficiency of the chemical processes taking place during the acid treatment and the residual iron content in the product following it depends not only on the above outlined requirements but also on the mineralogical composition of the starting material.

To establish the relationships valid for the diasporic Greek bauxite examinations were launched in 1983 at the Department of Mineralogy of the Chemical University of Veszprém under the guidance of Prof. E. Nemecz. The report drawn up on their result is enclosed as Annex I of this Report.

Tests done prior to and under the UNIDO Sub-Contract provided some information regarding the type of correlation that exists between the mineral composition of the bauxite feed and the conditions of leaching.

Tables 3/1 and 3/2 show the minerological composition of 9 bauxites received from Greece, their initial and eventual iron oxide content as well as some computed parameters.

Although the data is not sufficient for a full statistical correlation the following observations can be made:

Table 3/1

Initial minerological composition and iron oxide contents

of Greek bauxites tested in 1982

Samples	Minerological composition	Iron oxide content /%		
		initial	in in calcined product	
1/82	much goethite, little hematite	16.7	1.3	
2/82	little goethite, medium hematite	13.4	1.6	
3/82	little goethite, medium hematite	8.5	2.0	
4/82	little goethite, much hematite	13.7	2.2	

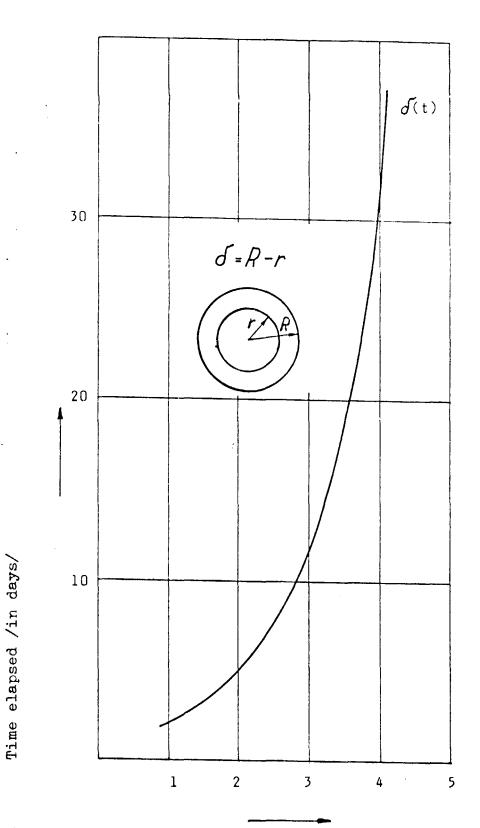
Table 3/2
The effect of initial minerological composition on the iron oxide content of the calcined beneficiated bauxite

		Sample						
	п¥п	иBи	60	475/1	475/2			
Hematite, %	11	. 22	7 /1.5/	8 /3/	14 /3/			
Goethite, %	2	2	13 /-/	4 /-/	5 /-/			
Initial Fe ₂ 0 ₃ , %	12.2	23	17.8	9.1	15.5			
Fe ₂ 0 ₃ in calcined product, %	2.45	3.54	2.5	2.8	2.7			
Diaspore, %	73	30	59	78	56			
Boehmite, %	4	35	8	6	20			
Diaspore Boehmite	18.25	0.86	7.4	13	2.8			
Hematite . 100, %	90.2	95.7	39.3	87.9	90.3			
initial Fe ₂ 0 ₃		7741						
Nematite, %	87.5	90.9	80	80.4	89.2			

- 1. Iron oxide remaining in the calcined refractory grade bauxite is not proportional to the initial iron oxide contents of the bauxite feed.
- 2. Goethite type iron oxides can fully dissolved with the Tatabánya process from diasporic bauxites.
- 3. Hematite type iron oxides can be removed with the Tatabánya process with an 80-91 % efficiency from diasporic bauxites.
- 4. The ratio of diaspore to boehmite does not appear to have an effect on the solubility of hematite.
- 5. The iron oxide content of the calcined beneficiated bauxite apparently also lepends on factors other than minerological composition.

The rate of acid treatment depends on the diffusive processes taking place in the particles. A lower concentration gradient reduces the rate of diffusion, with increasing particle size the time required for the reduction of iron grows exponentially.

If the particles of the Greek bauxite are 4 mm the acid treatment of the charge in one column takes 30 days. It has been established that a single mm larger particle size would increase beneficiation time by nearly 100 per cent while if the maximum particle size is, say, 3.15 mm, the beneficiation of the bauxite examined would require a shorter period of time /Figure 3/1/. It should however be remembered that the smaller the particle size to which the raw bauxite is ground, the more will be the dusty fraction, unsuited



Depth of leached zone; mm

Figure 3/1 leaching time vs. grain size sample D

for the intended process. Besides, the reduction of the particle size would increase the danger of disturbed flow, canal formation and disturbances in the permeability.

The formation of canals would result in uneven contact between the bauxite and the acid.

The smallest particle size and the pertaining maximum flow rate must be established by running tests. In the course of pilot tests the flow rate was determined at 7.5 cm/h /this is the rate at which the liquid level rises in an empty column/. The permissible smallest particle size may be somewhere between 0.5 and 1.0 mm.

The pilot scale test has shown that the hard diasporic structure of the Greek bauxite is one of its freat assets, i.e. it is not prone to fragmentation by the acid. Accordingly the probability of disturbances of flow is much lower than with gibbsitic bauxites.

The acid treatment cycle using five columns applied in the large-scale laboratory experiments was chosen on the basis of experience gained with Hungarian bauxites, but the higher stability of the Greek bauxites allows higher rates in washing and favourable modifications of the technological cycle.

According to the modified technology acid treatment cycles can be completed in 40 day with only four columns: three times ten days in each cycle for the acid treatment which leaves ten days for washing and discharge. This modification reduces the specific investment cost with Greek bauxite.

As regards further processing, the Greek bauxite offers a further advantage in that after washing, it needs no filtration. After draining via the bottom of the column the moisture content of the bauxite will be as low as 15 to 20 % /30 to 35 % with the Hungarian bauxite/.

From the diasporic bauxite examined, under a pressure of approximately 1000 bar, bricks can be produced, without bonding materials, by dry pressing or semi-dry pressing and calcined in tunnel kilms.

The calcined refractory grade bauxite made in the pilot scale test was found to be of medium quality, comparable to Chinese Grade I and Grade II bauxites. Its properties do not match those of the gray bauxite from the 3rd horizon of the "Parnassus-Kiona-Helikon" region tested in laboratory.

The by-product from the pilot scale beneficiation of bauxite could be turned into a pigment grade iron oxide comparable to Layferrox grade products.

4. Market appraisal and production programme for refractory grade bauxites

In this chapter comprehensive information on the market for refractory grade bauxites given by the UNILO Expert in his Report /2/ will be complemented and a production programme will be proposed for products based on Greek raw material.

4.1. Refractory grade bauxites

The bauxite world production is in the range of 80 to 90 million tons per year; nearly 95 % of all bauxite mined is converted to alumina by the layer process and only about one and a half percent is sold as refractory grade bauxite. The market leader is Guymin /Guayana/ with a 1985 production capacity of 950 thousand tons calcined bauxite, actually selling about 600 thousand tons refractory grade and about 100.000 tons abrasive grade waterial. China's export grows at a rapid pace and its sales are reportedly approaching 0.5 million tons a year, pard of this being calcined in Massa, near Livorno in Italy. Surinam, Malaysia and the USA are producing between 50.000 and 100.000 tons refractory grade bauxite each. India and Brazil are also producers, mostly for the domestic market. South Africa is selling a comparable material called Andalusite. Australia does not sell refractory grade bauxite yet, but a part of its calcined abrasive grade material /annual production in 1980=285.000 tons/ is processed into brown fused alumina and sold to refractory makers as high alumina, low iron material.

Refractory grade bauxite is hormally offered as a calcined

product. It typically has low iron and low silica content. The industry standard, against which the others are judged is the so-called Refractory Aggregate Supercalcined /RASC/, available from Guayana through the U.S. Company Philip Brothers. This product is specified to have max. 2.5 % Fe₂O₃, max. 7.5 % SiO₂, max. 3.5 % TiO₂ and min. 86,5 % Al₂O₃ /the latter often nearer to 90 %/. /5/

The Shanghai International Trading Co. handling the export of some of the Chinese refractory grade calcined backites only specifies Al₂O₃ and iron oxide contents. The latter is max. 2 % the former is min. 85 % for 'Super Grade' and min. 80 % for 'better Grade'. /6/

Minmetals also sells Chinese refractory grade bauxite. Two grades are offered: Grade I and Grade II. Grade I typically has 87.50 % Al₂O₃, 6 % SiO₂, 1.50 % Fe₂O₃, 3.70 % TiO₂, 0.20 loss on ignition and a bulk specific gravity of 3.10 g/cm³. Grade II has 84.50 % Al₂O₃, 6.50 % SiO₂, 1.50 % Fe₂O₃, 4 % TiO₂, 0.20 % loss on ignition and a BSG of 2.80 g/cm³ /5/. Most producers offer lower quality grades as well - Guymine has a 84 % min. Al₂O₃, BSG 3.0 material, the Chinese are selling "calcined diaspores" /having 50 to 70 % Al₂O₃/ /6/.

These materials are often treated before calcination. Chinese and Indian refractory bauxites are handselected, and partially calcined. Guayanese bauxite is washed, dried and screened. None of the products available on the market is known to have undergone hydrometallurgical beneficiation, though.

Technical development in the market is going in the direction of increased Al_2O_3 content /up to 93 %/, decreased Al_2O_3 /as low as 3.25 %/, Al_2O_3 /as low as 2.5 %/ and Al_2O_3 contents /as low as 1.25 %/. Guymine, the state owned producer of the RASC grade aims at a specification of ESC min. 3.6 and max. 5 % porosity and is reported to be working on a flotation method to reach this. /7/

4.2. Uses and applications

The main use for material called refractory grade bauxite is in the production of high alumina refractories. These represent a growing portion of the refractory materials used, as alumino-silicate based products lose ground due to inferior refractoriness and strength. Refractory grade bauxite's use, however, is not limited to refractory industry: it has other applications as anti-skid surfacing of roads and floors, is used in the manufacture of high strength porcelain as well as in coatings of welding rods, etc. The uncalcined low-iron bauxite has various additional applications, when processed into aluminum sulphate it is widely used in water purification and in paper manufacture. As thermally activated bauxite, it is sold in the USA as an absorbant for desulphurisation, decolorisation and as drying agent in food and chemical processing industries.

As the refractory use component of the demand decreases with technical process in steel making /specific consumption of refractories was reduced in W-Germany from 26 kg in 1960

to 15 kg in 1980 for a ton of steel/ and negative growth of the iron and steel industry in many parts of the world, and non-refractory uses of the same grade stagnate or grow/e.g. in water purification/ it is hard to say whether more or less high alumina, low iron bauxite can be sold in 5 or 10 years time.

4.3. Frices

Trice information appearing in Industrial Minerals, a monthly published by Metal Bulletin Journals, Ltd. in London provides a periodical guide to the prices obtained by producers and dealers. In the past 5 years the three major grades on the market have been quoted as appears in <u>Table</u> 4/1.

The RASC grade from Guayana have suffered some price loss in the early 1980's, but as the supply situation normalized, its price has not changed much in dollar terms. /European users able to buy full shipleads are of course paying up to 30 % less in their own currency now as the dollar fell./ Chinese producers could continuously improve the quality of their product and this is reflected in the prices, too. The present average prices of £ 85 and 77 are the equivalent of USD 127.5 and 115.5, respectively, which says that the RASC is still considered to be superior to the Chinese Grades. This difference is even more pronounced than it may seem since the price for the Guyanese material is quoted F.O.B. Baltimore and there is additional freight of USD 35-60/ton plus to be paid by the European user. The authors

Table 4/1

Prices of refractory grade bouxites

/Source: Industrial Minerals, London/

	Chinese bau ite Crade I /typical 85 % Al ₂ O ₃ / CIF Europea: port /L/t/	Chinese bauxite Grade II /typical 80 % Al ₂ O ₃ / CIF European port /L/t/	Guayana bauxite RASC FOB Baltimore /\$/t/
August, 1981	76	68 ·	204.16
January, 1982	73	62.5	214.53
September, 1982	73	62.5	200.95
February, 1983	- 73	62.5	172.23
December, 1984	78 - 80	71 - 73	164.28
February, 1985	78-80	71-73	164.28
July, 1985	84 – 86	7 6 - 78	164.28
March, 1986	84 - 86	76 - 78	164.28

are not aware of a numerical formula that could be used to weigh the influence of the various technical parameters on the price of this material. It is apparent, though, that Al_2O_3 and Fe_2O_3 contents are just two of the many factors effecting the price, one of the reasons being that the different grades are not necessarily substitutable for each other.

Actual contract prices are effected, among other things, by the customers' ability to by a full shipload on the one hand and on the grain size, delivery times, and vendor's reliability record, etc. Ground and sized grain fractions would typically sell at a 15 to 20 pct. premium over the standard RASC price. The cost to the refractory maker may include freight, customs dut, handling charges, commissions, etc. Of these freight, especially if by rail, can be quite significant. /Users in Central Europe are paying up to DM 100/ton for freight./

A comparison of the prices of refractory grade bauxite with abrasive grade /min. 86 % Al₂O₃, permitted to have up to 5 % each of iron oxide and titania/ shows this latter grade to have stable prices of about 20 % less than the Chinese Grade II. Industrial Minerals has been quoting a price range between £ 57 and 66/ton for the past 24 months /USD 85.5-99/ton/ for abrasive grade calcined bauxite, CIF European port.

Although refractory grade calcined bauxite prices have been fairly stable recently and this is what this report assumes for the life of the project as well, there is a potential for changes: Political disturbances can result in a shortage of supply as experienced towards the end of the 1970's, an increase in the energy prices /at their lowest level for 15 years in real terms at the time of writing/ could push up cost of calcination, etc.

As many refractory makers are developing new products and continuously try to cut costs to make existing ones there is plenty of opportunity to exploit the individual benefits of a new material.

4.4. The Greek products: possible markets

This report covers experimental work conducted on several Greek bauxite grades. In the authors opinion, which is in full agreement with the recommendation given by the UNILO Expert in his Report /2/, the objective of a project to beneficiate Greek bauxite should be to produce a product quality that is second to note in the market. This is definitely not the case with the product from the pilot scale test, although this material <u>is</u> a high alumina, low iron bauxite.

As a laboratory work showed /Table 2/1 and Table 2/2/
sample 3/82 could be processed into a calcined product
that is superior to RASC grade. As a result of the beneficiation, it would have a higher degree of reproducibility and
homogenity. There are indications from further tests
conducted by third parties, that this material could yield
an excellent quality brown fissed alumina, sold normally as
abrasive as well as raw material for high alumina refractories. Further work could expose additional benefits of the
material over competition, as well as further applications
possibilities.

