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# 12951

# MODEL BASIC ENGINEERING DESIGN

## FOR A

100 MTPD AMMONIA PLANT

# FINDINGS & CHARACTERISTICS OF 100 MTPD AMMONIA PLANT

UNIDO CONTRACT No. 82/48 PROJECT No. UD / INT / 82 / 101 ACTIVITY CODE: UD / INT / 30.3

Prepared for

## UNITED NATIONS INDUSTRIAL DEVELOPMENT ORGANISATION

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(JUNE 1983)



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## $CHAPTER-1$

#### 1. INTRODUCTION

1.1 At the NFC-UNIDO Seminar on MINI-FERTILIZER PLANTS, held at Lahore on November· 15-20, 1982, FEDO had presented a design document for a 100 Metric Tonnes/day Ammonia Plant. A modular concept had been adopted for the design using established process steps. In line with the logic of long-term availability of raw-materials in the majority of countries, two designs based on Coke and Fuel Oil as raw-materials and with many common elements were presenteJ in the document. The reasons for the choice of these two raw-materials had also been explained.

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- 1.2 The discussions at the Seminar favoured a modified-approach for the final design study. Based on current plant experience and the expected availability of small pockets of Natural Gas. in many developing countries, it was decided that designs based on Coal and Natural Gas also be done to supplement the available study on Fuel-oil. This has now been done, adopting fresh approaches from those previously followed. Two basic designs, one based on Natural gas and another based on Anthracite Coal are presented in this document. In effect, taking both documents together, UNIDO have four different designs and using the modular principle, many possible variants.
- 1.3 The very conception of mini-plants has been born out of the recognition of fertiliser as an important input in the economic evolution of Developing Countries and the uncertainties affecting supplies and prices following on the oil price crisis of 1973-75. The heavy impact of higher transport charges on prices favours local manufacture. To many countries the introduction to sophisticated industry could be through local mini-plants which will help to up-grade technical capabilities and promote better agriculture.
- 1.4 The rising price of energy has focussed attention on energy consumption per unit of fertiliser output. Large single stream plants are favourably placed on this score, since the effective gathering of fugitive waste heat is more easily done in these units. Most plants are self-contained in power requirement by adoption of designs which use process heat for meeting this need. A matching of available heat with the imperatives of design of principal machinery confine these achieved high efficiencies to the larger sizes of plants. A limitation on energy efficiency is placed on Mini-plants by the need to depend on purchased or generated electric power for the principal drives. Direct comparisons of energy consumption per unit of product are therefore not in order. However, it may be noted that a 1979 Study on Ammonia plants in North America\*, where a large number of the most advanced Ammonia synthesis plants are installed, shows an energy consumption of  $11.209 \times 10^6$  kilo calories per Metric Tonne of Ammonia. This figure may be taken as a fair index for comparison of new designs.
- 1.5 In presenting the first document the logic which governed the selection of preferred raw-material for a design of plant for wider application had been discussed (para 3.1 to 3.4). The assumption of the availability of local pockets of Natural gas based on the fairly laxge number of countries with potential (listed by BEICIP, a Bureau of the Institute Francaise du Petrole in 1980) is the logic for the Natural gas alternative presented here. It is assumed that many of these deposits are recoverable economically for local consumption and a price of about  $$4$  per 1000 cft. has been assumed in the costing. Anthracite Coal is a commodity in abundant supply and is the raw-material assumed for the second design.

[\*ENERGY AND FERTILIZER : IFDC TECHNICAL BULLETIN IFDC-T-19)

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# 2. PROCESSES and the set of the set

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#### 2.1 Ammonia Synthesis

The design of the Ammonia Synthesis module is common for both alternatives. From the receipt of purified Synthesis Gas till the delivery of liquid AMmonia into storage is the compass of this module. In order to reduce the number of rotating machinery, the simplest Ammonia process has been used, namely, the Casale Process which requires only the Synthesis Compressor. The system operates at 400 kg/cm2 which does not present any problems of availability of equipment at the small size of Plant being considered. No refrigeration machinery is involved. Circulation is by ejector which is trouble-free. Maximum recovery of Ammonia is effected by scrubbing the Purge gases and distilling off the Ammonia. The residual purge gases are available as fuel for process or off-sites use.

