



TOGETHER
for a sustainable future

OCCASION

This publication has been made available to the public on the occasion of the 50th anniversary of the United Nations Industrial Development Organisation.



TOGETHER
for a sustainable future

DISCLAIMER

This document has been produced without formal United Nations editing. The designations employed and the presentation of the material in this document do not imply the expression of any opinion whatsoever on the part of the Secretariat of the United Nations Industrial Development Organization (UNIDO) concerning the legal status of any country, territory, city or area or of its authorities, or concerning the delimitation of its frontiers or boundaries, or its economic system or degree of development. Designations such as “developed”, “industrialized” and “developing” are intended for statistical convenience and do not necessarily express a judgment about the stage reached by a particular country or area in the development process. Mention of firm names or commercial products does not constitute an endorsement by UNIDO.

FAIR USE POLICY

Any part of this publication may be quoted and referenced for educational and research purposes without additional permission from UNIDO. However, those who make use of quoting and referencing this publication are requested to follow the Fair Use Policy of giving due credit to UNIDO.

CONTACT

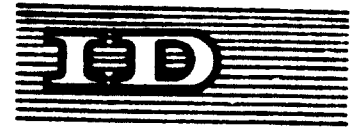
Please contact publications@unido.org for further information concerning UNIDO publications.

For more information about UNIDO, please visit us at www.unido.org

We regret that some of the pages in the microfiche copy of this report may not be up to the proper legibility standards, even though the best possible copy was used for preparing the master fiche



07563



United Nations Industrial Development Organization

Distr.
LIMITED
ID/WG.223/4
18 January 1977
Original: ENGLISH

Regional Symposium for Asia and the Pacific
on the Production and Promotion of Pesticides
and on Sub-Regional/Regional Co-operation in
the Pesticide Industries

Bangkok, Thailand
1 to 7 February 1977

METHYL PARATHION
AN INDUSTRY PROFILE^{1/}

by

C. L. Dhawan*

* Consultant

^{1/} The views and opinions expressed in this paper are those of the author and do not necessarily reflect the views of the secretariat of UNIDO. This document has been reproduced without formal editing.

id.77-287

GENERAL INFORMATION - AN INDUSTRY PROFILE

CONTENTS

1.	SUMMARY	3
2.	CHEMISTRY & INDUSTRIAL ACTIVITY	6
3.	FORMULATION & MARKETING	8
4.	DESCRIPTION OF PROCESS	9
5.	RAW MATERIAL REQUIREMENTS	12
6.	UTILITIES	13
7.	CAPITAL COSTS	14
8.	COST OF PRODUCTION	19
9.	LIST OF EQUIPMENT	20
10.	SPECIFICATIONS OF THE END PRODUCT	21
11.	RE-BOARDS & REFERENCE	22
12.	SAFETY & ENVIRONMENTAL CONSIDERATIONS	24
13.	MAN POWER REQUIREMENTS	25
14.	BASIC MANUFACTURERS	26
15.	FLOW DIAGRAM	

S U M M A R Y

Methyl Parathion belonging to the organo-phosphorous group of pesticides is one of the widely used insecticides due to its rapid knock-down effect and relatively lower application costs. However, the use of Parathions is banned or restricted in some countries. In some cases, use of Ethyl Parathion has been banned but use of Methyl Parathion has been allowed. Again, in some cases, due to higher toxicity hazards at the time of application, (even in the case of Methyl Parathion) while use of dusts is allowed, the use of emulsifiable concentrates has been restricted to Government supervised programmes as in India. Notwithstanding its toxicity, it is one of the most popular insecticides in most parts of the world.

Parathion requirements for Philippines, Thailand and Indonesia in 1975 were around 474 H.T. as reported by Messrs. C. Pope and J.J. Magee (UNEP/WHO 1973 - UNIDO/FAO Feasibility Survey of pesticide production and use in certain countries). Its use in India is around 1700 T.P.A. and installed capacity of 1700 T.P.A. Manufacturing facilities do not exist in other developing countries of the area.

Major basic raw materials for manufacture of Methyl Parathion are Phosphorous Pentasulphide, Methanol, Toluene, Thionyl Chloride, Phenol, Nitric acid and Sulphuric acid.

The manufacturing process involves three steps :

1. Preparation of Dimethyl Thiophosphoryl Chloride;
2. Preparation of Sodium - p - Nitrophenate; and
3. Condensation of Dimethyl Thiophosphoryl Chloride and Sodium - p - Nitrophenate to give Methyl Parathion.

The route for production of Methyl Parathion involves utilisation of dimethyl thiophosphoric acid which is a starting material for a number of organophosphorous compounds and is a most desirable route especially should it be decided to set up an organo-phosphorus pesticides complex to manufacture other products also.

Capital cost for a 20 T.P.A. plant (which is of an economic size) under Indian conditions works out to be Rs. 17.43 million. A detailed list of equipment, construction costs, off-site facilities etc. is provided in the report. Similarly, details of man-power requirements, maintenance and depreciation costs, utility costs etc. are provided. Also, the information on working capital is provided. The cost of production works out to be around Rs. 25,500/Kg.

The major by-products, which are saleable, are Hydrochloric acid (30%), sodium hydrogen sulphide, O - Nitrophenol and sodium chloride - the last is not economical to recover.

Some of the effluents even can be recovered as saleable products; others neutralized or incinerated. The disposal of effluents should not pose a major problem.