In Greece the beneficiated bauxite could substitute imports now running in the range of 10-15 thousand tons a year. It could be used also in Greece to produce high quality refractory products for export in the Middle East, in the Soviet Union and other countries. While the Soviet Union is

the biggest steelmaker in the world and it has its own refractory producers, Greek-Joviet trade relations may prove close enough to provide opportunity for the development of this business. The fact that the Soviet Union is playing a key role in the establishment of the new Greek alumina plant is probably important, as large Soviet exports are due that will have to be balanced by Greece.

Steel making is at a relatively low output level in the Middle East now. Current production is reportedly about 5 million tons a year. There is apparently no local refractory production in the region. As detailed statistics were not available to the authors, consumption of refractory grade bauxite and equivalents has been estimated as follows:

Relating EEC's imports of some 300.000 tons refractory grade calcined bauxite and its steel production of roughly 125 million tons to the steel production in the Middle East, the latter's consumption is put in the 12 to 15 thousand ton range.

It should also be possible to export the refractory grade calcined bauxite. In this case markets would include the Soviet Union. European countries. etc.

4.5 Production programme, revenues

The performance of a beneficiation plant producing refractory grade bauxite will depend to a large extent on its raw material, as demonstrated in chapters 2 and 7 of this report. The use of inferior raw material will result in lower product quality and may also increase production costs. As input material varies over time, one approach can be to blend the raw material and/or the product to ensure consistent quality for the latter. Another way can be to produce more than one final product grade. It may be practical from the marketing point of view also, to have several product grades at different prices, for different customers. As treatment time increases quasi-exponentially with decrease in residual ironoxide content and leaching time is an important element in the cost of operation, as described in Section 7.2, it may be prohibitive for cost reasons to keep processing an inferior raw material for an extended period of time.

The price of the beneficiated material will be affected by the cost to make it as well as by the supply/demand situation. The extent to which competing materials be substituted and vica versa will be important, too. Even in the hypothetical case of a full two-way substitutability, however, there would be differences in the prices obtained for the two equivalent materials. In this report the assumption has been made that the top quality material characterized in Tables 2/1 and 2/2 could be sold for net ex-plant price in the range of USD 165 to 100 /Drs. 22,000 to 26,600/, this range being centered on 90% of the CIF price of the RASC grade, lumps, loose /1 USD = Drs 135/.

The medium grade produced in the pilot plant and characterized in Tables 2/7, 2/8 and 2/9 could be used as high alumina, low iron refractory material, too. Frice of this material has been related in this report to that of the Chinese refractory bauxites. The net ex-plant selling price range for this grade has been selected as between Drs 15,400 and 18,200/ton /USD 114-135/, centered on the average of prices of the Chinese Grade I and II materials.

Revenue estimates in <u>Table 4/2</u> for plant capacities ranging from 20,000 to 80,000 tons have been made for the two extreme cases of producing 100 % top grade or 100 % medium grade product, taking both low and high limits of the price ranges into account. Price premium possible for ground material sold in bags is not considered in the Table, but a 6 percent export subsidy is.

Table 4/2

Export¹ revenue from bauxite sales
/in *000,000 Drs/

Feed grade		85		- 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1	83
Output in	201	40	60	30	
Revenue at 1. USD 114 per ton /Drs 15,400 + 6 %/	323	653	979	1,306	-
2. USD 165 per ton /Brs 22,400 + 6 %/				_	95C
3. USD 135 per ton /Drs 18,200 + 8 %/	38 ₀	7 72	1,158	1,543	_
4. USD 197 per ton /Frs 26, CO + 6 %/	-	_	-	-	1,128

O Export subsidy: 6 %

5. Iron oxide

5.1. Introduction

In the TATABANYA process refractory grade bauxite is produced by removing a major part of the initial iron oxide content of the feedstock. The iron chemically bound in iron chloride at the end of the leaching stage is turned into iron oxide again in the acid regeneration stage. This iron oxide has a various impurities depending on the composition of the liquor, on the roasting temperature, and on other factors. It is apparent from Table 2/10 in Chapter 2 that there is a substantial amount of impurities in the iron oxide produced from the intreated liquors of the leaching/washing stage. Impurities could be significantly reduced by intermediate ion-exchange purification, described in detail in the Interim Report. This however is a costly process. The alternative way, that is to purify iron oxide afterwards, is much cheaper. s Table 2/11 in Chapter 2 shows this material is similar to what is used as pigment for colouring concrete and some other building materials. Respective laboratory tests described in Chapter 2 confirmed this and indicated further that it may be possible to use the purified and processed re enerated iron oxide in alkyd resin paint systems. Based on consultations with RTR-Ruthner GmbH the authors feel, however that it will not be economically feasible to process iron oxide from Greek bauxite for use in the Cerrite industry, as proposed by the UNIDO expert in his Report /2/, purity requirements there being more strict.

5.2. The market

Iron oxide is the second largest colour in the world, after titanium dioxide; its market share exceeds that of all other coloured pigments combined. Present estimated world market is 670.000 tons, about 50 % of which is supplied by Eayer of Leverkusen, Cermany and its subsidiaries. Various chemical processes are used to supply about 85 % of the market; the balance is natural iron oxides. The most important use /60 % worldwide/ is to colour concrete. /Coloured mortar, concrete paving stones, concrete roof tiles, etc./ Coating applications /29 % wooldwide/ including anticorrosion paint, latex, etc. as well as use in the plastic. /trash bags/ paper and other industries should also be mentioned. Iron oxide red accounts for approximately 43 % of the US iron oxide consumption, with yellow, black and brown following /8/. The authors estimate that the Greek market for regenerated iron oxide pi ment could be developped to reach 5.000 tons a year.

Iron oxide pigments contribute little to the cost of the coloured product, as they are relatively cheap and are used in low percentages /Bayer recommends the use of 3 parts pigment for 100 parts cement for concrete colouring applications/. Manufacturers normally offer a range of shades in red, brown, yellow and black. The prices of the Bayer products are up to 30 % above those of the competition. Bayererox Brown 645 T, a grade similar in shade to the pigment made from the material used in the pilot scale test, is currently sold in Austria for 3S 1988 /USD 136/ - per 100 kg

/minimum lot size 500 kg/, ex warehouse in Jienna. The export quotation price for the same grade for a 1000 tons/year has been stable for the past years at around IM 2000/ton/USD 1000/ton/ FCO Furth/Germany. Red and yellow grades are sold at within 10 % of this price, black is about 35 % cheaper. An operator of Ruthmer spray-roasting equipment in W-Germany is obtaining between IM 500 and 800/ton iron oxide /USD .25 - .50/kg/, which however requires further conditioning when used as a bigment.

5.3. Revenue from iron oxide sales

To estimate the revenue from iron oxide sales it is assumed that iron oxide produced in the bauxite beneficiation plant will be processed to pigment grade using a proprietary technology. Considering the fine granulometry of the spray roasted iron oxide, this is best done within the plant. For simplicity's sake, one grade only is considered, although it should be possible to produce different shades of brown and possibly red.

Table 5/1 shows the revenues for various plant sizes based on the assumption that the total amount is sold at one price. Quantities used in this table have been taken from the material balance in Section 6.1. and from similar balances not appearing in the Report. In order to make a sensitivity analysis possible, a price range has been used. This is centered on Drs 21/kg/USD .15/kg/, and covers the Drs 14/USD .10/ to Drs 28/USD .20/ range. The low price is proposed to attract sufficient interest for the product and to leave room for cost of promotion and distribution.

Table 5/1

Revenue from iron oxide pigment sales /in million Drs/

Saumite Feed made		63			
.event 11ant cizo	20	40	60	୫୦	40
ri tisht produce /*COO tols boliss/	3 5.5	10.95	16.4	21.9	5.7
to at tow.PD /wrs 14/ per he solids	77	153	230	307	3 0
.2. at 180.20 //ars 23/ pac its solius	154	303	400	6 17	leō
in these 700 thank sold in theece and balance alread /1/, at the 28 per kg solids	(161	324	436	64 8	171

/1/
| Lmport Subsidy: 6 %

6. Processes and Plant

In this chapter technical and technological information is given on the Tatabánya process and on a plant with an annual output of 40.000 tons of calcined refractory grade bauxite, based on the use of the same raw material as in the pilot plant test. Estimated material flows and balances appear also for the high grade raw material /83/known from earlier testing for the sake of comparison as well as to make the financial and business evaluation in subsequent parts of this Report possible.

Using the findings from laboratory and pilot plant tests, detailed information is provided on the material flows of the plant, on the individual processes and technological steps including pollution control and environmental protection measures. Basic engineering data is presented for the equipment and machinery needed. Location criteria are reviewed briefly and a layout is proposed.

Data from this chapter together with data on other plant variants generated similarly is used in subsequent part of this Report for purposes of financial and economic evaluation.

6.1. Material balances

Material balances for the medium rade raw material are lased on the analysis appearing in Column 3 of Table 2/5 for the feed, in Column 2 of Table 2/7 for the bauxite product and in Table 2/10 for the iron oxide product.

The relevant chemical reactions taking place in the leaching step are as follows:

^{Al} 2 ⁰ 3	+ 6 NCl	2 A1C1 ₃	+ 3	H ₂ 0
Fe ₂ 0 ₃	+ 6 hCl	2 FeCl ₃	+ 3	1120
TiO ₂	+ 4 HCl	TiCl ₄	+ 2	H ₂ 0
CaO	+ 2 HO1	CaCl ₂	+	11 ₂ 0
Hg0	+ 2 HC1	MgCl ₂	+	H ₂ 0

elements of the medium-frace and high grade bauxite feeds between the 'white' bauxite product and the iron oxide product, indicating also the yields of the above chemical reactions for the medium grade material. Results for the high material are for orie, tation only.

The full material balance lased on the pilot scale test of the medium grade material appears in Table 6/3.

A Shanky diagram of material flows connected with bauxite appear in <u>Table 6/4</u>.

In <u>Figure 6/1</u> material flows for medium grade /85/ raw material are presented in a block diagram.

Table 6/1
Distribution of elements of the medium grade feed between the products

	blemen in the			ents salcined te /2/			Elements in the iron oxid product	
	%	ton	%	TON	ton	%	ton	%
Al ₂ 0 ₃	54.0	36,616	89.51	33,804	812	2.2	597 /4/	5.1
Fe ₂ 0 ₃	17.8	12,070	2.53	1,012	11,058	91.6	11,058	94.1
SiO ₂	2.0	1,356	3.37	1,348	8	0.6	8	0.07
TiO ₂	2.3	1,560	3. (8	1,472	88	5.6	88	0.7
CaO	2.9	1,966	0.1	4Ü	1,920	(98.0	- /4/	
Ози	1.9	1,238	0.2	80	1,208	93.8	- /4/	
K20+Na20	0.1	68	0.12	48	20	29.4	- /4/	-
moisture	4.9	3,523	-	-	3,3233	-	-	-
loss on ign.	14.1	9,561	0.49	1 96	9 , 36 9 /3/		-	
Total	100,0	67 , 308	100,0	40,000	27, 803		11,751	99.97

Notes: /1/ Taken from Table 2/5 Column 3 and adjusted for 4.9 % moisture

/2/ Taken from an Analysis by Surveillance, Athens

/3/ Evaporates during drying and calcination

/4/ Missing material discharged as waste gas or effluent

<u>lable 5/2</u> Distribution of the elements of the high grade feed between the products

	lleme in the	nts /1/ feed	Mements in the /2/ calcined bauxite			
	%	ton	%	lon	,3	ton
Al ₂ O ₃	65 .7 8	38 , 024	91.82	56 , 6 73	11.3	712 /4/
Fe ₂ 0 ₃	9.25	5,345	0.86	343	33.3	5,002
SiO ₂	2.67	1,543	3. 86	1,538	0.1	5
TiO ₂	2.76	1,593	3. 28	1,310	:.8	288
de⊅≨Ω0	2.00	*A *	0,1,	O.A.		- /1/
ĭa ₂ +⊼ ₂ 0	0.125	72	0.13	52	-	- /4/
moisture /3/	4.91	2,839	-	-	-	-
loss on ign. /3/	13.45	7.77 5	-	-	-	-
Cotal	90,095	5 7, 902	·100,16	40,000	100.0	6,007

_otos: /1/ Taken from Lable 2/1, Column 1 and adjusted for 4.9 % moisture /2/ Taken from Table 2/1, Column 2

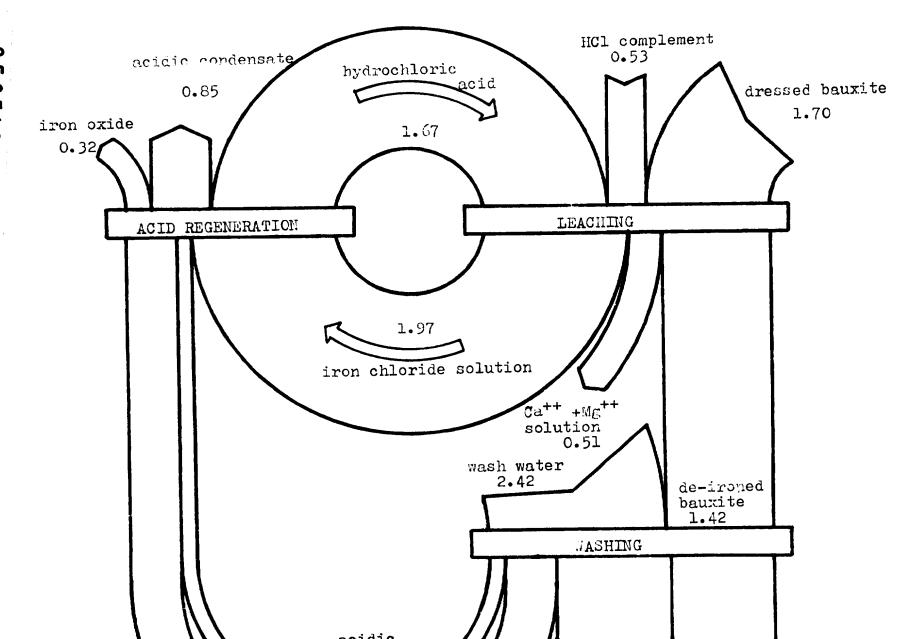
/3/ and /4/ see Table 6/1

Table 4/3
Material balances of a 40,000 ton plant to process medium grade /85/ bauxite