#### 2.2 Synthesis Gas - based on Natural Gas Reforming

- 2.2.1 The design adopted is based on the orthodox flow-sheet of Natural Gas Reforming Plant. Natural Gas at 40 kg/cm<sup>2</sup> is preheated before passing to sulfur removal by Zinc Oxide. The gas is then mixed with steam and reformed in two steps, primary and secondary. Heat recovery is effected from the flue gas by heating process streams and combustion air. The heat in the process gases is used to raise steam for process purposes. There will be a surplus of steam which can be used in offsi tes.
- 2.2.2 CO conversion in two steps, High Temperature & Low Temperature reduces CO in the process gas stream by combining CO with steam to give volume for volume conversion to Hydrogen. The gas stream which contains Hydroyen and Nitrogen as well as  $CO<sub>2</sub>$  is then processed for removal of  $CO_2$ .

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 $2.2.3$ The Benfield Process, which is widely accepted has been adopted in this case for the  $CO<sub>2</sub>$  removal step. Physical absorption processes like Selexol Process are available: but are yet to make an impact and hence choice of the more widely adopted Benfield Process. The bulk of  $CO<sub>2</sub>$  in the gas stream is removed in this step. The output gas stream is essentially in the ratio needed for Ammonia Synthesis and needs only the removal of the small fraction of Carbon Oxides still left. This is done by methanation which converts carbon oxides to process-neutral methane. The gas is now ready for compression which is the first step in the synthesis module.

#### 2.3 Synthesis Gas - based on Anthracite Coal

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- 2.3.1 Even though Coal and Coke were the raw-materials used at the beginning of the Ammonia industry, these widely available raw-materials were replaced by the more easily handled Petroleum derivatives, Naphtha and Fuel Oil where Natural Gas was not available. The changing pricing structure of fuels provokes a revision of attitude to traditional fuels. There are several popular processes for their use in gas making which have made their strong presence in improved versions in recent years. For use in Mini-Ammonia Plants, where simplicity and dependability are important criteria, the Wellman Gasifier using enriched air is being proposed. The use of PSA unit for providing the enriched air is a simpler alternative to the cryogenic units, normally used for Partial Oxidation Processes.
- 2.3.2 The PSA unit uses molecular sieves and the pressure swing principle to provide a supply of enriched air with oxygen content of 46% which will ensure a properly proportioned Synthesis Gas at the end of the gas making module. Since the Gasifier itself operates at near atmospheric pressure the enriched air is delivered at low pressure.
- 2.3.3 The Wellman Gasifier is a well established equipment which has proved itself in many installations. Multiple units- are used (three in this case) sc that even in the event of maintenance of one unit, Gas supplies to the synthesis plant is not interrupted. The gas leaving the producer is scrubbed free of dust and higher hydrocarbons and sent to a gas holder which serves to even out process flews.
- 2.3.4 The gas has to be scrubbed free of any sulphur content before further processing. The Benfield Hi-pure Process is used after compressing the gas to about 35  $kg/cm<sup>2</sup>$ . Since the Benfield Process is used after CO-conversion to remove CO<sub>2</sub> from the gas stream, the Hi-pure Process fits well into the general scheme. The compressing of gas to the intermediate pressure enables the purified gases to be delivered to the synthesis module at the same pressure as in the case of the Natural Gas alternative.
- 2.3.5 CO-conversion is carried out in High Temperature and Low temperature Stages. The Carbon Di-cxide formed is removed in the absorption towers of the Benfield Process and the gas passes on to Methanation. With carbon oxides reduced by conversion to Methane to a residual of less than 10 ppm. the gases enter the suction of the synthesis compressor.
- 2.3.6 The Gasification module is tailored around the Wellman Gasifier which is considered simpler and easier to operate than other coal gasification equipment. It will be noted that unit production costs are higher in this case in spite of lower raw-material costs largely on account of electricity and steam costs as well as depreciation. The electricity and steam consumptions

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suggest a Co-generation cycle Power Generation at the site itself thereby making substantial savings due to higher energy efficiency.