The safety and environmental considerations in case of Parathion manufacture and use is of great importance. Incidents of explosion in manufacture and over-exposure in handling, storage and application leading to fatalities have been reported and as such extraordinary precautions are needed from the manufacturing stage to end-use of the product.

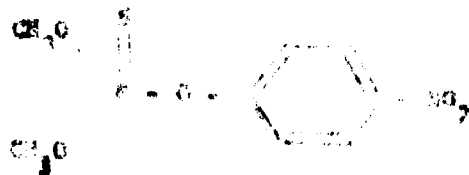
Specifications for the end-product as laid down by U.N.O. and Indian Standards Institution (ISI) are provided in the report.

Patents for manufacture of Parathion have expired all-over the world. There are six known manufacturers of Parathion - the two largest being BASF AG and MONSIEUR.

CHEMISTRY & INSECTICIDAL ACTIVITY

Methyl parathion (2,2-dimethyl-3-(4-nitrophenyl) phosphorothioate) is an insecticide and an acaricide with broad spectrum of activity but with high mammalian toxicity (LD 50; rats oral 14-21 mg/kg body weight; dermal 67 mg/kg body weight).

Parathion is a high boiling deep brown to yellow liquid which possesses a characteristic odour. Technical grade material has a specific gravity of 1.23 and is soluble in most organic solvents but is slightly soluble in aliphatic hydrocarbons. Its solubility in water is very low (55-60 mg/litre). Parathion is stable to hydrolysis in distilled water, hard water and acid solutions. It is, however, rapidly hydrolysed in the presence of alkaline materials such as lime, sodium hydroxide and lime sulfur etc.



Structural formula

After treatment with parathion, some phytotoxicity to green house crops like cucumber and tomatoes has been observed when emulsions are used. But, if parathion is used as a wettable powder or diluted dusts, phytotoxicity is rarely observed. Other vegetable crops and ornamentals can tolerate quite large doses without any harmful effect.

It has been used with success for the control of green
 leaf hoppers, white flies, grasshoppers, cut worms, rice ear
 worms, leaf rollers, rice bugs, leafy cutworms in cereals,
 spotted bill worms, walking beetles, leaf miners, thrips, aphids
 and mites in cotton, leaf beetles, weevils, beetles, thrips,
 leaf miners, cutworms, grasshoppers, cabbage worms, fruit
 worms, moths, beetles, and worms. It is used in vegetables;
 and against flies, house caterpillars, aphids and leaf
 miners in house crops.

It is also used to control the most troublesome
 insects of the house, such as the cockroach (15-25) and dusting
 powder (15-25).

1

FORMULATION REQUIREMENTS

Methyl Parathion is formulated as emulsifiable concentrates (25-50%), wettable powders (15-25%) and dusting powders (1-2%). Formulation facilities for E.C.'s and dusting powders exist in many of the countries. However, as mentioned earlier, the dusts should, perhaps, find favour for manufacture due to their lower application hazards.

Emulsifiable concentrates of Methyl Parathion have been used extensively either alone or in combination with other products especially DDT, M. Parathion and DDT combination dusts are also quite popular in many countries especially for cotton pest control. The demand for this product in the developing countries of Asia including India is estimated to be no more than 2,000 TPA. India has an installed capacity for 1700 TPA and may not be fully utilized due to restricted use under the Insecticides Act. No known restrictions exist in other countries but such possibilities cannot be ruled out in the future. This matter would have to be discussed with country governments and officials concerned prior to establishment of manufacturing facilities.

DESCRIPTION OF PROCESSES FOR METHYL PARATHION

The process of manufacture involves three steps:

- (a) Preparation of Dimethyl thiophosphoryl chloride.
- (b) Preparation of Sodium-p-nitrophenate.
- (c) Condensation of dimethyl thiophosphoryl chloride and Sodium-p-nitrophenate to give Methyl Parathion.

(a) Preparation of Dimethyl thiophosphoryl chloride.

In the present method of commercial production of Dimethyl thiophosphoryl chloride, phosphorus pentasulphide and methanol are reacted in a solvent like toluene and then chlorinated with either chlorine or thionyl chloride.

There are other methods also starting from phosphorus trichloride and dialkyl dithiophosphorus dichloride which, however, have not found much use. The method described earlier has the advantage of utilizing dimethyl thiophosphoric acid which is a starting material for a number of organo-phosphorus pesticides and thus is most suited for organo-phosphorus pesticide complex.

(b) Preparation of Sodium-p-Nitrophenate.

Phenol is nitrated in presence of nitric acid and sulphuric acid at a regulated temperature to yield a mixture of p and o-nitrophenol which are separated by either fractional distillation or steam distillation and converted into sodium salt

by sodium hydroxide. The yield of *o*-nitrophenol has been achieved to the extent of 50-70%.

(c) Preparation of Methyl Parathion.

There are number of methods which have been used for condensation of dimethyl thiocarbonyl chloride and sodium-*p*-nitrophenate. This exothermic reaction can be carried out at 90°C to 100°C, when the time required is 3 hours and also at -10°C to +10°C when the time required is 20 hours. The reaction is conducted in liquid phase and the solvent like alcohol, benzene, chlorobenzene, methyl ethyl ketone etc. has been used. Some of the catalysts like sodium, potassium, barium, triethylamine, trimethylamine, *o*-thymol, *o*-cresol, *o*-nitrophenol, *o*-nitroaniline etc. are known to accelerate the condensation step.