	Quan m ⁵ /yr.	tity t/yr.	Specific weight t/m3	Remarks
Leaching				
In: Bauxite	46,128	67,808	1.47	5 % moisture
Hydrochloric acid	76,737	88,247	1.15	330 g/l HCl
Out: Iron chloride liquor	59,343	86,641	1.46	130 g/1 Fe
De-ironed bauxite -	34,002	56,783	1.67	20 % moistui
Calcium + Magnesium solution	15,396	20,631	1.34	
Washing				
In: De-ironed bauxite	34,002	56 ,7 83	1.67	20% moistur€
Process water	96,582	96,582	1.00	
Out: 'White' bauxite	33, 948	55 , 675	1.64	
Acidic wash water I.	6,823	7,232	1.06	150 g/1 HC1
Acidic wash water II.	27,371	27,918	1.02	50 g/1 HC1
Acidic wash water III	. 61,921	62,540	1.01	1 g/1 HC1
Acid rememeration				
In: Iron chloride liquor	53 , 864	78,641	1.46	130 g/1 HC1
Acidic wash water I.	6,823	7,232	1.06	150 g/1 HC1
Acidic wash water II.	27,371	27,918	1.02	50 g/1 HC1
Process water	3 9,500	33,500	1.00	
Cooling water	1,676,500	1,676,500	1.00	
Fuel oil	2,700	2.500	0.93	Caloric value 41 MJ/t
Steam	-	29,500		160 °C, saturated

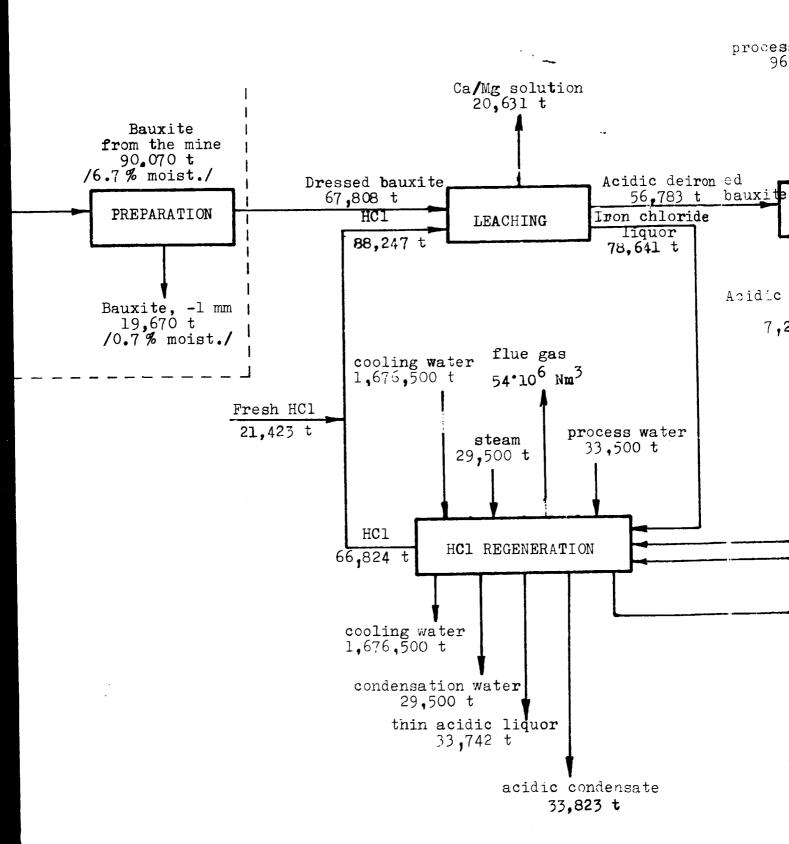
Table 6/3 /Cont'd/

	હ્ય	antity	Specific weight	Remarks
	m²/yr.	t/yr.	t/m3	TOMOTRO
Out: Iron oxide	27,136	12,754	0.47	
Spent	. 676 EOO	1 676 500	1.00	
	L,676,500	1,676,500	1.00	'000 lm ³ /year
Flue gas HCl	53,000	- 66 934	י זיב	. •
Thin acidic liquor	58,108	•		
Condensation water	•	•		0.0 E/I noi
Acidic condensate	25 , 000	33,823		6.8 g/l HCl
moraro convensa ec		JJ. 32 J		0.0 8/1 1101
Calcination		,		
In: White bauxite	33,358	55 , 6 7 5	1.67	20 % moisture
Fuel oil	5,700	5,300	0.93	Caloric value
Slaked lime	250	374	1.50	
Water for flue gas scrubber	187,000	187,000	1.00	
Out: Calcined bauxite	26.315	40,000	1.52	_
Flue gas	136 000	-		'COO hm ⁵ /year
Waste liquor from flue gas scrubber	1 57.376	158,950	1.01	
Production of iron oxide	pi nt		•	
In: Iron oxide	27,135	12,754	0.47	
Process water	31 8,350	318,850	1.00	
Slaked lime	17	25	1.50	
Additive	40	48	1.20	
Cut: Iron oxide slurry	1 8 333	22,000	1.20	50 % water
Spent wash water		•		

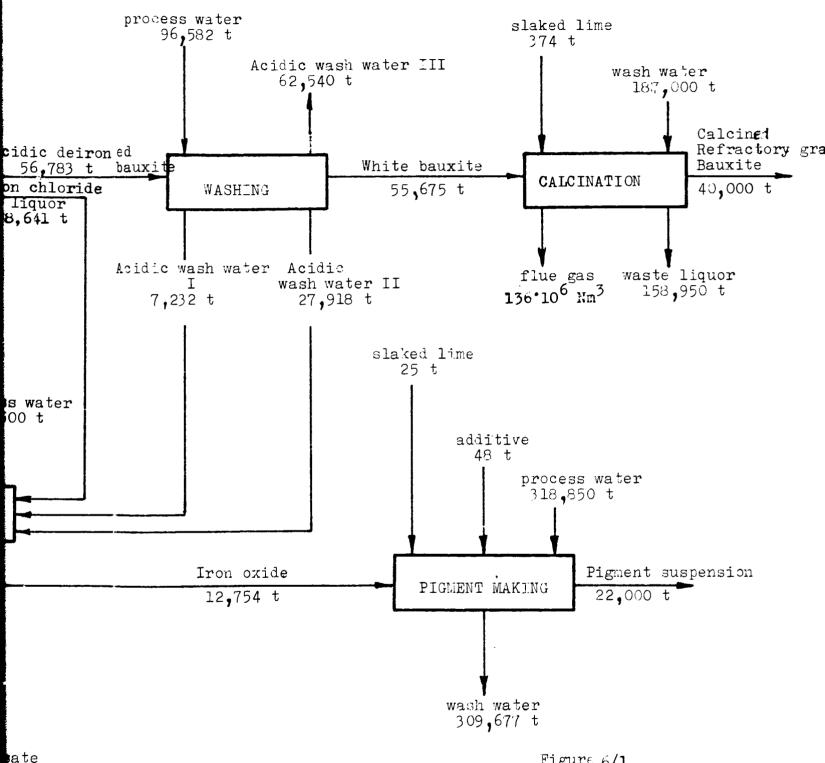


SEC

Z



SECTION 1



SECTION 2

Figure 6/1

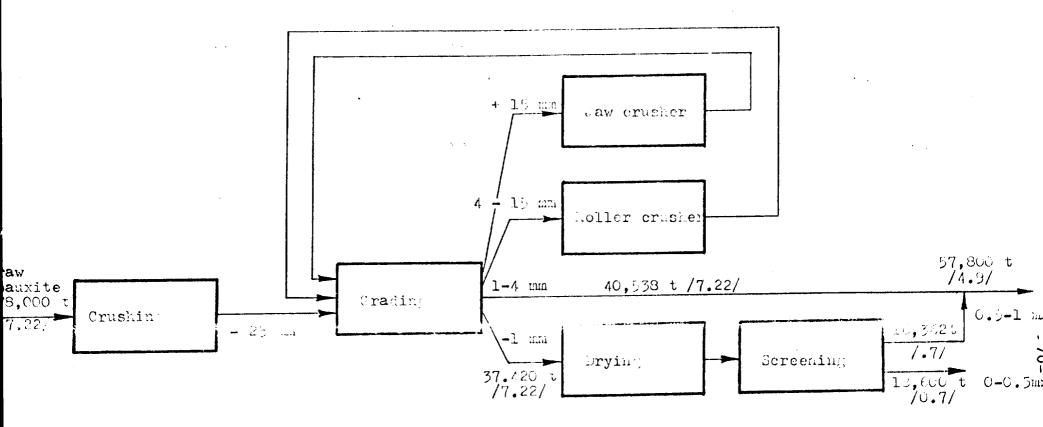
Block schematic with material flows /Plant size: 40.000 tons/yr/ Feed grade: '85'

C.2. Processes

6.2.1 Ore preparation

Crushing and screening tests of the medium grade bauxite are outside the Terms of a derence for this project as it was felt that the mines asked to supply the raw material would deliver the required grain size. It appeared, however, that feed for the plant will have to be purchased from a number of mines, possibly 3 or 4. These will be able to supply a consistent grain size bauxite, though this may differ from mine to mine.

If this situation prevails in the future, a dressing facility has to be provided at the auxite processing plant. Incoming material has to be stored /estimated storage requirement is 75 days' supply i.e. some 15,000 tons bauxite/, crushed, graded, ground, dried, sor ened, possibly blended and stored. Figure 5/2 shows a flowshest developed for the high grade /83/ raw material based on the limited information available on this material. The dressing unit is laid out for 78,000 tons bauxite of as-ined grade /moisture content: 7.22 %/. Apart from the production of 57,800 tons bauxite suitable for processing in the leaching stage, some 13,000 tons of fines /-0.5 mm/ is produced /moisture content: 0.7 %/.



Note: Figures in parenthesis indicate moisture content

Fi ure 6/2

Material flows for preparation of bauxite for a 40,000 ton plant using high grade /83/ feedstock

6.2.2 Leachin / Tashing

Lauxite having suitable grain size is treated with concentrated hydrochloric acid. Columns are filled with bauxite and the acid is slowly flowing through the fill.

The acid cycle /see Figure 3/ is carried out in four columns numbered by 1, 2, 3 and 4. Let's assume that column 1 is empty and ready for filling, column 2 is fed with acid, column 3 receives the solution leaving column 2 and column 4 delivers the solution to the collecting tank of the hydrochloric acid regenerator:

After a certain time, when the ion concentration of the solution leaving column 4 t gins to decrease and reaches a determined limit value /110-120 g/l Fe/, the cycle is stepped forward. The solution of column 4 is delivered to the previously filled or just being filled column 1. Column 2 is disconnected from the acid, rinsing is started in it. The other columns are also stepped up. In column 1. after saturation with solution, the so-called "calcium/ /magnesium solution" appear. which is neutralized. After decomposition of the carbon stes, Fe3+ concentration increases in the solution leaving column 1 and at a given concentration the solution is switched over towards the pyroliser. 10 days after the start of charging, the Fe3+ content in the solution begins to decrease and the cycle is stepped up again. Column 3 is rinsed, column 2 is charged. In this sequence column 1 is being charged again on the 40th day.

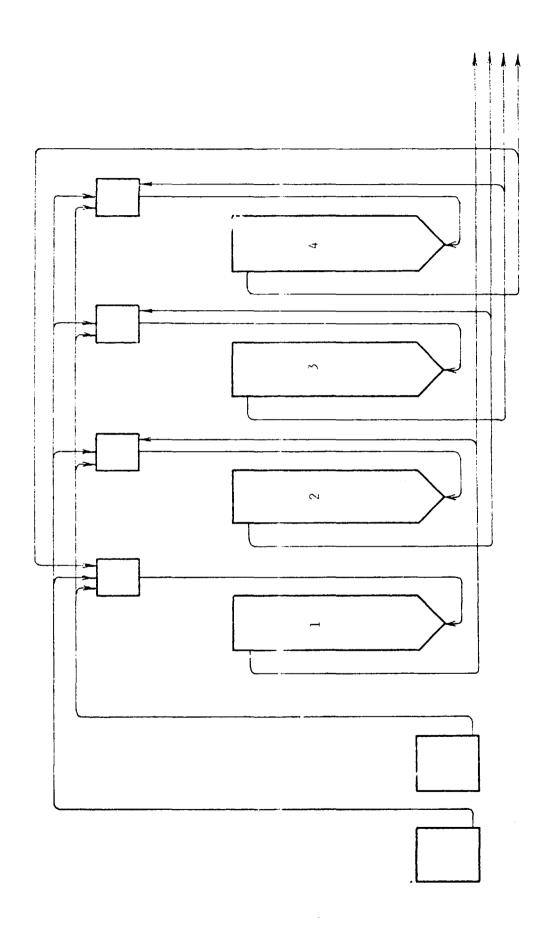


Figure 6/3 Schematic for leaching

The line diagram of above cycle is shown below. The abbreviations used mean /in c monological order of operations/:

- A = rinsing + bankite removal
- B = filling + draining the caldium/magnesium solution + production of iron chloride solution for the pyrolizer
- C = leaching, phase two
- D = leaching, phase three

Column number			C	per	atio	n s	
1	3	C	D	À	2.5	Ü	
2	A	3	Ü	D	A	В	
3	D	A	E	C	D	A	
4	Ü	Э	A	В	C	D	
	0 1	.0	20 3	4	.0	50	60 /days/

Phase "A" comprises the following steps:

- draining the acid from the bauxite; 12 hours
- rinsing with water; 192 hours
- draining with wash water; 8 hours
- removing the bauxite; 28 hours

The time required for phase "" is as follows:

- charging: filling the column voids in the bauxite fill of the iron chloride solution; 60 hours
- draining the calcium/magnesium solution; 25 hours
- iron chloride solution to the pyrolizer, 155 hours

The bauxite is filled into 24 steel tanks of 4.40 m diameter, 9 m high, having acid resistant rubber lining, arranged in two double rows. These columns are filled from above, by a conveyor track moving on a discharge car. This equipment should be laid out to transport 10 tons/hour of the bauxite to permit the continuous filling of a column within the 10 days cycle in each of the six leaching units. After filling the tanks are closed with a cover preventing the escape of acid fumes. Figure 6/4 shows the sectional view of the leaching unit.

During leaching 1098 1/hour of 30 % HCl should flow through the bauxite charge. A definite hydrostatic pressure must be provided for this purpose at the bottom of the bauxite charge. From the pilot plant experiences, this can be achieved by placing the feed tank 5 m higher than the upper level of the bauxite charge. The leached and thoroughly rinsed white bauxite is removed through large openings at the bottom of the column. The bauxite is fed on to a conveyor which in turn forwards it to a covered up area, where the water is allowed to drain.