#### 3.0 DOWNSTREAM PLANTS

7he recommendation made regarding down-stream plants in Chapter-I - para-7 of the earlier document is<br>retained. Product costs have been presented based on new costing of Ammonia.

#### 4.0 GENERAL

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- 4.1 In presenting revised designs the following requirements i.ave been met.
- $4.1.1$ Alternate designs for the use of Coal and Natural Gas have been prepared as required. The modular pattern enables the substitution of part of the Fuel Oil design earlier presented to combine with the new Ammonia Module.
- 4. 1. 2 The design for Natural Gas processing follows classical steps since none of the newer process steps publicised have been established yet and the savings indicated are marginal and still to be proven. The design for coal has used the simplest and fairly widely accepted Gasifier in combination with a standard PSA oxygen enrichment unit. Benfield Process has been adopted for both routes as it is presently the most widely accepted purification step. The Benfield Hi-pure Process for removal of sulfur incorporates easily into the system.
- 4. 1. 3 The generation of power has to be done in an off-sites unit. The surplus steam from the Natural Gas Reforming process and the requirement of power and steam at low and

medium pressure in the Coal Gasification Process can be suitably integrated with the off-sites unit. This may be done more effectively in specific cases since locational factors are involved.

- 4.1.4 Investment and production costs based on a location in India have been presented.
- 4.1.5 Two Gasification modes and a Synthesis Process have been provided in this document. Each of these processes can be standardised with limited modifications made for local factors. Moreover the modules within each group can be replaced if an improved process becomes available. Particular attention has been paid bo choosing the most easily operable process units and schemes.
- 4.2 Energy Efficiency
- 4.2.1 Quoted below is an extract from IFDC Bulletin "ENERGY & FERTILIZER":

"There do not seem to be any commercially feasible technological breakthroughs in Nitrogen manufacturing that would result in major energy saving".

What this means is that while further energy savings are desirable, those publicised are at best may . ginal and unlikely to affect the perceived advantages of economy of local manufacture. Quoting again from the same source "For example, operation of an Ammonia Plant at 60% capacity may use 25% more energy per tonne of product than continuous operation at full capacity", it is clear that the utmost emphasis needs to be paid to design features which make uninterrupted operation possible.

4.2.2 In the larger plants with complex process linkages, it has been observed that a single interruption in a month increases the monthly average consumption of energy by as much as 10%. The ability to operate for long periods without interruption, engineered with comparative ease into the simpler designs presented in this document may therefore be their best recommendation of Mini-Plants for developing countries.

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#### MINI AMMONIA PLANT

#### **PROCESS**

#### 1. Natural Gas steam reforming

- 1.1 Design basis: 100 MT per day of Anhydrous Anmonia
- 1.2 Raw Materials: Natural gas of following composition by volume & equivalent to make up gas required for  $100$  MT per day of Anhydrous Anmonia:



#### 1.3 The Plant

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- 1. The feed natural gas of above canposition is assumed to be available at the batter $_{Y}$  limit at 40 Kg/cm $^{2}$ g pressure and ambient temperature. Natural gas is used for fuel use also.
- 2. Auxiliary steam raising facility has been provided as a buffer to meet the steam requirement in the plant during start-up and other abnormalities. The plant is self sufficient in steam requirement at normal plant load.
- 3. Although only Zinc Oxide has been, at present, provided for the feedstock sulphur removal, provision has been made for Comox catalyst bed also for use, if required.
- 4. Benfield process has been adopted for the decarbonation of the make-up gas.