A. The reaction is carried out in a stirred jacketed closed vessel system to give a yield of about 60%. The reaction product mixture may be pumped through a pleated filter to remove gummy impurities and filtrate is separated into aqueous acid oily layer. The oily layer may then be washed with a dilute sodium carbonate solution and the unreacted dimethyl thiophosphate is removed by steam distillation. After cooling and settling, the organic layer can be dried by heating under vacuum to give the product.

B. A product containing 95-98% Methyl Parathion is obtained when the reaction of Dimethyl thiocarbonyl chloride with *o*-Nitrophenol is conducted in presence of anhydrous potassium

carbonate in acetone with the use of sufficiently pure starting material. This product for convenience of handling and transportation is diluted with 10-15% xylene.

C. The most important industrial method is the reaction of Dimethyl chlorothiophosphate with p-Nitrophenol in presence of Hydrogen chloride acceptors or with Sodium-p-Nitrophenolate. The reaction is carried out in water in presence of emulsifiers (e.g. ammonium or amine naphthalenes). By the aqueous method, the compound is obtained with 85-90% of the principal product.

Choice Of The Method.

Out of the three condensation methods described above A and C are in common use. Process A, because of the problem of evaporation of organic solvents at the lowest possible temperature (through film evaporation techniques) required to avoid any possible explosions may not be suitable for less industrialised developing country and, therefore, process C becomes a method of choice.

RAW MATERIAL REQUIREMENT PER LB. OF METHYL PARATHION

P ₂ S ₅	0.48
Methanol	0.30
Form	0.20
Phenol	0.87
Nitric Acid	0.36
Sulphuric Acid	0.72
Chlorine	0.30
Toluene	0.15
Emulsifier	0.06

REQUIREMENTS

Utility	Specifications	Quantity	Application	Equipment required
Water	Softness to 40-50 ppm of hardness limit	160,000 gpd	For process requirement, scrubbing of effluent gases, by-product recovery, make up water in cooling towers, boiler feed etc.	1) Cooling tower ii) Cooling water circulating pumps iii) Cooling tower sump
Steam	10 psig and 50 galg	5 tonnes/day	Required in the process for p-nitrophenol making, methoxy parathion and solvent recovery units.	i) Boiler 1.5 tonnes/hr ii) Oil can with storage tank iii) Hydraulic pressure test pump
Power	-	52556 units per day	For main plant, pumps lighting etc.	400 KW
Refrigeration	-	50 tonnes	Required at most steps of manufacture	2 Drive units of 25 tonnes/capacity.

SCHEDULE

(Figures in thousands)

1.	land and land development (20 acre of land)	2000.00
2.	Equipment cost including effluent treatment plant	1980.00
3.	Plant building, civil and structure	800.00
4.	Painting and installation	500.00
5.	erection	200.00
6.	Know-how and detailed engineering	2000.00
7.	Commissioning	3000.00

11000.00

CAPITAL INVESTMENT

A) Plant & Machinery

1) Methyl-Dichlorophosphate unit

(Rupees in thousands)

i) M.S. glasslined vessel with reflux condenser (200 litres) (1 No.)	300.00
ii) M.S. batch tank for toluene 200 litres (1No)	0.60
/batch iii) MS/tank for methanol 600 ltr. (1 No)	1.00

2) Dimethyl sulphathiosulphate unit

i) MS glasslined vessel 500 ltr. with reflux condenser (2 Nos.)	300.00
ii) storage tank lead-lined 7000 litres cap. (1 No)	19.00

3) Condensation unit

i) MS glasslined vessel 1000 ltr. with reflux condenser (1 No.)	300.00
--	--------

4) Methyl-nitrosulphate unit

i) SS reactor 1000 ltr. (1 No)	75.00
ii) Leadlined wash tank 1000 ltr. (1 No)	6.00
iii) MS batch tank for H_2SO_4 200 litres (1 No.)	0.40
iv) MS-AR brick lined batch tank 500 ltr. (1No.)	8.00
v) SS batch tank for Na_2CO_3 200 ltr. (1 No)	4.00
vi) M.S. distillation unit (1No)	20.00
vii) SS storage tank β -nitro-phenol (1 No)	20.00

5) Miscellaneous

i) For effluent treatment unit	300.00
ii) Misc. equipment e.g. storage tank, pumps etc.	100.00
iii) Pipe line fittings etc.	20.00
iv) Misc. (contingencies)	10.00

B) Land and building offsite facilities and misc.

15000.00

17000.00

ESTIMATE OF SUPPLY FACILITIES AND MISC. EXPENDITURES

(Dollars in thousands)

1.	Water	
	Overhead tank 2000' construction & water supply	40.00
2.	Power receiving and distribution system including emergency power	2000.00
3.	Water softening plant capacity 250 ³ /hr	100.00
4.	Air for instrumentation 500 ³ /hr	75.00
5.	Refrigeration capacity 50 tonnes	400.00
6.	Cooling tower	400.00
7.	Yard piping	350.00
8.	Compound wall and fencing	150.00
9.	Roads as per layout	50.00
10.	Time office, change room and welfare	85.00
11.	Canteen	100.00
12.	Laboratory	255.00
13.	Dispensary	30.00
14.	General store	65.00
15.	Administration block	450.00
16.	Transport Vehicle	50.00
17.	Security and guard room	30.00
18.	Material handling equipment	200.00
19.	<u>Storage</u>	
	Underground	
	i) Methanol	300.00
	ii) Toluene	
20.	Genetic seeds and sulphuric acid	200.00
21.	Raw material and drum storage	50.00
22.	Fuel oil storage	100.00
23.	Methyl parathion	50.00
24.	Steam generator plant 1.5 tonnes/hr	350.00
25.	<u>Misc.</u>	
	i) Preliminary	200.00
	ii) Pre-operative expenses	1600.00
	iii) Contingencies	500.00
	iv) Interest during construction	1200.00
	Total:	<u>2410.00</u>