6.2.3 Calcination

In accordance with the requirements of the tunnel kiln used for calcination the bauxite is pressed into bricks. The crude white bauxite should have 5 - 7 % moisture for this operation. Therefore, it must be dried before pressing. Moist white bauxite spread in a layer of 30 cm thickness will dry to a sufficient extent within 7 - 9 days without forced drying. The dry

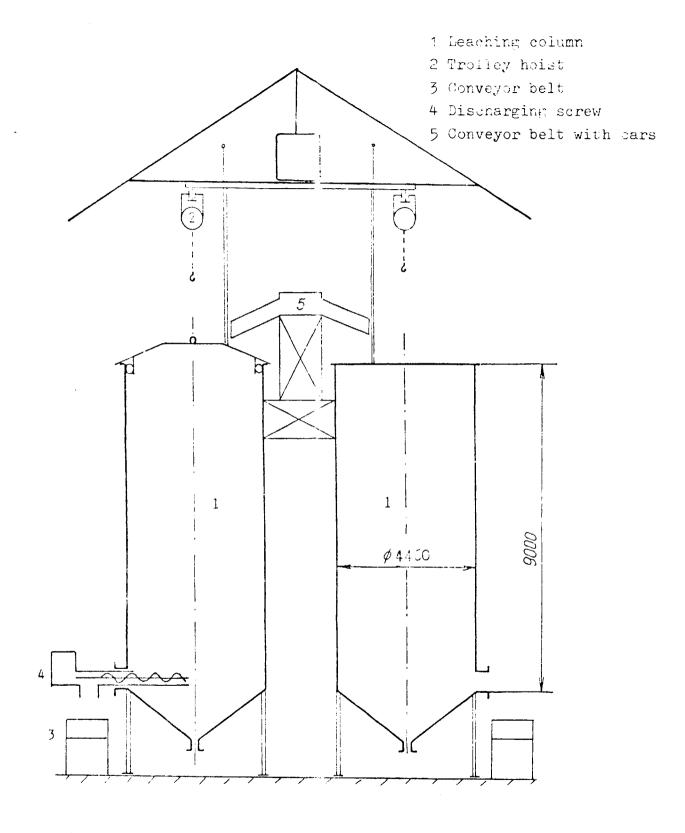


Figure 6/4
Sectional view of the leaching plant

white lauxite is charged onto a rubber conveyor belt transporting it to the hopper of the press.

The Urichs from the press are placed by an automatic charger onto the kiln trolleys. 9.5 cms of brick may be loaded on each 2.5 x 2.5 m trolley. Defore calcination, the moisture content of the bricks must be reduced; the loaded kiln trolleys spend 43 hours at 80 - 110 °C in the drying tunnel. Lext the cars pass into the tunnel kiln for calcination.

After calcination the bricks discharged from the trolleys are crushed and screened as required.

The medium grade /85/ bauxite is expected to yield relatively little fines, due to its medianical properties. A jaw crusher can be used to break it to -40 mm. In a further crushing stage, -10 mm material is produced. Roller crushers and a 2-level vibration screet can be used to produce various grain sizes, such as the one most in demand: 1-3 mm as well as 0.5 - 1.6 mm, 1-3 mm, 5-6 mm, etc.

6.2.4 Hydrochloric acid reguleration

During the 240 hours' leaching cycle, a high FeCl₃ solution /130-140 g/l Fe³⁺/ may be drain off from the freshly charged bauxite for approx. 155 hours and delivered for regeneration to the pyrolysis type acid regeneration unit. For a continuous operation of the pyrolizer 600 m³ solution has to be stored. During the storme the solid particles floating in the iron chloride solution will settle and can be removed.

Apart from the iron chloride liquor two effluents from the washing operations are also recycled through the pyroliner and their HOL content is also recovered.

The regenerated acid from the pyrolizer is fed into a buffer tank, where fresh acid is added to compensate for losses.

6.2.5 Iroduction of iron swide pigment

The method used in the pilot plant test to upgrade the iron oxide powder produced in hydrochloric acid regeneration an ion exchange procedure has been found to require much equipment in comparison with the possible benefit and produced a high amount of waste solutions with high salt content.

Therefore an alternative process is considered to remove the contaminants from the iron oxide powder and condition it for use as a pigment with good coloring properties. The iron exide powder leaving the pyrolizer is mixed in a tank with agitator with water and a lime milk solution according to its chloride content. The slurry is pumped into a tank with conical bottom for settling. The thick part is separated from most of the liquid on a vacuum filter. The retained iron oxide is again mixed with water in the tank with agitator. The slurry is settled and filtered. Wasling with water is repeated. The washing efficiency is increased by sprinkling wash water onto the filter dur. If filtering.

The required particle size of the iron oxide pigment is obtained by wet grinding. A slucry with 50 % iron oxide content is prepared from the filtered iron oxide and the additives for adjusting the quality parameters of the end product are blended into this. Two ball mills connected in series are used for fine grinding. The pigment slurry is stored in tanks with agitators. The product can be delivered in plastic drums, or plastic containers, etc.

Figure 4/5 shows the tentative flow-sheet of iron oxide purification and pigment production.

5.2.6 Process and cooling water treatment

The two large water consumers are the hydrochloric acid regenerator and the iron oxide purifier. If through flow type cooling is used, cooling after used in the pyrolizer can be used for rinsing the ir a oxide. The following requirement is to be made on the cooling water in this case:

- content of solids max. 5 mg/l

- pN value 7-8

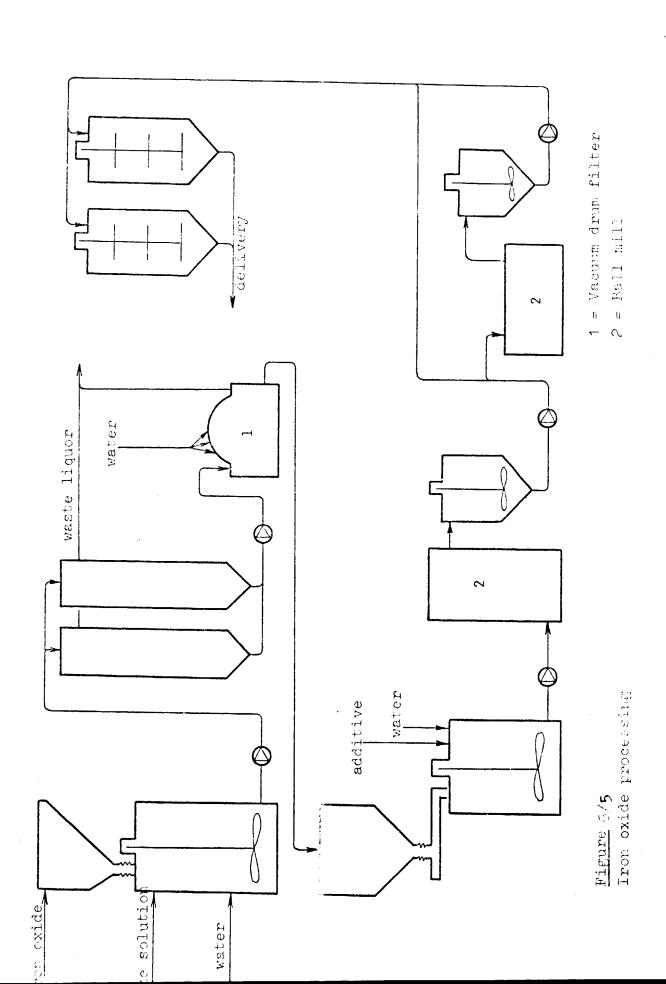
- carbonate hardness max. 17 German degree

- IMnO_A consumption 4,5 mg/1

- temperature max. 22 °C

- free CO₂ not permitted

Water treatment is determined by the quality characteristics of the available sweet water. As a site for the plant has not yet been selected, it is a sumed that sufficient volume



of water with the required temperature and quality will be available at the plant site, so water treatment procedure is restricted to storage, ascration and filtering through sand filters.

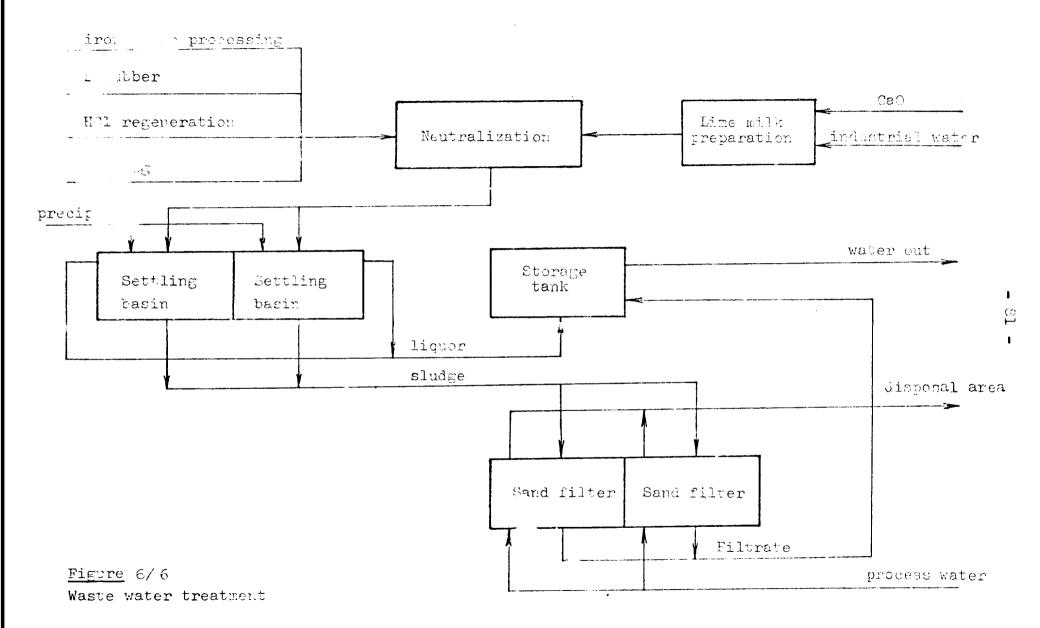
2.2.7 Lavinomicht protection

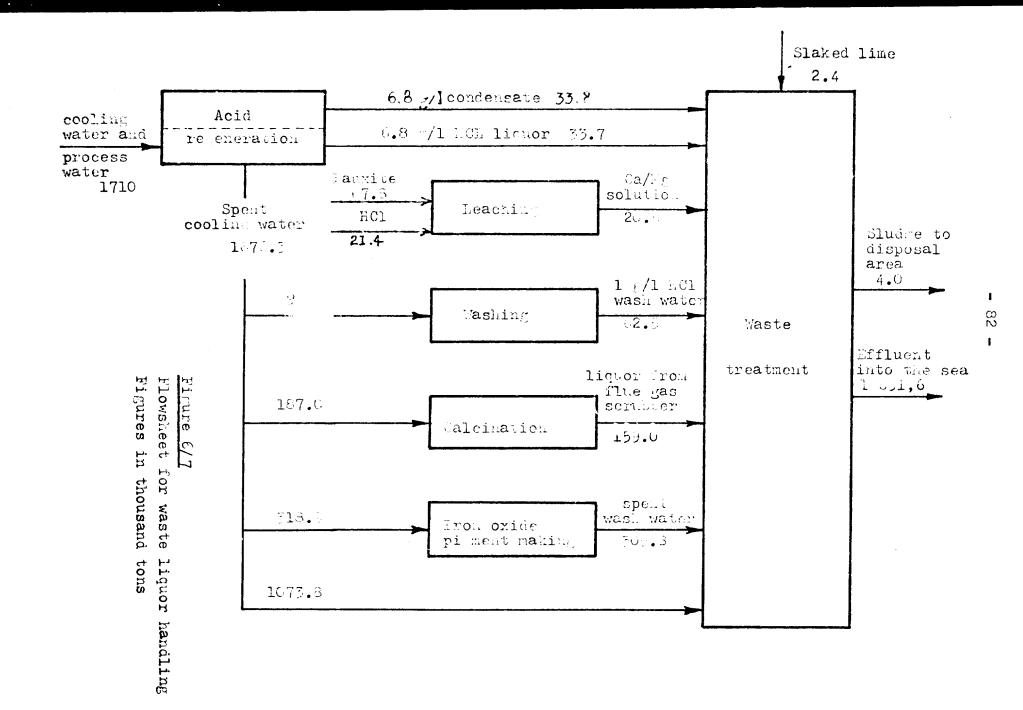
The Tatabúnya process is aimed at the complex utilisation of the bauxite ore, more than 70 % of non-water weight of the feed turned into products for sale. Hydrochloric acid used in the bauxite beneficiation plant is recirculated and re-used. The bulk of balance leaves the plant together with the hydrochloric acid losses and the oxidation products from the fuel.

a/ Licuid wastes

About 8.3 m 5 waste water is produced in the plant for each ton of raw bauxite. This is collected and treated as shown in Figure 7/6.

The actual quantities of liquid waste are shown in <u>Figure C/7</u>. Spent cooling water from the pyrollzer with a temperature of approximately 40 °C is used as process water for the other units and the remaining 60 % is used to dilute the effluent obtained after approximately 4,000 tons of sludge is segmated from it.





Later:

The studye is expected to have the following main components:

Al/GH/3: 3 %
CaSO, 2H₂C: 28 %
Iron oxide powder: 19 %

The salt of ent of technological origin in the waste water emit from the plant is estimated not to exceed

CaCl₂ 0.21 %

M Cl₂ 0.17 %

NaCl+RCl 0.005 %

45 %

b/ Flue ases, Pumes

The sources of flue gas in the plant are the pyrolizer, the funnel kiln, and the steam generator.

The pyrolizer should be fitted with a dust removal unit to catch iron oxide dust. All flue gases should go through a wet scrubber.

Table 6/5 shows the estimated parameters of the flue gases emitted from the plant.

Table 0/5
Flue gas parameter

_e arameter	Unit	Pyrolizer	kiln + Steam Generator
Lempera ture	o _U	opprox. 70	approx. 70
01 content	m_{γ}/n^3	.ax. 100	-
30 ₂ content	m / m_2	-	m ax. 50
Joliús content	_{mg/m} 3	ax. 100	max. 100
uantity	nm ³ /ton or raw bauxin.	600	2,600

6.2.5 fuel oil grounde

As natural gas is not available for industrial purposes in the part of Greece in question, fuel oil will have to be used for operating the hydrochloric acid regeneration equipment, the tunnel kiln and the steam generator. It is proposed that 3 % of the annual fuel requirement of 10,670 tons, i.e. about 300 tons fuel is stored in tanks at the plant.

6.2.9 Tank park

The tank park serves for the storage and the buffering of material circulation among the different solutions, fresh LCl, regenerated HCl, etc.