#### 1.4 \_ Plant Description (Drg. No. 9484-237-11120)

The natural gas at 40 Kg/cm $^2$ g pressure at the battery limit is freed of liquid condensate in the knock out drum 301 and is preheated to 400°C in the preheater 581 for the feed gas desulphurisation. The natural gas for fuel requirement is tapped from the main stream prior to the preheater but after the liquid knock-out drum. The feed gas at 400°C from the preheater is passed through the ZnO bed in the desulphurisation vessel 302. The desulphurised feed is then mixed with the required quantity of process steam to make up a steam carbon ratio of  $3.5:1'$  at the Primary Reformer inlet. preheated again to 510°C in the Primary Reformer convection train and fed to the primary reformer 582. The Reformer is a conventional top fired furnace with 64 numbers of centrifugally cast tubes of HK 40 alloy and 40 numbers of gas fired burners in the radiant box. Following equipments are housed in the convection train of the reforming furnace :

1) Feed Preheater

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- 2) Process Air Preheater
- 3) Process Steam Superheater
- 4) Flue Gas Boiler and
- 5) Combustion air preheater

The reformer effluents at about 820°C and a methane slip of 10.78% is further refonned in the Secondary Reformer 303. This is an adiabatic reactor wherein required quantity of the preheated process air at 470°C is admitted and further reforming of the Methane takes place at a higher temperature than in the Primary Reformer.

The Secondary Reformer effluent at 1000°Cwith a methane content of 0.33% is then  $\infty$ oled to 380°C in the Reformed Gas Boiler 502. The hydrogen content in the gas is further erhanced by conversion of the carbon monoxide to carbon dioxide in presence of steam in the high and low temperature shift reactors 305 and 306 respectively. The gas is fed to the high temperature Shift Converter 305 at 380°C.

The exothermic heat in the HTS effluent is removed in the converted gas boiler 503 and process heat exchangers 504 and 505 and the temperature reduced frcm 440°C to 223°C at the inlet of the low tenperature Shift Reactor (LTS) 306.

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The carbon monoxide content in the gas exit of the LTS is 0.3% and carbon dioxide about 17.9%. The heat content in the gas is used for the reboiler duty in the decarbonation section based on the Benfield process.

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The carbon dioxide in the synthesis make up gas is reduced to 0.1% in the decarbonation section. The solutkon is 30% potassium carbonato, 3% DEA and Vanadium Pentoxide as inhibitor. SS 304 slotted rings are used as the tower packing.

The carbon oxides in the gas exit the decarbonation section are converted into Methane in the Methanator 311. The total inerts in the make-up gas after the Methanator will be 1.15% consisting of Argon 0.32% & Methane 0.83%.

The gas at 24 kg/cm<sup>2</sup>g pressure and 45°C is sent to the ammonia synthesis section as make-up gas.

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#### COAL GASIFICATION: SECTION  $2.$

- 2.1 Design Basis: 100 MT per day of Anhydrous Ammonia.
- 2.2 Raw Materials: Anthracite coal of following proximate analysis :



#### 2.3 The Plant:

Pressure Swing Adsorption (PSA) System has been included in the scheme for meeting the requirement of enriched air with 46% oxygen for the gasifiers and also to ensure required quantity of nitrogen in the synthesis make-up gas.

- 2. The gasifier is a fixed bed type operating at near atmospheric pressure.
- 3. A gas compressor has been incorporated to boost the gas pressure from atmospheric to medium pressure of 35 kg/cm $^2$ g prior to the gas purification section.
- 4. Hi-Pure Benfield process has been considered most suitable for the gas desulphurisation and decarbonation processes.
- 5. Final purification of the gas is *b'J* Methanation where the carbon oxides are reduced to less than 10 ppn level.