STATEMENT OF CAPITAL OUTLAY

	(Dollars in thousands)
1. Land and land development	2000.00
2. Main plant and equipment including erection and commissioning	8780.00
3. Services, offsite facilities and Misc. expenditure	11700.00
4. Working capital	8075.00
5. Misc.	215.99
	<u>18768.99</u>

e) **MINERAL CAPITAL**

(based on three months requirements)

(Report in thousands)

H_2S_3	278.00
Methanol	178.00
Phenol	670.00
HNO_3	180.00
H_2SO_4	94.00
Cl_2	85.00
Toluene	75.00
Emulsifier	100.00
NaOH	61.00
	<hr/>
	2075.00
	<hr/>

UNIT OF PRODUCTION

1. Raw Material Cost:-		(Rupees in thousands)
F_2S_3	21x11000 @ Rs.1100/- MT	2310.00
Methanol	150x4000 @ 4000/- MT	600.00
HNO_3	180x4000 @ Rs.4000/- MT	720.00
Phenol	15x28000 @ Rs.8000/- MT	1400.00
H_2SO_4	30x500 @ Rs.500/- MT	15.00
Cl_2	150x700 @ Rs.700/- MT	105.00
Toluene	7000 @ Rs.30 @ 45000/- KL	300.00
Emulsifier	20x20000 @ Rs.20000/- MT	400.00
NaOH	100x1640 @ Rs.1640/- MT	164.00
		<u>2895.00</u>

2. Utilities:-		
Water	100,000 gallons/ton @ Rs.1/- KL	100.00
Power	3456x1000kwh @ Rs.25/- per kWh	850.00
Steam	3000 MT/tonne	30.00
		<u>980.00</u>

3. Salaries and Wages:-		
Labour (direct)		720.00
Labour (indirect) and Supervisory staff		
Overhead 50% of direct labour		
4. Maintenance @ 5% on plant & Machinery		90.00
5. Depreciation @ 15% on plant & Machinery		270.00
6. Interest on total investment (involving capital and capital cost @ 12%)		370.00
		<u>1450.00</u>

1.	2895.00
2.	980.00
3 to 6	1450.00
	<u>5325.00</u>

Cost of production per ton: **87.50**

LIST OF EQUIPMENTS FOR MEXCEL PARASOLIC PLANT

	Quantity	Material of construction	Capacity
<u>REA UNIT</u>			
1. Reaction kettle	1 No	GL	1000 lts.
2. Reflux condenser	1 "	GL	" "
3. Toluene batch tank	1 "	MS	600 "
4. Wash batch tank	1 "	MS	300 "
<u>SALICYLHYDROXYBENZOIC ACID UNIT</u>			
5. Chlorinator	2 "	GL	500 "
6. Reflux condenser	2 "	GL	" "
7. Storage tank	1 "	SL	7000 "
<u>MEXCEL PARASOLIC UNIT</u>			
8. Reaction kettle	1 "	GL	1000 "
9. Reflux condenser	1 "	GL	" "
10. Toluene recovery vessel	1 "	MS	1000 "
11. Batch tank	1 "	MS	300 "
<u>MISCELLANEOUS EQUIPMENTS</u>			
12. Pumps for pumping and transfer of material from one vessel to another	4 Nos	GI	" "
13. Storage tank for H_2SO_4 , NaOH, toluene, nitric acid, emulsifier phenol, hydrochloric acid, Sod. MS etc.	10 "		7 days storage capacity
<u>MSD. p-NITROPHENOLIC UNIT</u>			
14. Reaction kettle	1 "	GL	1000 lts.
15. Wash tank	1 "	SL	1000 "
16. Batch tank	1 "	MS	300 "
17. Batch tank	1 "	MS	300 "

SPECIFICATIONS OF THE END PRODUCT

a) **Indian Standard Specifications (IS-2572-1957)**

Acidity (as H_2SO_4)	0.15 (max.)
Active content	95 (min.)
Water	0.15 (max.)
Material insoluble in acetone	0.15 (max.)
Sp. gravity at 25° C	1.817 (min.)

b) **M.S.S. Specifications (MSD/STP/11.2.1)**

Sr.No.	Requirement	Limits	
		Max.	Min.
1.	O,O-Diethyl O-4 (nitrophenyl) Phosphorothioate content % by weight	90.0	-
2.	Acidity % by weight, calculated as H_2SO_4	-	0.5
3.	Solid matter insoluble in acetone % by weight	-	0.5
4.	Water content, % by weight	-	0.5
5.	Sp. gravity at 25° C/15° C	1.8577	-

BY-PRODUCTS and EFFLUENTS.

<u>1. By-Products.</u>	<u>Quantities</u>
i) Hydrochloric Acid - 30%	105 Tons .
ii) Sodium hydrogen sulphide.	65 Tons .
iii) O-Nitrophenol.	240 Tons .
iv) Sodium chloride.	Not economical to recover.

2. Effluents.

(a) Gasous Effluent.

Gasous effluent consist of the following:

(i) Hydrogen Sulphide.