6.2.10 Minter operation

At temperatures below 15 °C, bauxite treatment does not provide satisfactory results. For continuous production, a temperature above this minimum must be provided in the leaching columns. In the winter season if leaching columns without heat insulation are to be used the solutions fed into it should have a minimum temperature of 25-30 °C. Craphite heat exchangers can be connected to each column to increase the temperature of the particular solutions.

6.3. Equipment list

A tentative list of the major pieces of equipment for a plant with a production capacity of 40,000 t per year and its technical features are given in the Table 6/6. A more detailed and accurate equipment list can be produced as a result of the detail engineering documentation of the plant.

Letter symbols in the "Number of units" column have the following meaning:

- A = 10. of installed units
- b = No. of functioning units
- C = No. of stand-by units.

thalloyed structural steet is suitable for cooling water, steam and fuel oil pipes. Hard PE pipes are proposed for hydruchloric acid, iron o'lloride and waste solutions. Special piping material necessary for the high temperature operation of the pyrolizer is assumed to be supplied by the Contractor delivering the pyrolizer unit.

Table 6/6 page i

Equipment list of 40.000 ton plant

Technological stop	Serial	Name of unit	Technical data	Numb A	er of B	unit C
1. Leaching, ashing		Front charger	Spoon size: 1 m3		1	1
	2.	Chargin; conveyor	W = 600 mm, $L = 35 ma$	2	2	-
	3.	Conveyor with discharging car	W = 600 mm L = 65 mm	2	2	-
	∳	leld thesesson to column	acid-resistant rubber coating dia. 4.46 x 3 m	24	24	
	5.	Overflow tank	Folyethylene tank dia. 0.5 x 9 m	24	24	-
	6.	Iron chloride collector tank	Steel tank with acid-resistant rubber lining dia. 6.3 x 5 m	4	4	-
	7.	Waste acid tank	Steel tank with acid-resistant rubber lining dia. 6.3 x 5 m	2	2	
	8.	Dilute acid tank	Steel tank with acid-resistant rubber lining dia. 6.3 x y m	£	4	-

Table 6/3 page ii

Equipment list of 40.000 ton plant

Technological	step Serial	lame of unit	Technical data		er of	unit C
	9.	Hydrochloric acid tank	Steel tank with acid-resistant rubber lining dia. 6.3 x 5 m	1	1	_
	10.	Water tank	Steel tank dia. 4 x 4 m	1	1	_
	11.	Feed tent	$\frac{\text{Polyndian}}{\text{V}} = 2 \text{ m}^{\frac{3}{3}}$	48	48	
	12.	Underground shaft tank	Steel tank with acid-resistant rub-ber lining and agitator, dia. 2 x 2 m	3	3	_
	13.	Shaft pump	Acid resistant Q = 12 m3/hour H = 16 m	5	3	2
	14.	Transfer pump	Acid resistant Q = 1 m3/hour H = 24 m	5 2	48	4
	15.	HCl pump	Acid resistant $Q = 10 \text{ m3/nour}$ $H = 24 \text{ m}$	2	1	1
	16. I	ron chloride pump	Acid resistant k = 10 mb/ lour N = 24 ll	2	1	1.

Taile (/k page iii

Equipment list of 40.000 ton plant

Technological step	Serial no.	Name of unit	fechnical data	lvumb A	cr of E	unit C
	17.	Waste liquor pump	Acid Desistant C = 12 m./hour E = 16 m	1	1	-
	13.	Diluted HCl pump	Acid resistant Q = 12 m3/hour H = 16 m	1	1	-
	19.	Water pump	Q = 10 m3/hour $C = 24 m$	1	1	-
	20.	lischarging screw	With gear & motor cty: C t/hour dia. 250 x 5.000 m	4	4	-
	21.	Trolley hoist	Electrical operation manual operation for 1 t load		4	_
4	22.	White bauxite discharge conveyor	W = 600 min $L = 70 m$	4	4	
	23.	White bauxite collecting conveyor	W = 800 ma $L = 20 m$	1	1	-
	24.	Conveyor	W = 800 mm $L = 35 m$	1	1.	_
	25.	Covered store e yard	Area: 2.200 m2	1	1	

Table 6/6 page v

Equipment list of 40.000 ton plant

Technological step	Serial	kame of unit	Technical data	Kumbe	er of	unita
	no.			A	ظ	C
	14.	Conveyor	W = 600 mm, $L = 5 m$	2	2	-
	15.	Jaw crusher	hax. lump size: 80 mm Cty.: 6 t/hour _	1	1	-
	16.	Gas scrubber	Cty.: 10.900 Nm ³ /hour	1	l	-
	17.	Roll crusher	Max. lump size: 15 mm Cty.: 6 t/hour	1	1	-
	, 18.	Conveyor	W = 600 mm, L = 12 m	2	2	-
	19.	Front charger	Scoop size: 1 m ³	~	1	-
3. HCl regeneration	1.	Andritz- athner iron chloride do-composition equipment	Oty.: 7.25 m ³ /hour hydrochloric acid with with 330 g/l	1	1	
	2.	Iron chloride settling tank	Steel, with conical bottom and acid-resistant rubber lining dia.: 2.5 x 10 m	3	3	_
	3.	Emergency water tank	Steel, dia.: 2 x 2 m	1	ı	-
	4.	Iron chloride pump	Acid-registant Q = 10 m3/hour H = 24 m	2	1	1

Table 6/6 page vi

Equipment list of 40.000 ton plant

Technological step	Serial	Name of unit	Technical data	- Numb A	or of	unit C
	5.	Slurry pump	Acid resistant			
	∫•	orariy pamp	Q = 12 m3/hour	_		
			H = 16 m	1	1	-
	6.	Water pump	C = 4 m3/hour H = 36 m	1	1	
	7.	Iron oxide bunker	$V = 50 \text{ m}^3$	1	1	_
	1 •	Tron oxide bunker	v = 50 m	.L	1	_
4. Iron oxide pig-			3,.			· · · · · · · · · · · · · · · · · · ·
ment production	1.	Bucket charger	Cty.: 2 m ³ /hour	1	1	-
	2.	Charging bunker with balance	$V = 10 \text{ m}^3, 5 \text{ t}$	1	1	_
	3.	Tank with agitator	Steel tank V = 75 m3	1	1	-
	4.	Slurry pump	© = 40 m3/hour H = 24 m	1	1	_
	5.	Settling tank	Steel V = 40 m3/hour	2	2	
	6.	Slurry pump	Q = 40 m3/hour H = 24 m	1	1	
	7.	Vacuum drum filter	Cty.: 20 m5/hour iron oxide slurry	1	1	-
	8.	Discharding screw	Cty.: 2 t/hour moist iron oxide	1	1	-

Table 6/6 page vii

Equipment list of 40.000 ton plant

Technological step	Scrial	Name of unit	Technical data		er of	
	no.			A	В	Ü
	9.	Storage tank for filtered iron oxide	V = 8 m3	1	1	-
	10.	Charging scew with balance	Cty.: 6 t/hour	1	1	-
	11.	Tank with agitator	V = 10 m3	1	1	-
	12.	Slurry pump	Q = 10 m3/hour H - 16 m	1	1	_
	13.	Dall mill	l: ing volume: 4.5 m3	1	1	-
	14.	Storage tank	Steel ink with agitator V = 10 m3	2	2	-
	15.	Ball mill	Milling volume: 2.2 m3	1	1	_
	16.	Slurry pump	Q = 12 m3/hour H = 24 m	2	2	_
	17.	Storage tank	Steel tank with agitator dia. 6.3×5 m	2	2	-
5. Neutralization of waste liquors	; l.	Slaking tank	Steel, dia. 1.5xl.5 m	1	1	
or won to miduois	2.	Discharge screw	Cty.: 200 kg/hour	1	1	_
	3.	Lime milk tank	V = 10 m3	1	1	_
	4.	Neutralizing tank	V = 25 m	1	1	_
	4. 5.	Settling basin	V = 200 m	2	2	
	→	Sarata, Sarati		_	_	

Table 6/6 page viii
Equipment list of 40.000 ton plant

Technological step	Serial no.	Mame of unit	Technical data	Rumb A	er of B	unit C
	6.	Flocculant tank	V = 5 m3	1	1	-
	7.	Slurry pump	Q = 2 m3/hour H = 12 m	2	2	_
	8.	Pump	C = 50 m3/hour $H = 24 m$	2	1	1
6. Water treatment	1.	Collector tasin with acrator	V = 1.000 m3	1.	1	-
	2.	Water pump	Q = 250 m3/hour H = 55 m	2	1	1
	3.	Sand filter	P = m3/hour	2	2	-
	4.	Filter rinsing pump	Q = 25 m3/hour H = 36 m	1	1	
7. Fuel oil storage	1.	Gear pump	Q = 380 l/minute p = 5 bar	2	1	1
	2.	Gear pump	<pre>Q = 16 1/minute p = 5 bar</pre>	2	1	1
	3.	Storage tank	Steel, V = 450 m3 dia.: 6.3 x 5 m	6	6	-
	4.	Oil filter	Q = 400 l/minute	2	1	1
	5.	Oil filter	Q = 20 l/minute	4	2	2

Table 6/6 page ix

Equipment list of 40.000 ton plant

Те	chnological step Se	erial	Name of unit	recasical data	Numb A	er of B	uni t C
8.	Winter operation	1.	Condensate tank	Steel, V = 50 m3	1	1	-
		2.	Graphite heat exchanger	Surface: 0.5 - 1.5 m ²	48	48	_
		3.	Warm water pump	$T = 90^{\circ} C$ Q = 25 m3/hour H = 36 m	2	1	1
9 .	Automatic measurement and control system	1.	Central processor	For 1024 TTL charmels Interface NS 232	2	1	1
		2.	Colour display with disc drive	2 x 27 HByte Winchester disk	2	1	1
		3.	Magnetic disc unit	1 MTyte floppy	4	2	2
		4.	Printer		2	1	1
		5.	Level sensors	Inductive			
		6.	Temperature sensors				
		7.	Alarm sensors				
		8.	Pump activators	with magnetic valve			

Table 6/6 page x

Equipment list of 40.000 ton plant

rechno	<u>-</u>	rial	Rame of unit	rechnical data		er of	units C
10. Au	xiliary equipmen	t	ayadangalagangan geliga a ayada ma diba a gelimpa menga a ambana men			, 	
	Compressed air				_		
	supply	1.	Compressor	Cty.: 150 Nm3/hour	1	1	-
		2.	Compressed air tank	V = 20 m3, p = 6 bar	ı	ı	
10.2.	HCl complement	1.	Storage tank	Steel, with acid- resistant rubber lining, dia.: 6.3 x 5 m	3 .	3	-
		2.	Discharge pump	Acid-resistant, Q = 24 m3/hour H = 24 m	1	1	_
10.3.	Steam genera- tion		Steam generator incl. water treatment unit	3 tons/hour 158 °C saturated stram	2	2	1

6.4 Location criteria, utility requirements

When site selection criteria for the bauxite beneficiation plant are developped availability and cost optimum issues have to be considered.

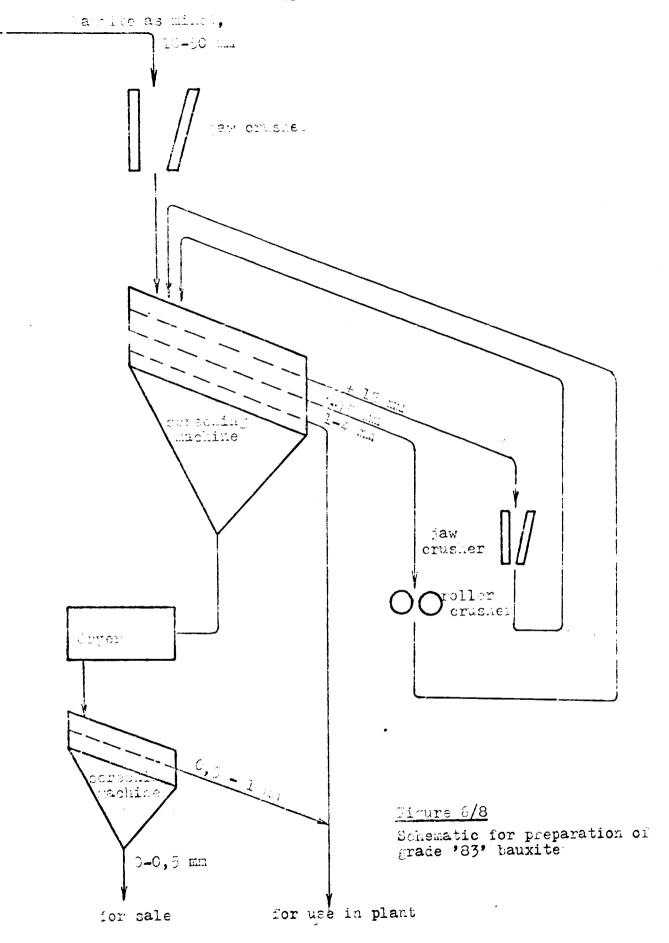
Requirement data from earlier sections of this Chapter 6 is used to calculate the parameters for a 40,000 tons plant laid out to process the medium grade /'85'/ bauxite arriving to the plant in 0.5 to 4 mm grain size. Following a suggestion from 1.170, estimates for a plant with similar output capacity processing Grade '35' bauxite are also given. These, however, necessarily have a higher confidence margin as laboratory results were used to generate them.

1. Land_

Grade '83'
290 x 540 m
15.7 ha
38.7 acre

This is based on the tentative layout in Section 6.5. Figure proposed for Grade '83' includes bauxite handling. A schematic diagram of this appears in Figure 6/8. /If it proves to be necessary to dry the 0-1 mm fraction, waste heat from the tunnel kiln should be used./

An increase of the plant output capacity by 20 % in either case would require an additional land area of up to 10 %.



lable .//
Lable ..//
Lable .

<u> </u>	rad e ' 85'	lrale '83'	mode of transport		
auxito, as misso, 1990 t/yr	(20 -	79	loose by scal/ rowl		
aamite, 0.5-4 ms, 1800 t/yr	7.8	, p 7. 8	looss ty sca, / roul		
ydroelloric acid,	21.4	ક.∴	in talles by sea / road		
dlaked line, foot	2.8	2.0	/ loogs bv ses / row.		
Calcined retractory Lauxife, / 200 t/jr	∴ 0.0	40.0	loose or it bags by read / sed		
or dried() % moisturd dide watelies, 1988 type, up to	30 . C	50 . 0	/ loose /lj road / sea		
llaon chi le piquent, thi i solide, 1000 cypp	22.0	11.2	in talks by road / sea		
curive reject D = 0.3 mm, 'COO t/yr	19.7	13.6	in ba, s by sea / road		
Notal, 'GCC t/yr, up to	273	2 26			

2. Lain raw material and products

Data for this section is taken from Tables 6/3, and 7/4. Results are in <u>lable 6/7</u>. The total material forwarding requirement for the plant is in the order of 275,000 tons /225,000 tons in the case of high quality feed/ per year.