## 2.4 Process Description [ Drg.No.9484-237-11121 J

The production of raw gas is by using Wellman Producer. The gas is produced by controlled partial oxidation of coal in the fixed bed type producer in an atmosphere of oxygen provided by enriched air. Steam is used to control the canbustion. The gas generated contains mostly carbon monoxide and Hydrogen along with Nitrogen, carbon dioxide as well as solid impurities. Major reactions in the producer are

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Three gasifiers, 583, are used in parallel for the production of the raw gas. The coal is fed from a hopper situated over the gasifier. The gasifier consists of a water jacketed retort, a two compartment fuel bin with segmental gates, force feed pipes, rotating grates, ash bin etc. The upper compartment of the fuel bin serves as a storage bin while lower compartment functions as sealed continuous fue! feed system. Coal from the gas tight lower bin flows continuously through the feed pipes to fill the retort. Oxygen required for the gasification is supplied through the enriched air fran the PSA system and the steam, supplementing steam that is generated in the jacket is supplied from Offsites. The gas leaves the gasifier from the top at about 300 $^{\circ}$ C. The ash as and when formed is discharged below through. the rotating grates.

The gas leaving the gasifier is sent to the water scrubber 312 to remove solid impurities. Scrubber circulating pumps 602 maintain water circu-· lation in the scrubber. The condensate in the cooled gas from the scrubber is knocked out in the knock out drum 313 and stored in the gas holder  $402$ .

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The gas from the gasholder is compressed to 30 to 35 kg/cm $^2$ g pressure in the gas compressor 754 prior to its further purification processes. About 10% of the carbon dioxide and  $0.2$ % of sulphur in the raw cooled and compressed gas is removed in the Hi-pure Benfield section where the gas is washed with DEA & Potassium Carbonate solution in separate columns. The gas exit the section will be free of Carbon Dioxide and Sulphur.

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The carbon monoxide in the gas exit the desulphurisation section is converted into carbon dioxide in the High (HTS) 316 and low (LTS) 317 temperature shift converters, which operate respectively at 370°C and 210°C. Additional steam at 14.4 TPH rate will be required for the shift reaction in both the HTS & LTS, which together bring down the Carbon Monoxide content from 43% to 0.2% and the Carbon Dioxide builds upto 30%.

The Carbon Dioxide is removed in the decarbonation section, where 30% carbonate solution with 3% DEA and Vanadium Pentoxide constituting the Benfield solution is used. The carbon dioxide in the gas exit, this section will be  $0.1$ .

The carbon oxides in the gas are finally removed in the Methanator 320 where they are catalytically converted into Methane. Make up synthesis gas containing about 1.92% inerts, consisting of 0.89% Methane and 1.03% Argon, is then sent to the ammonia synthesis section.

#### AMMONIA SYNTHESIS SECTION  $3.$

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#### 3.1 Process Description [ Drg. No.9484-237-11107 J

The synthesis gas after the final purification in the Methanator is compressed in the synthesis gas compressor (755) to 450 kg/cm<sup>2</sup> pressure. The gas after oil separation in the oil filter (325) is then sent through the ejector (921) where part of its pressure energy is used up in boosting the recycle gas to the converter inlet pressure of  $400 \text{ kg/cm}^2$ .

The synthesis system used is as per Casale design. The synthesis make up gas mixed with the recycle gas in the ejector enters the converter  $(3/6)$ . Inlet gas heating is by exchange of heat with the converter effluent which leaves the converter at about 260°C. The gas enters the catalyst bed at 370°C. The exothermic reaction takes place to produce ammonia

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R_2 \times N_2 + 3/2 H_2 \longrightarrow NH_3 + \Delta H_{25} \quad -22.2 \times 10^3 \text{ Kcal/kg mole N}_2 \text{ reached.}
$$

The converter effluent at 260°C is cooled with cooling water in the Ammonia condenser (530). The bulk of the Anmonia in the gas stream is condensed in a single stage in this condenser.