This is either recovered by absorbing in sodium hydroxide solution to produce sodium hydrogen sulphide (used extensively in leather tanning industry) or disposed of by burning.

(ii) Chlorine.

Trace of residual chlorine is removed by scrubbing with caustic solution.

(iii) Hydrochloric acid.

Exhaust absorption system is required with re-circulation system to produce saleable hydrochloric acid.

(d) WATER EFFLUENT.

The liquid effluent normally contains dissolved solids mainly NaCl, NaOH and traces of organophosphorous compounds, oils and solvents. Treatment of waste water normally is done by inbuilt facilities in process like hydrolysis with hot alkali, extraction

with solvents etc. followed by incinerators which consist of absorption towers containing regenerable granular active carbon etc. The final water is then collected in a common separator for final separation of solid, oil and solvent. This after bio-oxidation is then discharged out of factory.

Normally down stream water is not recommended for any other use.

Waste Product.

- (1) Oils and solids are about 0.25 tons per day and are disposed off by incineration.
- (2) Liquid about 2000 gallons per day.

SAFETY AND ENVIRONMENTAL CONSIDERATIONS

- i) The condensation reaction has to be carried out in a closed vessel as the bye-product gases are quite toxic and also a nuisance.
- ii) During recovery of solvents at the purification stage incidence of explosions have been reported.
- iii) Methyl parathion is highly toxic to warm blooded animal both through inhalation and skin absorption. It is, therefore of utmost importance that due care is taken for workers safety during manufacturing state and also at the time of application.
- iv) The process evolves such toxic gaseous effluents like H_2S , nitrous oxide fumes etc. and liquid effluent containing traces of methyl parathion thiophosphoric acid, nitrophenols etc. Due care is required to control exit of these effluents into environment.
- v) Due to highly toxic nature of this pesticide, stricter control measures must be adopted in regard to packaging, storage, distribution, application etc.

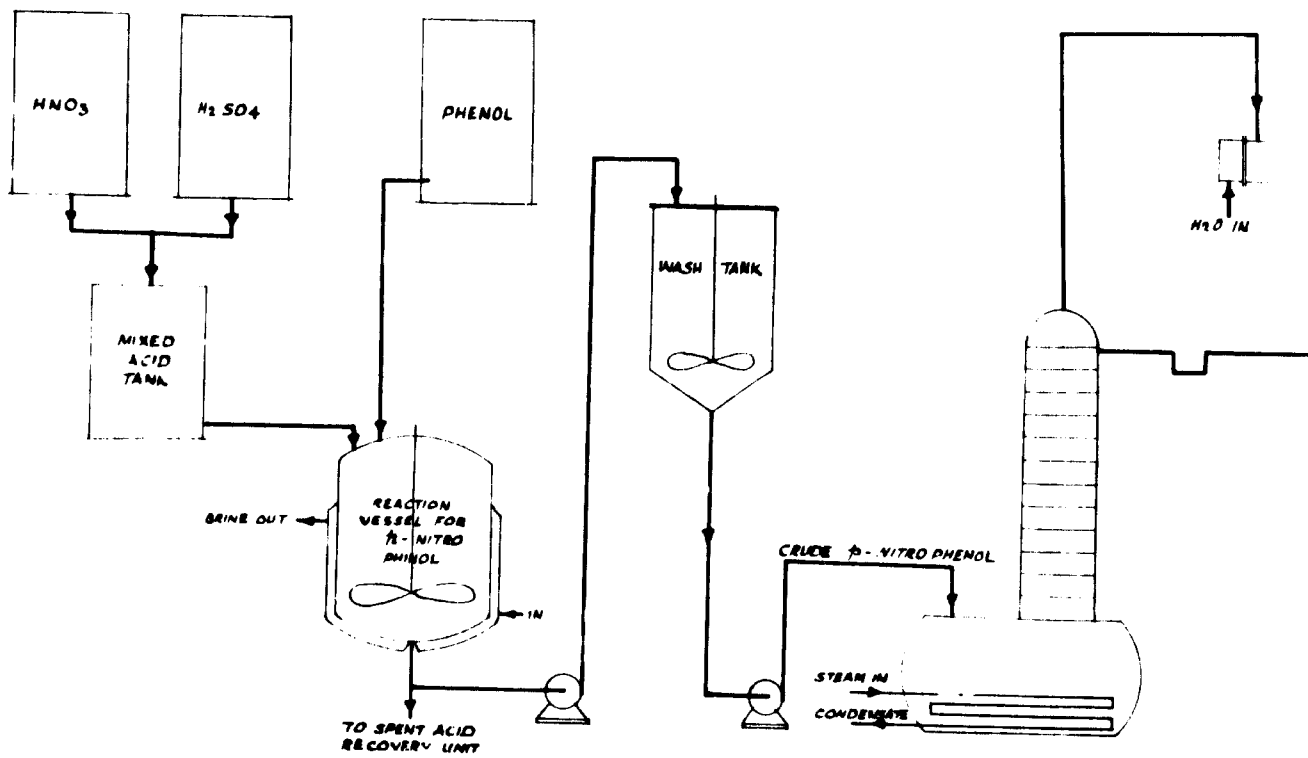
MAN POWER REQUIREMENT

<u>Sr.No.</u>	<u>Personnel</u>	<u>No.</u>
1.	General Manager	1
2.	Production Manager	1
3.	Engineering Manager	1
4.	Finance Manager	1
5.	Commercial Manager	1
6.	Personnel Manager	1
7.	Security Officer	1
8.	Stores Officer	1
9.	Accountants	2
10.	PA to GM	1
11.	Receptionist	1
12.	Typist clerks	4
13.	Driver	1
14.	Peon	1
15.	Watchman	6
16.	Shift Supervisor	4
17.	Maintenance Supervisor	3
18.	Plant Operators	8
19.	Skilled workers	4
20.	Unskilled workers	14
21.	Chemists	2
22.	Boiler operator	4
23.	Refrigeration operator	4
24.	Fitter/welder	4
25.	Fitter helper	4
26.	Electrician	2
27.	Electrician helper	2
28.	Instrument mechanic	1