3. Ltilities_

The consumption figures are derived in essence from the basic engineering data, and list of equipment appearing in Sections 6.2 and 6.3. An input/output schematic has been drawn up for the 40/85 plant and is shown in https://c.com/listing/figures-are-summarized-in-fable-c/6.

4. Wastes

This is described in detail in Section 6.2.7. The 40/85 plant will produce approximately the following wastes:

Ilua gases,	fumes	190	•	10 ⁶	Nm ³ /yr
waste water		1.7	•	10 ⁶	t/yr
solid waste	/sludge/	4.0			t/yr

The 40/85 plant would produce roughly the same amount of flue gas, about 40% of the waste water produced by the 40/85 plant, that is about 0.63.10° ton/yr, since this is more or less proportional to the LCL consumption; Solid waste is estimated for 2.5 ton/yr.

5. Usight of plant equipment

This is derived from Section 6.3 and estimated as 4,700 tons for the 40/85 plant.

6. Labour requirements

Pased on the experience from MIRAOVA's PÉCS plant, prior to commissioning a workforce of 30 to 35 is needed in the 4 months' period of earth moving / foundation laying,

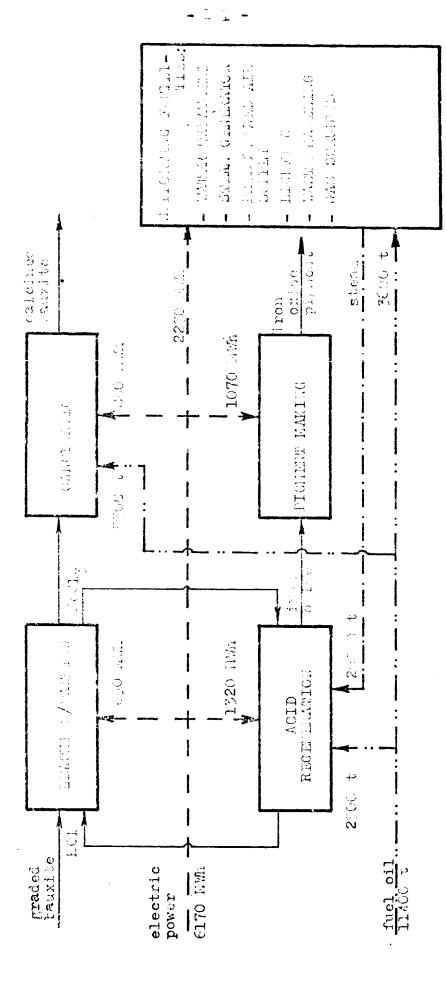


Figure 6/9
Phergy used in the 70/35 plant

lable 1/2
Utilities

	Grade '89'	erale 'dj'
lectric pawor, Concemption, 1711/yr 1811, 1821, 171	6,170 50	4 , 570
ivel oil lon	11,400	7,300
uroc sa waler .971m	230	130
uotable water m²/m	٤	4

average of the 14 months of eventury, start-up, etc.
There is any all as the personnel of the owner of the plant.

for the operation of the plant, altojether of periods will be required, you which is management. As yo workers are expected to work in continuous work order, there would be a maniful of 37 people in work at any one time.

7. dive propies into

irotably the most difficult requirement to satisfy is that or major supply. The 230 m³/hr value is for flow-through type cooling of the pyroliser. If cooling water is recycled, we 40/85 plant's consumption of process water is above 100 m³/hr, subject to the maximum requirements on chlorides in the effluent.

vicinity to an alumina plant is advantageous from the point of view steam supply, leboratory facilities, possible emphasis of materials, /out-of-spec alumina for the state of materials.

Olosomess to a mine becomes less important less pply is to tecovores arom several mines at various places.

hocation close to or on the seaside helps to save transport costs for both raw materials as well as the products. Onlorides in the effluent of the plant are all natural components of seawater.

(.5 Lagout

Pinure 6/10 shows the tentative layout for a bauxite feneficiation plant with an annual output of (0,000 tons /Feed: Grade '8)'/. The arrangement was based on the principle of keeping feed stock as well as iron oxide as far as possible from the calcined white bauxite in order to reduce the danger of contamination of the end product.

The figures in the drawing are explained below:

- 1. Raw material preparation
- 2. Lunker for -0.5 mm fraction
- 3. Dryer
- 4. Gas scrubber
- 5. Paumite feed conveyor belt
- J. Leachin, / Mashing column
- 7. 'Maile' bauxite conveyor belt
- 3. Demadering area for white bauxite
- 9. Erick press
- 10. Trolley loading machine -
- 11. Drying turnel
- 12. Tunnel kiln
- 13. Trolley unloading machine
- 14. Calcined bauxite, handling
- 15. Calcined bauxite, storage
- 11. Cooling water system
- 17. Fuel oil tanks
- 18. Steam generator
- 19. Tyrolizer
- 20. Iron oxide, washing
- 21. Pigment shop
- 22. Tank park
- 23. Waste liquor treatment

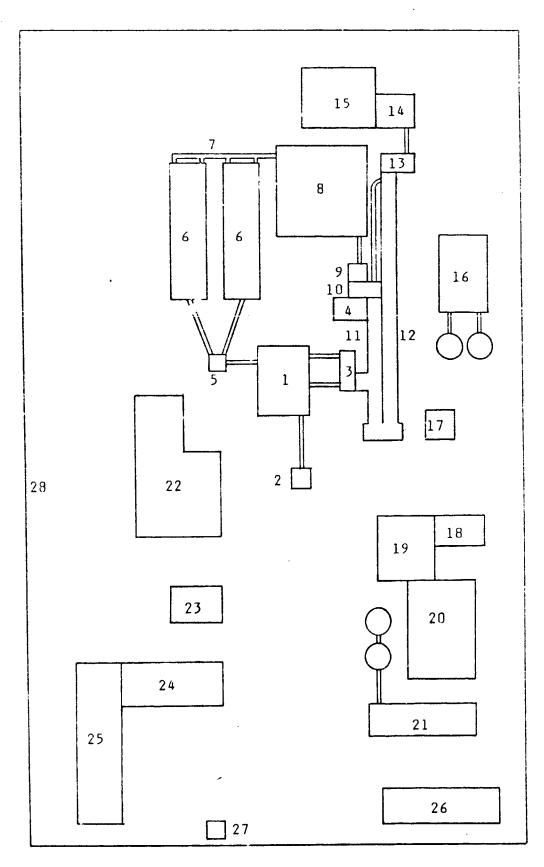


Figure 6/10
Tentative layout /40/85/

- 24. Corvice building
- 2). Offices and lass
- 26. Thops and warehouses
- 27. Jate
- 23. Bolice

7. Cost estimates

In this chapter investment and operating costs will be estimated for alternative plant capacities for processing a raw material similar to the grade tested in the pilot plant /see Tables 2/3 and 2/4/, to be referred to as Grade 85. Four plant sizes with outputs of 20, 40, 60 and 80 thousand tons refractory grade bauxite, respectively will be covered. Investment and operating costs given have been reached on individual assessment in each case. The 40 thousand ton plant is selected as base variant. The cost figures used in the report are, as a rule, actual at the time of writing. Prices of imported equipment include sea freight, but no import duty.

Although limited amount of engineering information could be developed for the bauxite grade tested in laboratory in 1932/83 with excellent results /to be referred to as Grade 83/, an attempt has been made in this section to estimate, for the purposes of comparison, the investment and operating costs for a 40.000 ton Grade 83 plant.

Eased on tables containing relevant data developed for altogether 5 variants, comparisons will be made and an evaluation will be given.

7.1. Investment cost

Investment cost of realizing the project will depend on site selection, on the scope of the project, feedstock grade, capacity installed, and on other factors. In this section estimates for the investment cost will be given taking the effect of the feedstock grade on the one hand and that of the installed capacity on the other, in consideration. Following a request by HIMIC, local and Forex components of cost of equipment and machinery have been estimated for each item.

Table 7/1 lists cost of equipment, installation cost, as well as the cost of building and construction for the base variant of 40.000-ton annual capacity. Cost of land is estimated for Drs 500,000 per acre unit cost inclusive site preparation. The cost of making the necessary amount of electrical power and water available to the plant as well as those of making the plant accessible by road and/or rail are not included in the figures given. The cost estimates for the individual plant sections have been based on the equipment list in Section 6.3, as well as on respective data from MINAOVA's own Pécs plant now under construction. Although it is not clear at this time, if bauxite can or cannot be purchased from the mines in the required grain size range of 0.5 to 4 mm, and it may prove necessary to allow for in-plant dressing facilities, these have not been included in the Table, as too little is known at this stage

Table 7/1

Cost of fixed assets for a 40.000-ton capacity plant /Grade '85'/ /all figures in /000.000 Drs/

Plant section	Cost of Building and Construction	Dau	t of ipment hinery	and	Cost of Installation	Total
Site, developped	17	Tota -	1/Fore: -	:/Greek -	-	17
Subtotal, site	17	•	-	-	-	17
Leaching/Washing	75	528	40	488	92	681
HCL regeneration	7	564	520	44	79	650 \
Calcination	65	380	380	-	65	510
Gas Scrubber	11	72	72	-	12	95
Waste Freatment	2	18	-	18	2	23
Pigment Processing	45	76	76	-	14	135
Subtotal, Frocess Equipment	205	1638	1071	550	251	2094

Plant section	Cost of Euilding and Construction	.lgu	t of ipment ninery	and	Cost of Installation	Total
		Logal	rorel	uneek		
Piping, Sewers and loads	40	14	-	14	3	62
Meetric lower Jupply, Lighting	9	15	•	15	5	29
Steam Generator incl. Water Preatment	9	51	51	•	5	. 29
Instrumentation and Process Control	3· -	77	37	· _	11	48
Toler Supply	22	12	-	12	4	38
Compressed Air Supply	_	8		8	1	9
Fuel Oil Storage and Handl	ing 3	9	-	9	3	15
Pical Project Handling	12	7		7	2	21
Distotal, Auriliana Encilities	95	153	03	, 5	42	290
Maintenance shops and sloc	115 22	11	y rang garantig derektigiské i franksette filozóf í Saldan	11	1	34
Fant Languero a militara e mon facilities	3.ld 29	-	general species of the species of th	Photo and the second se	and	2 9
Sectoral, Prilaters	51	i.L	-	11	1	63
onal dinatarrata	3.13	1.302	13.79	er var i November	2.72	2464

about the behaviour of the material.

of application, he foreign exchange component of the cost of application, it was assumed that the leaching/washing columns which be represented in Greece. Hel representation equipment, estemperation below that the passes as well as pigment shop equipment will be imported, as all these incorporate special engineering and high grade materials. These assumptions are rather conservative for lack of precise information and it may well be possible to further decrease the Forex component.

Cally 7/2 shows the cost of equipment and machinery for different sastalts. The sources of equipment price estimates are an follows: Some are unit prices actually incurred for equipment item, in MINLOVA's out plant at Pécs. Others are quotivious obtained from suppliers including Andritz of Ametria, Lettsch of TRG, A46M REck Electric of hungary and others. It can be seen, that overall cost approximately doubles when the capacity increases fourfold. The use of Grade '85' over the ideal Aral. '65' makes it necessary to process more material and to recycle more HCL for a given output. This and other factors bring about an increase of approximately 10 to 20 % in equipment costs, even though bauxite drepsing is included for Grade '83', as requested by HIMIC in the review meeting.

16010 2/2

Cost of rguipson, and acchingy for different plant capacities /in /ct3 toms/

Reedstock

Start deriview						
in. 20 40 60 80 10.	Flant gerlion		(:J:ad	\$ 0		Track 32.
- Analysis	Lant captoin	20		9	80	40
### 1900	ed Language be	1		ı	Î	47.
- ener 510s		5	(7) (1) (2)	o A	() ()	
225 2-50 423 A72 120 2-44 277 250 1 244 277 250 1 244 277 250	TI we ener dies	100	\$	130	844	20,000
250 75 05 122 1 254 277 850 1 255 1 257 2 125 2 255		\$0\ 0\ 2\ 2\	9	n) en st	3	098
1 244 277 250 1 245 1 202 2 125 2 204	:	S. S. E.		(f) (f)	C: C:	©
		and and a	전 (건 (건)	577	830	23.7
		1	2000 C T	04 04 05 100		O Commence design

the last provides a birallal total investment cost famour total or the disclerent variants. The cost of estational literact for the use of the proprietory handles be estationed for 10 f of cost of process equipment, plus notation may be the been estimated for 5 f of the obtained cost of the estational has been estimated for 5 f of the obtain cost of lived accets, pre-investment studies as 1 f of the total investment cost. These fi was are purely estimated for the purposes of the present accurate assentant overcise and do not represent actual offer by any party.

In has been ascumed, that a stail of 5 qualified persons cay us necessary for a period of up to 2 years, for the material of the implementation of the Indicated including a convision and overall coordination of the installation, towar man and take cover of civil veries, equipment and plant, regardleds of the installed cay with washes on a non-bly dyspace solary of 10.000 Taudimas, the actual cost estimate for this activity is 20 million brackmas.