The condensed Ammonia is separated in the separator(327) and the recycle gas after a purge, is drawn to the recycle ejector, mixed with the synthesis make up gas and fed to the converter.

The anmonia in the separator is let down to 20 kg/ $\mathrm{cm}^2$  in the let down vessel (328). The dissolved gas released in the let dam vessel joins the purge gas and goes to the ammonia recovery section for ammonia recovery. The gases scrubbed free of Amronia is used as fuel.

The purge gas from synthesis loop and the flash gas from let-down vessel are sent to the scrubber (329) where the Anmonia is scrubbed with water. A part of the ammonia solution is fed to the middle section of the scrubber through a solution oooler (531) as circulating solution and the remaining solution is fed to the stripper (330).

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In the Ammonia Stripper, the Ammonia solution is counter-currently heated with steam admitted at the bottom of the tower. The overhead gaseous product of the stiipper is 99.5%  $M_{3}$  in vapour phase and this is condensed by a reflux condenser (534) and collected in a separator (331). A part of it is used as reflux and the remainder is sent to the let down The bottom product of the stripper is vessel as recovered ammonia. sent to the scrubber via solution exchanger (533) as absorbing fluid.

#### CHAPTER III

MINI AMMONIA PLANT

**TECHNICAL DATA** 

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# List of Equipment with outline specifications

# 1. Natural Gas Steam Reforming Section (Ref:Drg.No.9484-237-11120)



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2. COAL GASIFICATION SECTION (Ref.Drg.No.9484-237-11121)

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51. No.	Item	Description	
1.	325	Make-up Gas Oil Separator	$1$ No.
		Diameter	350 mm
		Height	2000 mm
		Internal	filter elements.
2.	326	Ammonia Converter	$1$ No.
		Operating Pressure	350 - 400 kg/cm <sup>2</sup> g
		Internals	Heat Exchangers,
			Electric Heater,
			Cooling Tubes &
			Catalyst. 3.2 $\mathbb{M}^3$
5.	327	Ammonia Separator	$1$ No.
		Diameter'	500 mm
		Height	4000 mm
4.	328	Ammonia let down Vessel	1 No.
		Diameter	1250 mm
		Height	3800 mm
5.	329	Scrubber	1 No.
		Diameter	350 mm
		Height	16500 mm
		Packing	25 mm SWRR 1.2 $M^3$
6.	330	Stripper	$1$ No.
		Diameter	750 mm
		Height	20000 mm
		Packing	25 mm SWRR 4.5 $M^3$
7.	331	Separator	1 No.
		Diameter	650 mm
		Height	1200 mm

3. APPIONIA SYNTHESIS SECTION (Ref: Drg.No.9484-237-11107)

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with make up gas  $12,550$ <br>NM<sup>3</sup>/hr. at 450 kg/cm<sup>2</sup>g.

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#### CHAPTER-IV

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#### 100 TPD MINI-AMMONIA PLANT:

#### Capital and Operating Costs

Capital cost and operating cost for 100 MTPD Annonia and matching end-product plant Calcium Ammonium Nitrate have been worked out and given below on the basis of Anthracite Coal and Natural Gas as feed-stocks for Ammonia production. In both cases, the cost of associated facilities have been added to the main plant costs. The plant site assumed is a typical Indian location (Udyogamandal). It is assumed that the Site is a developed one with infra-structural facilities like road/rail access, power and water supply upto batterylimits. The rates given at the end of the estimate list out the various assumptions underlying the estimate.



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Table-1

#### 100 MIPD AMMONIA PLANT

#### Summary of Capital Cost

Scope in this battery-limits:

- a) Coal based plant Feedstock gasification, Benfield desulfurisation, CO conversion, Benfield CO2 removal, Methanation and Ammonia synthesis, PSA for enriched oxygen supply.
- b) Natural Gas based plant - Steam reforming, CO conversion, CO2 removal, methanation and Ammonia synthesis.