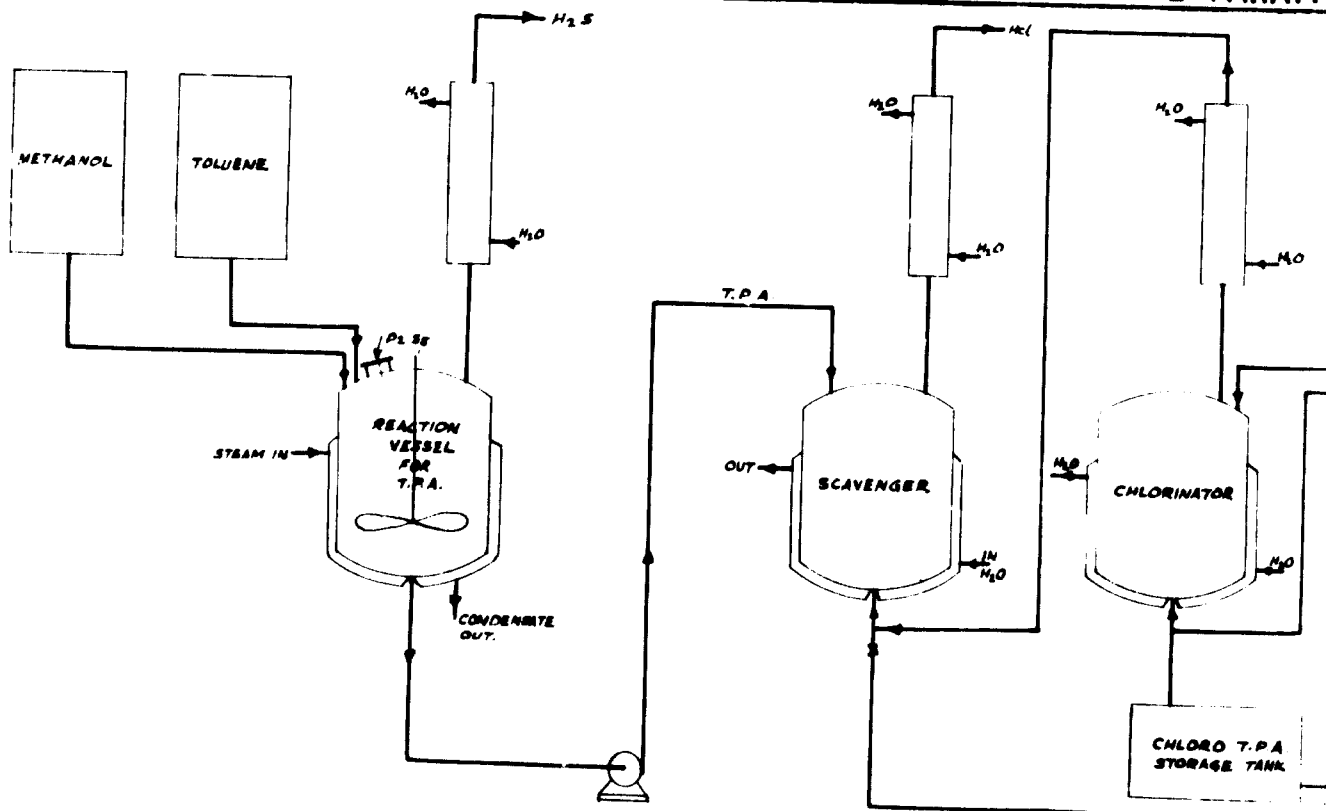
BASIC MANUFACTURERS OF METHYL PARATHION & THEIR ADDRESSES

1. BAYER AG
SPATE PFLANZENSCHUTZ ANWENDUNGSTECHNIK
309, LEVERKUSEN BAYERWERK
FEDERAL REPUBLIC OF GERMANY
2. CHEVINOVA
P.O. BOX 9 BK 7620 LEHUS, DENMARK
TELEPHONE: (07) 88-41-00
3. KERR-McGEE CHEMICAL CORPORATION
KERR-McGEE CENTRE
OKLAHOMA CITY, OKLAHOMA, U.S.A. - 73129
TELEPHONE: (405) 236-1313
4. MONSANTO AGRICULTURAL PRODUCTS COMPANY
800 N LINDBERGH
ST. LOUIS, MISSOURI, U.S.A. - 63166
TELEPHONE: (314) 694-1000
5. STAUFFER CHEMICAL COMPANY
AGRICULTURAL CHEMICALS DIVISION
WEST PORT, CONNECTICUT, U.S.A. - 06880
TELEPHONE: (203) 226-1511
6. BAYER INDEA LIMITED
EXPRESS TOWERS
WARDIAN POINT, BOMBAY, INDIA - 400 001