A Legarive them is dereseen at 10 , of the total investment

The working capital requirement estimates for the various variants are based on the cost of purchasing 75 days

Table 7/3

Total investment cost

/All cost Tigures in *000.000 Drs/

Estimate for			${ m F}\epsilon$	edstock		
Tlant capacity in •000 tons		Cr	ade '85	,	Grade '	33'
	20	40	60	. 80	40	40 [¥]
Plant site, developed	12	17	21	27	19	15
Sub-Subtotal, Site	12	17	21	27	19	15
Crude Bauxite Preparation	-	_	•	-	88	88
Leaching/Washing	442	€81	7 95	920	584 5	.84
HC1 Regeneration	440	6 5 0	780	870	450	150
Calcination	433	510	570	590	510	-
Cas Scrubber	77	95	103	116	84	55
Maste Treatment	19	23	27	<i>:</i> 0	23	23
Pigment Processing	94	135	160	17 5	95	Jö
Sub-Gabtotal, Process Equipment	1,510	094ر2	2,440	2,701	1,834 1,2	235

Estimate for			Feed	stock		····	******
rlant capacity in 000 tons	<u> </u>	Grade	'85'		Grad	e '83'	
	20	40	60	80	40 ^{**}	40	
Piping, Sewers and Roads	54	6 2	71	77	58	58	
Electric Power Supply/ Lighting	25	2 9	33	3 5	27	29	
Steam Generator, incl. Water Troatment	46	68	81	98	55	55	
Instrumentation and Process Control	46	48	49	49	40	47	
Water Supply	30	38	41	45	30	30	
Compressed Air Supply	8	9	10	11	3	8	
Fuel Oil Storage/Handling	12	15	17	18	10	12	
Final Product Handling/ Packing	17	21	24	26	ć	21	
Jub-subtotal, Luxiliary racilities	238	290	32 6	359	2 34	260	
flant Fanagement and Welfare	31	34	36	33	34	34	
Maintenanco Shops, Stores, e	tc. 29	2 9	31	32	2 9	29	
Sub-Subtotal, Buildings	60	63	67	70	63	63	
Sub-Total, Fixed Assets	1 820	2 464	2 854	3 154	1 :07 2	176	

Service of the control of the contro		THE RESIDENCE OF THE PROPERTY		
A CONTRACTOR AND		নহন : ক ^{ানি} নীয়	orade 1691	•
The state took and the state of		en die Afficiation de minimization en dies sites dies	40	* -(
A COUNTRY TEST AND TO COUNTRY TEST AND THE PARTY AND THE P			0	130
10 00 00 01 1 01 1 01 1	98 () 2 () 3 () 4 ()	te s	: N	53
Tankin - Oli			52	S
	3 143 158	æ	1.09	83
A Company of the Comp			<u>c</u> a	25
the state of the			689	25.2
			- ; 9)	· ·
	1			
	٠.	V .,		(M) (M)

Market to a soid, his of the most oil the projected for use.

investible cost is to be sport on process equipment for leading/stream, 18-20 % on HOL recemenation equipment materials of the process and of the appropriate and conjugate additional items of the up to 40 % of the total cost of putting up a plant and solution as a cost of putting up a plant and solution and investment cost about the cost of the voll investment cost about as a cost of the 80 thousand-toning, the cost of decision cakes savings of approximately 12 % and the cost of the 20 % to the revenue of the plant, appear to 20 % to the revenue of the plant, appears to 20 % to the revenue of the plant, appears to 20 % to the revenue of the plant, appears to 50 %, but the actual additional investment costs of disposal is still lower.

production with the problem of the 20.000, the contract of the 20.000, the contract of the 20.000, the contract of the contrac

proposition of the state of the proposition and different varieties.

constant to addition the plant is baunite gradel and constant to addition the plant is baunite gradel and constant a moisture against the constant and the cons

the first of the operation occurrent the 4 variants

on the contract the operation occurrent, as well as estimated

on the contract the contract this latter is taken in as-which

on the contract.

variants rounded to the costs were obtained from variants rave by a variant from some sources as given below.

- Markito: Using the formula proposed by the UNIXO export in his Report /2/, Grade '85' should be available at big Alth.-/ton, Grade '83' at Drs 511C.-/ton.(Mine op propose have been quoting somewhat higher prices.)

Table 7/4
Consumption data estimated for various plants

Variants

Input	unit	20/85	70/85	GU/85	80/85	40/85					
Bauxite	ton	3 3.904	67,808	101,712	135,816	57,800					
нсі	ton	11,800	21,400	31,200	40,500	(10,000					
Slaked lime	ton	1,400	2,800	4,200	5,000	2,000					
Precipitant	kg	11,800	22,000	31,700	42,000	10,000					
Additive for Pigment making	kg	24,000	48,000	72,000	96,000	24,000					
Fuel Oil	ton	5,675	11,400	17,125	22,850	7,350					
Steam ·	ton	14,750	29,500	44,250	59,000	16,500					
Compressed Air	•oco m ³	820	900	980	1,060	860					
Electric Power	MWh	3,600	6,170	8,700	11,000	4,570					
Process water	• 000 m ³	895	1,710	2,590	3,470	990					
Potable Water	~ 000 m ³	25	27	30	33	27					
Spares and wear Farts	∕ 000 Drs	36 , 000	40,000	42,000	43,000	35,000					

Table 7/5
Cost of operation estimated for various plants /Plant capacities in '000 tons/
/all cost figures in '000.000 Drs/

Unit	Cost item			Cost	for		
Drs/unit	Feedstock	G:	rade '35	7		Gr	ade '83'
	Plant capacity	20	40	60	80	40	% of total operating cost
4,455 6,193	Crude Bauxite, ton	151	3 02	453	604	35 3	42.27
7	Cost of Transportation	7	14	21	28	16	1.89
6,100	Hydrochloric acid, ton	7 2	130	190	247	61	7.2
- `	All other materials	12	22	31	40	16	1.89
-	Sub-Subtotal, Materials	242	4 63	695	379	451	53.25
24,000	Fuel Oil, ton	136	274	411	548	176	20.77
7,000	Electric Power, kWh	25	43	60	77	32	3.77
-	Other Utilities/supplies	3	5	7	9	3	0.35
-	Sub-Subtotal, Utilities	164	3 22	473	634	211	24.91
-	Royalty on refr. sales	7	14	21	28	22	2.6
-	Royalty on pigment sales	3	16	24	3 2	9	1.6
-	Sub-Subtotal, Royalties	15	30	45	60	31	3. 66
-	Subtotal, Variable costs	417	314	1,211	1,604	691	81.53

Cost item	+		Cost	for		
Feedstock	 	Gr	ade '35'			Grade '83'
Plant capacity	20	40	60	80	40	% of total operating cost
Wages and Salaries	100-	109	113	127	109	12.37
Factory overhead	40	42	43	45	36	4.25
Administrative costs	3	11	13	14	11	1.3
Subtotal, Fixed Costs	148	162	174	186	156	18.42
Total	565	976	1,395	1,790	347	100.00

The purchase price of Drs 4455/ton is used in this Report for the C.S to 4 mm medium grade /'85'/ bauxite assuming a 10 % surcharge for dressing. A different assumption is made for the high grade raw material as agreed at the Review Resting held in Dudapest and récs on 22-25 October 1986 /9/: raw material bought as-mined for USD 42,-/ton /Drs 5676/ton/ and fines from grading and screening resold at LSD 39,-/ton /Drs 5265/ton/. Table 7/6 shows how a unit price of Drs 6195/ton was finally arrived at, based on the material flows of Figure 6/8.

- Cost of transportation: Drs. 7.-/ton kilometer /source: HIMIC./An average road distance of 30 km is assumed.
- Hydrochloric acid: Drs 6100/ton /30 %/
- Fuel oil: Unit cost of Drs 24,000 has been assumed, although HIMIC stated Drs 27,150.

It should be noted, however that present world market prices for suitable quality fuel oil grades are up to 50 % lower /from USD 80/ton, equal Drs 10,800/.

- Electric power: Drs 7000/kWh /source: HIMIC/
- Additive for pigment making: Drs 220/kg /preliminary information obtained from owner of technology/
- Precipitant: Drs 300/kg
- Royalty on refractory sales: running at a rate of 2 %,

 /preliminary information obtained from owner of
 technology/
- Royalty on pigment sales: running at a rate of 5 % /preliminary information obtained from owner of technology/.

Table 7/6

Cost estimate for Grade '83' bauxite
/Flant capacity: 40.000 tons/

	Quantity (tons)	Price (Drs/ton)	Cost (CCC Drs)
Raw material procurement Transportation	78,000	5,670/1/	442,260 13,650 ^{/2/}
Loss sold rejected raw material	18,600	5, ₂₆₅ /3/	/97,929/
Raw material input	57 , 800	6,193	35 7, 981

^{/1/} Equal to USD 42/ton

^{/2/ 174.98} Drs/ton

^{/3/} Equal to USD 39/ton

The manning requirements for a shift in a 40.000 ton plant are given in Table 7/7. Based on a 3-hour working day and a five-day week, taking 20 working days as the average paid holidays and further 10 working days as lost, the continuous operation of the plant for 8000 hours a year makes it necessary to budget for 4.5 shifts.

The cost of man-power for the same plant is shown in Table 7/6. Wage and salary rates are based on HIMIC's information, cost-to-employer figures have been calculated on the basis of 14 monthly pays a year. Social security contributions are estimated at a rate of 35 %. In the larger plants two additional semi skilled workers are foreseen for each shift in the leaching/washing department, resulting in an additional cost-to-employer of Drs 9.36 million for each 20.000 ton capacity increment.

Factory overhead includes spare and wear parts estimated as 1.0 to 1.5 % of the total cost of fixed assets, as well as auxiliary materials and supplies. Administrative costs including cost of repair, communications, office supplies, etc. have been estimated as approximately 10 % of the salaries.

Depreciation figures have been calculated from the original value of fixed assets using the straight line method, applying a uniform rate of 9.25 %, derived from 10 % for equipment and 5 % for buildings. / Table 7/9./

Table 7/7
Hamming requirements

rocess / Location	g	8	ازبا	iali	iica	tion	k pla Code	es	with 1
Eauxite preparation		(1)							
Leaching/Nashing			1				1	1	
Dryin			1						
Calcination			1				1		
nOl regeneration						2			
Material handling			2						
Laboratory	3							1	
Haintenance and Repair					7				ı
Flant management				4				2	4
Total	3	25	16	Ľ,	7	9	9	4	5
Note: Number of jobs i	n cc	ntin	uos i	work.	-ord	er a	ppear	rs in	

Table 773
Laurever costs
Laurever in Laure 2000 rae as/

U y filig	2.3	gadification/lost	Colory /Conth	Auring
1	5	Lanagem e r t	110-250	20,000
9	PL	Desiratories/ Adaptatica	158-118	:
<u>z</u>		A gir sering webbbl	30	5 · ·
	3	in min se r t op Octobel	75	•
		e tillei halbar	70	•
		o de p r o cupal	of C	3 ,
		wente-villed lylor	9 0	
:	;	ari-Millad Imbor/ ozerk	55	<u></u>
		The illeriation	50	· · · •
	And the second s			1
7117 213	e de la Proposición de la Prop			= 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1

Table 7/9

Depreciation schedule /straight line depreciation/

	2 0/85	40/85	60/85	80/85	40/83
Auxiliary facilities	238	2 90	32 6	359	2 60
Process equipment	1 522	2 111	2 461	2 725	1 853
Building s	60	63	67	70	63
Licences	152	211	246	273	185
Capital expenditure	3 88	5 2 0	605	672	463
Total	2 360	3 195	3 7 05	4 099	2 824
Annual cost /9,25 %/	218	2 96	343	379	261
Residual value after 10 years	180	23 5	27 5	30 9	214

As can be seen from Table 7/5 the inputs with the largest relative cost are /a/ the raw material /24 to 31 % for Grade '85' and 42 % for Grade '83'/ and /b/ the cost of fuel /22 to 27 %/. Operating costs for a plant processing high grade /83/ material are some 20 % below those of the comparable Grade '85' plants, despite the former's higher bauxite cost.

If the white bauxite is left uncalcined, the estimated savings for 40,000 ton output areas follows:

electric power	million Drs 5.4
fuel oil	128
spare parts, supplies	8
labor	12.8
	154.2

The total cost of operation of the 40/85 plant is reduced to Ers 822 million, that of the 40/83 plant to Ers 693 million.

5. percuic and all amoial evaluation

In this complain which a sile years processing the court countries of the point of view or their estimated ecc unic and financial performances.

for a case in which the total production of refractor, that enter the calculate the calculate from the plant is exported, 700 tone of the iros emids pi nead is sold in treeds, the balance is emported. The calculation is based on maximum prices from a spicers / and 5 and uses 6 % export subsidy. The we do not be appear in 1251s 8/1.

Cost of operation taken from Table 7/9 and depreciation taken from Table 7/9 have Teen took to calculate the profitability of operation, shown in <u>Table 8/2</u>. Tay lack period is calculated using the total investment cost figures from Table 7/3 and allowing for the 25 % grant as per law 1202/82.

All plants except for the 20/85 variant make a profit in the maximum prices case. Plant 20/85 would break even if its revenue would increase by 3 %. If 75 % of the depreciation calculated in Table 7/9 is included in the cost of operation, 40/85 and 60/85 plants join 20/85 in making a loss rather than a profit. To break even 40/85 plant's revenue would have to increase by 12 %, 60/85 plant's by 2.5 %.

Table 8/1
Revenue /in million Drs/

Feed grade			85			83
Plant size	ton	20	4C	60	80	40
Refractory bauxite	19,292	38 6	772	1 158	1 543	_
100 % export)	28,196	-	-	-	-	1 128
Iron oxi de pigment 700 t so ld in Greece	28,000	2 0	2 0	20	20	20
the balance for export	29,680	1 41	304	466	6 2 8	151
Total		547	1 096	1 644	2 191	1 299

Table 3/2
Profitability of operation
Drs figures in millions

Variants	20/85	40/85	60/85	80/85	40/85
tevenue	547	1 096	1 644	2 191	1 299
Cost of operation less depreciation	5 6 5	97 5	1 385	1 790	847
Gross profit /loss/	/18/	120	259	401	452
Total investment cost less {rant /l/	1 851	2 537	2 980	5 35 3	2 242
Pay-back period /years/	-	21	11.5	8.3	5.C
Depreciation/Ty	135	254	2 98	333	224
Net profit//loss/ after /ك/	/203/	/134/	/35/	ં8	223

^{/1/} Grant: 25 % of Total Investment Cost /Law 1202/82/

8.2. Sensitivity

The effect of the price of the raw material, the single biggest item in cost of production, can be examined using the sensitivity curve of <u>Figure 8/1</u>. Feedstock price has been varied in a range of \pm 25 % of the median value. Gross profit of 40/85 plant increases by Drs 140 million to almost 4-fold over the \pm 25 % bauxite feed prices range. This effect is even more pronounced for larger plants.

The effect of the fuel price can be seen in <u>Figure 8/2</u>. A range from Drs 24,000, the price chosen for use in the calculation of the operating cost, to Drs 14,850 /USD 110/ has been selected. Gross profit of the 40/85 plant increases by Drs 100 million over the full fuel price range.

In <u>Figure 3/3</u> the effect of the iron oxide content of the feed on the cost of spent acid regeneration is shown. Cost of equipment was kept constant for the sake of simplicity. As iron in the feed gets higher decreasing cost of bauxite feed cannot balance increasing cost of regeneration /in-cluding cost of complementary HCl/.

Figures 8/4 and 8/5 show the effect of the selling price of the calcined refractory bauxite and the level of production on gross profits of the 40/85 and 40/83 plants respectively. Changes in the iron oxide pigment sales proceeds would have about half of the effect in the case of the 40/85 plant, and about one-seventh in the 40/83 plant, compared to those caused by changes in the proceeds from bauxite.