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## 100 MIPD AMMONIA PLANT

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Table-2

# Cost of production and Sensitivity Study

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Base Case - 90% Production level



## 3. Sensitivity analysis





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 $=-26.760$ 7. Total Investment ...

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 $Table<sup>4</sup>$ 

#### END PRODUCT PLANT (CAN)

#### Cost of production and Sensitivity study

## Base Case - Production from NH<sub>2</sub> availability at 90% level



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#### 100 MIPD MINI AMMONIA PLANT

Assumptions on Cost Estimate

- $1.$ Capital Cost
- $1.1$ Equipment cost and materials are assumed at Indian Prices (Jan. 83)
- $1.2$ Spares @ 10% of equipment cost have been included.
- $1.3$ Transport costs include 15% towards transportation and associated expenses (marine and inland).
- $1.4$ Site erection and civil works costs are based on prevailing rates at Udvogamandal (India).
- $1.5$ No duties or taxes are included in the estimates.
- 1.6 Off-site facilities included are Power Receiving and Distribution within plant area. Water Treatment, Steam Generation, Cooling towers, Raw-material receiving storage and handling, Effluent Disposal, Workshops, Laboratory, General Stores and Administrative buildings.
- $1.7$ It is assumed that developed land with infrastructural facilities is available and hence no cost has been provided towards land and development, utility supply upto battery limit, rail road access and  $\ddot{\phantom{0}}$ township.
- 1.8 Contingency has been provided @ 10% of fixed capital.
- $1.9$ As method of financing is not decided, interest during construction has not been considered in the capital costs.
- 1.10 The effect of inflation during construction has also not been considered. The impact of a 10% increase in cost has been shown in the sensitivity study.
- $1.11$ In working capital, the following are included:
	- i) 30 days' stock of coal feed for Ammonia (no storage of N.G.)
	- ii) 5 days' stock of Ammonia

iii) 15 days' stock of end-product fertilizer CAN

- $2<sub>r</sub>$ Cost of production
- $2.1$ Plant availability:

Number of stream days considered for this study is 320. Base case is taken at 90% production level of Ammonia. End-product capacities are fixed on the basis of Ammonia availability.

 $2.2$ For this study, the following feed-stock prices are assumed:



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- 2.3 Utility prices assumed are  $$0.05/M^3$  for water and  $$0.05/kWh$ for power commensurate with the prevailing rates in Udyogamands  $(India)$ .
- 2.4 Maintenance is taken  $\theta$  3% of erected plant  $\cos t$ .
- 2.5 Average salary is taken at \$2000.- per year per person. Overheads have been added at 100% of salaries.
- 2.6 Straight-line depreciation  $e$  10% is taken for computing cost of production.
- 2.7 No interest on capital is assumed.

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- 2.8 Sensitivity study gives the cost of production with 10% return on total investment.
- 2.9 Sensitivity study also gives the impact of 10% change in production level, 10% increase in capital cost and 10% increase in feed-stock cost.





SECTION 2



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DECARBONATION SHIFT CONVERSION  $\frac{1}{2}$   $\times$   $\frac{1}{4}$ 







SAS<br>ER

 $\mathbb{Z}^2$ .

 $\frac{1}{\sqrt{2}}$ 

 $\ast$ 

SYNTHESIS  $-\left( 6\right)$ 











 $\mathbf{r}=\mathbf{r}$ 





 $\mathbf{u} = \mathbf{u} \times \mathbf{u}$  ,  $\mathbf{u} = \mathbf{u}$ 

DESIGN DRAWN  $CHECKE$ APPRO  $\Box$ ATE SCALE





 $\mathbb{R}^2$ 

 $\mathcal{L}^{\text{max}}$ 

1

# SECTION 7







 $\hat{\mathcal{L}}$ 

 $\mathcal{L}_{\text{max}}$  , where  $\mathcal{L}_{\text{max}}$ 





 $\overline{\phantom{a}}$ 

 $\bar{1}$ 



 $\bar{\rm I}$ 