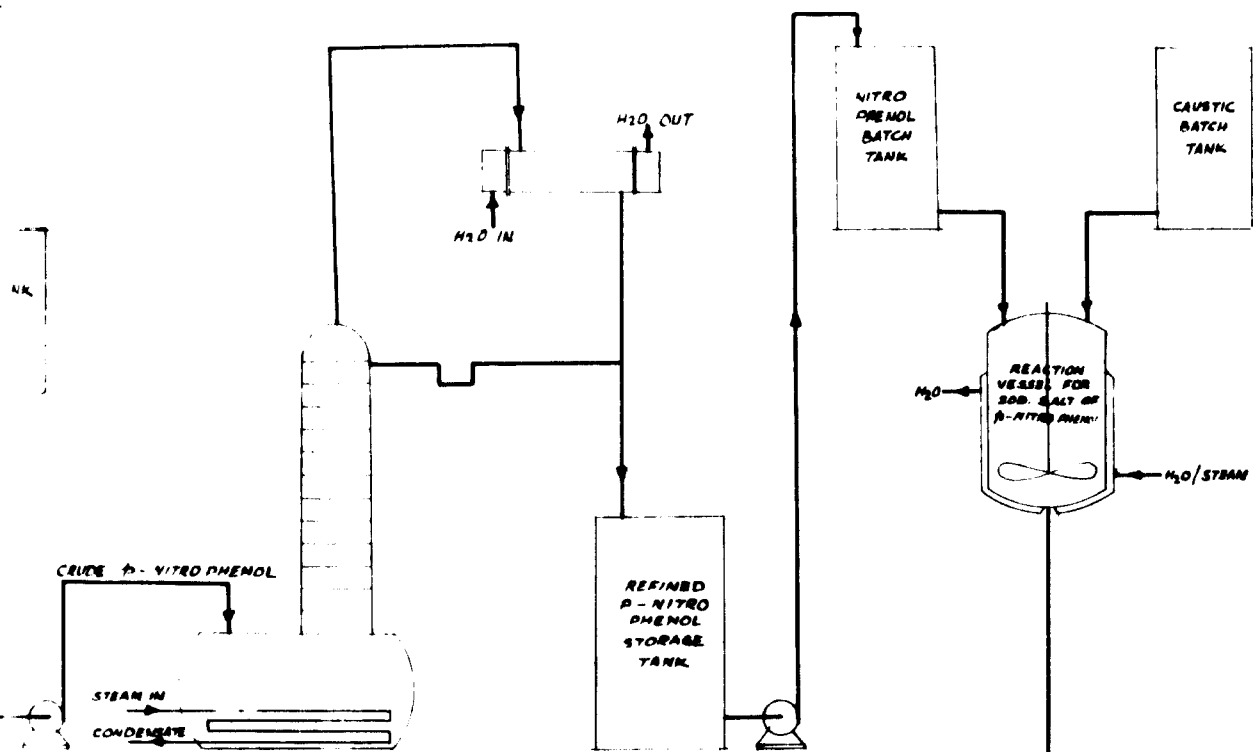
MANUFACTURE OF P-NITRO PHENOL



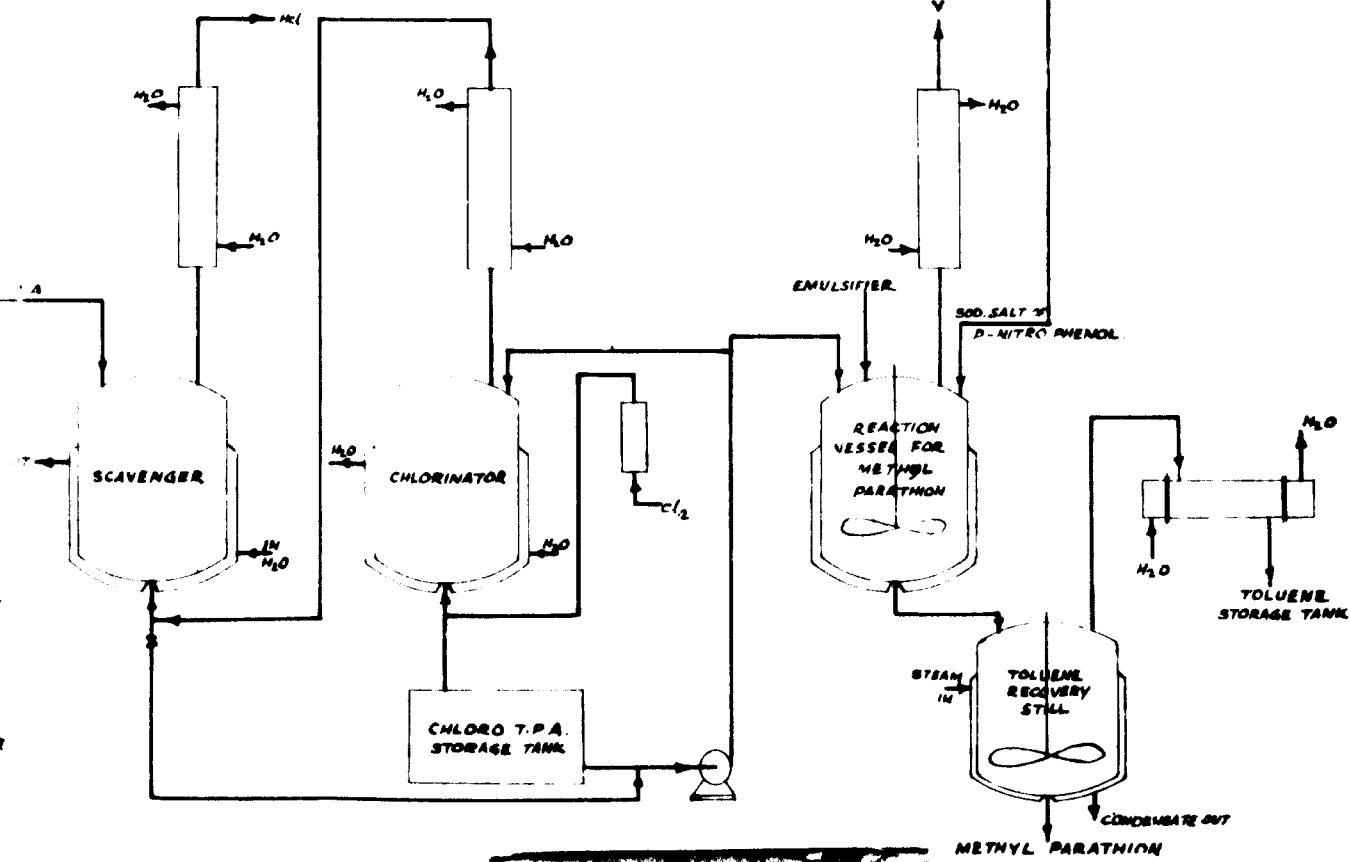
MANUFACTURE OF METHYL PARATHI