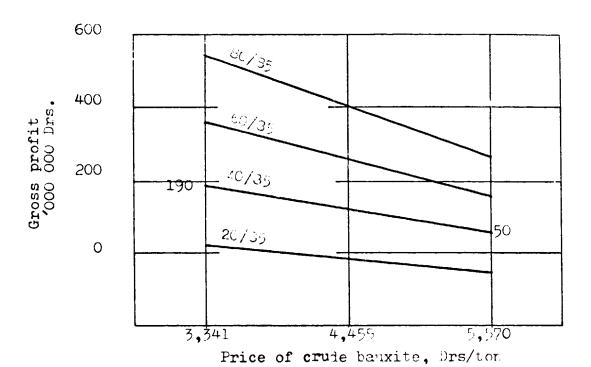


Figure 8/1: Effect of bauxite purchase pric on gross profit of plant, using Grade '85' feedstock

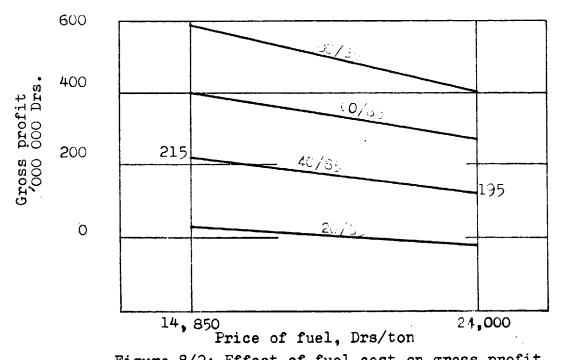
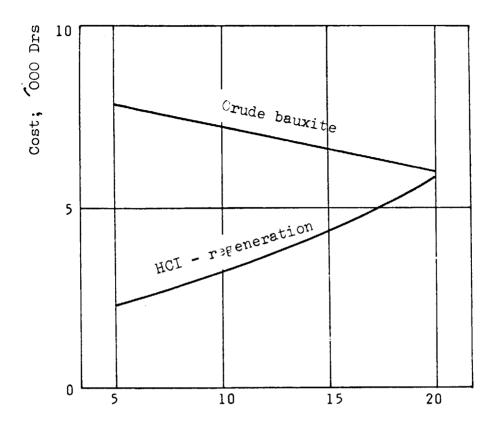


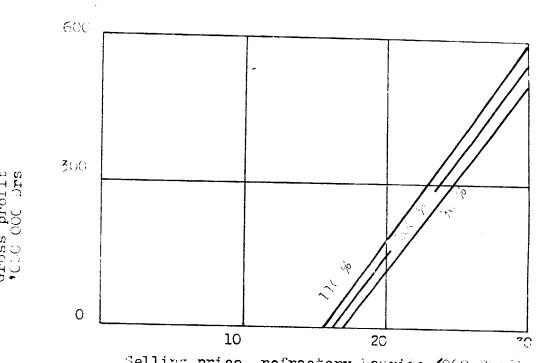
Figure 8/2: Effect of fuel cost on gross profit of plant using Grade '85' feedstock



iron oxide in crude bauxite; %

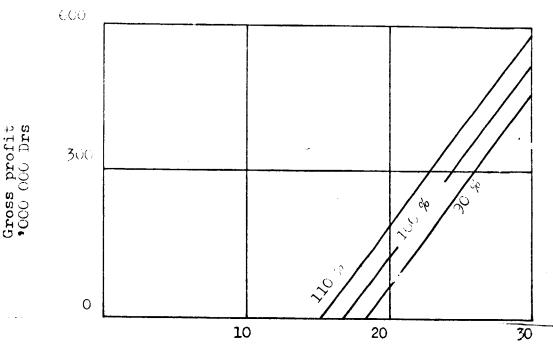
Figure 8/3

Cost of feed and cost of HCI regeneration vs iron oxide content of feed.



Selling price, refractory bauxite /000 Drs/ton

The effect of the selling price of the calcined refractory bauxile and of the level of production on the gross profit of the 40/85 plant



Selling price, refractory bauxite /000 Drs/ton

Figure 8/5
The effect of the selling price of the calcined refractory bauxite and of the level of production on the gross profit of the 40/33 plant

Figures 8/6 and 8/7 show at what level of production the 40/85 and 40/83 plant break even, respectively.

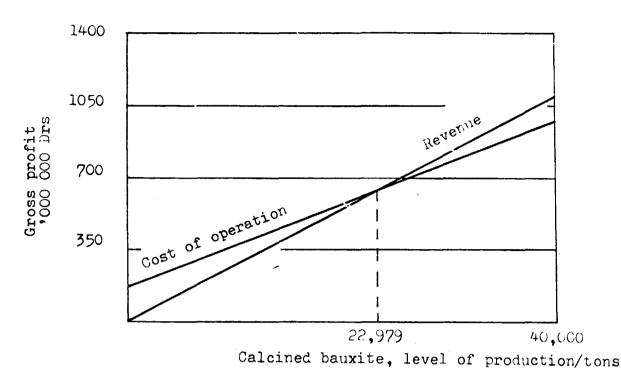
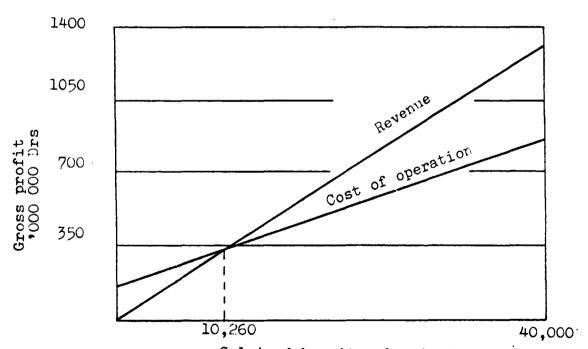


Figure 3/6 Break even point for the 40/35 plant



Calcined bauxite, level of production/tons
Figure 8/7 Break even point for the 40/83 plant

8.3. Financial evaluation

As apparent from the section 8.1 plants having an output capacity of 40, 60 and 80,000 tons show a gross profit when the products are sold at the maximum projected prices. Only the largest plant considered shows a net profit, however, under the same price conditions. The use of the medium grade bauxite to feed a beneficiation plant can therefore not be recommended unless fuel price and bauxite cost are significantly reduced.

In an attempt to demonstrate the level of economic viability to be expected from using the high grade feed, two financing models are considered. In one it is assumed that promoters and collaborators will provide 75 % of the capital requirement of the project, 25 % coming from a Covernment Grant. In the other model 35 % foreign loan is assumed, 40 % is brought in by the promoters/collaborators and 25 % is covered by Government Grant. The use of Greek loan is not considered because of its high interest cost.

Working capital is financed from a short term loan subject to 21.5% interest.

Net cash-flow and total net present value analyses for the first model appear in <u>Tables 8/3</u> and 8/4, for the second model in Tables 8/5 and 8/6.

Table 3/3
Projected income statement for the plant 40/83; financing model No. 1*
/values in <000,000 Drs/

Year	1	5	3	4	5	6	7	3	9	10	11	12	Residual value	Total.
Sales revenues	-	-	344	1 299	1 299	1 299	1 299	1	1 299	1 299	1 299	1 299	-	
Cost of operation	-	-	600	347	847	847	847	847	847	847	847	847	- !	ı
Gross Profit		-	244	452	452	452	452	452	452	452	452	452	-	
Depreciation	_	-	261	261	261	261	261	261	261	261	261	261	214	2 324
Taxable Income	-	-	/17/	191	191	191	191	191	191	191	191	191	-	
Taxes		-	-	34	92	92	92	92	92	92	9 2.	32	-	
Net profit	_	-	/17/	107	99	93	99	99	99	99	99	99	-	882
Net cash flow	_	-	244	3 63 (3 60	360	360	360	360	360	360	360	214	3 7 06

^{*} Total Investment Outlay: 2 989 Mio Drs

Grant : 25 % = 717 Mio Drs

Equity: 75 % = 2 242 Mio Drs

Tax Rate: 49 %, no tax holiday

Net cash flow: net profit plus depreciation

Table 8/4

Calculation of present value and internal rate of return on owners, equity /financing model
No 1/ for plant 40/83

/walues in <000,000 Drs/

	Year	1	2		3	4	5		6	7		8	9	10	11	12	Salvago' value	Tota
Α.	Cash inflow 1. Sales revenue	•	•		844	1 299	1	2 99	1 29 9	1 2	299	1 299	1 2 99	1 2 99	1 299	1 299	_	
в.	Cash outflow /1+2+3/	900	1	342	600	931		939	939	9	39	93 9	939	939	939	939	/750/	
	1. Total inv. outlay	900	1	342	-	-		-			-	-	-	-	-	_	/750/	
	2. Operating cost	-	-		600	847		847	847	8	347	847	847	847	847	847	_	
	3. Taxes	-			-	84	·	92	92		92	. 92	92	92	92	92	-	
c.	Net cash flow /A-B/	/900/	/1	342/	244	368		360	360	3	360	360	360	360	360	360	750 1	253
D.	Present value ³	/900/	/1	21 3/	200	273		242	220	1	199	180	163	147	133	121	227	- 8

Notes; 1. Frices changes, appreciation not considered.

2. Salvage value: Land: 17,

of buildings: 200, of plant equipment: 375, working capital: 165

3. Present value: discounted at the internal rate of return of 10.5 %

- 140

Table 3/5
Frojected Income Statement for the plant 40/83; financing model No.2*
/Values in voo.000 Drs/

												.		1
Year	1	2	3	4	5	6	7	3	9	10	11	1.2	Residual value	Total
Sales revenues	-	-	344	1 299	1 299	1 293	1 299	1 299	1 299	1 299	1 299	1 299	-	
Cost of operation	-	-	600	847	847	847	947	847	347	847	847	8,47	_	
Gross profit	_		244	452	452	452	452	452	452	452	452	452	-	
Interest on Forex loan	_	_	-	74.6	65	55.4	45.8	36.2	26.5	16.8	7.2	-	•	
Interest on short term loan	-	_	3 5•5	35.5	35.5	35.5	35.5	35.5	35.5	35.5	35•5	35.5	-	
Depreciation	_	-	261	261	261	261	261	261	261	261	261	261	214	5 854
Taxable Income			/52.2/	45.4	90.5	100.1	109.7	119.3	129.0	133.7	143.3	155.5	•	
Tax .	_	-	_	21.3	43.4	43	52.7	57.3	62	66.6	71.2	74.6	-	
Net profit			/52.2/	59.1	47.1	52.1	57.0	62	67	72.1	77.1	30.9		522
Loan principal repayment	-	_	_	137.4	137.4	137.4	137.4	137.4	137.4	137.4	137.4	-	-	1 099
Net cash flow	-	-	203.8	132.7	170.7	175.7	130.6	135.6	190.6	195.7	200.7	341.9		2 247
								<u> </u>						

*Total investment cost: 2 834 Nio Drs

Working capital: 165 Mio Drs

Grant: 25 % = 703 Mio Drs

Forex loan: 35 % = 992 Nio Drs

Equity: 40 % = 1 134 hio Drs

Interest rate of Forex loan: 7 %, Repayment in 16 semesters; grace period of two semesters

Interest rate for short term loam: 21.5 %

Tax Rate: 4'%, no tax holiday

Hot card, flow: set profit plus depressas

Notes:

5°

					-										
0 0 0 0 0 0	Year	1	2	3	4	5	6	7	8	9	10	11		Sal- vage ² value	100 1000
Table 8, Table 8	A. Cash inflow 1. Sales revenue	-	-	844	12 99	1299	12 99	12 99	1299	1299	1299	1299	1299	-	
	<pre>B. Cash outflow /1+2+3/</pre>	400	734	635.5	1116.3	1128.3	1123.2	1118.4	1113.4	1108.4	1103.4	1098.4	95 7. 1	/7 50	
3.] 4. E	1. Investment costs													/750	/
Dis of Rou	a/ Equity funds	400	734	-	-			_		_	_	_	_		İ
count retur nded	b/ Forex loan principal repayment	_	_	-	137.4	137.4	137.4	137.4	137.4	137.4	137.4	137.4		_	1099
at of guz	c/ Interest on Forex loan	_	-	-	74.6	65	55.4	45.8	36.2	26.5	16.8	7.2	-	_	10)
the 14.0	d/ Interest on short term loan			75 5	75 5	-	7.5								
% F.	10an			35.5	35.5	35.5	35.5	35.5	35. 5	35.5	35. 5	35.5	- 35.5	-	
tea	2. Operation cost	-	-	600	847	847	847	847	847	847	847	847	847	-	
internal %	3. Taxes	-	-	_	21.8	43.4	48	52.7	57.3	6 2	66.6	71.2	74.6	-	
	C. Net cash flow A/A+B/	/400/	/754/	209	183	171	17 6	181	186	191	1 96	201	342	7 50	1650
	D. Present value ⁴	/400/	/644/	160	123	1 01	91	82	74	67	60	54	81	155	

on the owners' equity results. Since the Greek system of investment promotion offers no tax holiday the profitability of the venture can be improved if outside capital available at relatively low cost is used. Under the selected financing model including outside financing the IRR calculated from the discounted cash flews is 14.0 %. This together with the pay-back period of 5 years appearing in Table 8/2 make the beneficiation of high grade Greek bauxite by the TATABANYA leaching/acid recovery technology an interesting idea to pursue.

The main reasons for the superior economic and financial performance of the 40/83 plant over 40/85 are as follows:

- leaching is more efficient as bauxite feed has more favourable minerology,
- less hydrochloric açid is lost as Ca+Mg level in feed is lower,
- material flows are lower.
- sales revenue is higher, despite lower pigment output, as top grade calcined refractory grade bauxite can be produced.

8.4. Rational economic evaluation

The production of calcined deironed bauxite from raw material available in Greek would have the following positive effects on the Greek national economy:

- Utilisation of local mineral resources for products with value added possible.
- Annual imports of some 10-15 thousand tons of refractory bauxite worth some 1.5 million US Dollars substituted by low import content local product.
- Exports potential of a few million US Dollars a year created for either refractory grade calcined bauxite, or refractory product made of it.
- Employment for 70 to 100 persons in a modern plant.

List of references

- 1. Greek Patent No. 65 874 issued to Tatabánya Coalmines and Necsek Ore Mines
- 2. Technical Report DP/GRE/83/005 dated 13 Nov. 1984, by Dr. D. Lack
- 3. Interim Report DP/GRE/83/005 dated October, 1985, by Dr. Riederauer et al.
- 4. Guyana Bauxite profile of a visit. Industrial Minerals,
 December 1984, p.53
- 5. bauxite China snaps at Guyana's heels.

 Industrial Minerals Refractory Survey, 1981, p.83
- 6. Quotation by the Shanghai International Frading Co.
- 7. Same as item 4
- 8. Iron oxide pigments
 Industrial Minerals Pigments and Extenders Supplement,
 May 1985, p.25
- 9. Minutes of the tripartite discussions held in Humpary, 22-25 October 1986