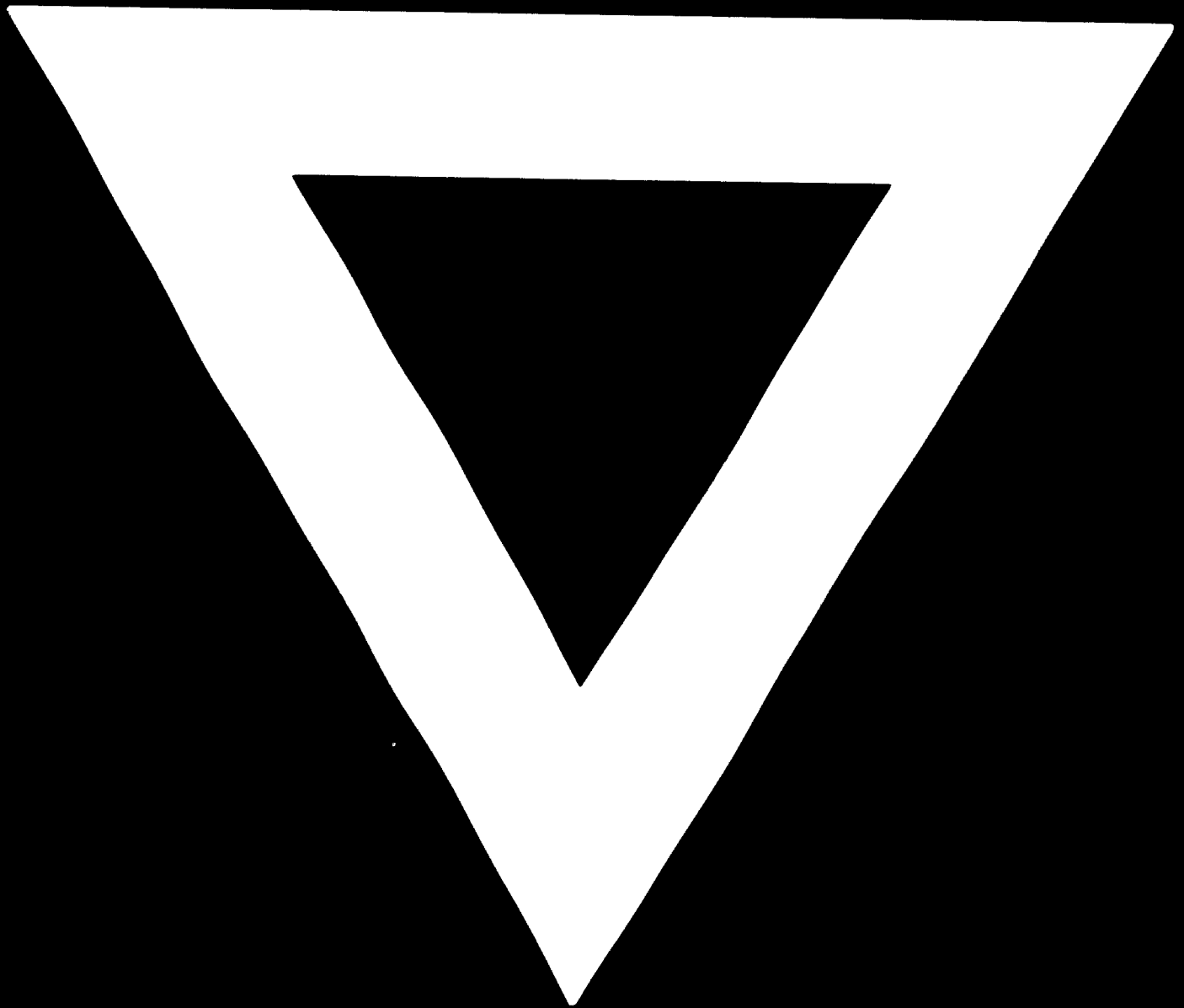
MANUFACTURE OF P-NITRO PHENOL SODIUM SALT



MANUFACTURE OF METHYL PARATHION



C-671



78. 11. 09