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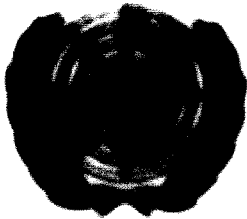
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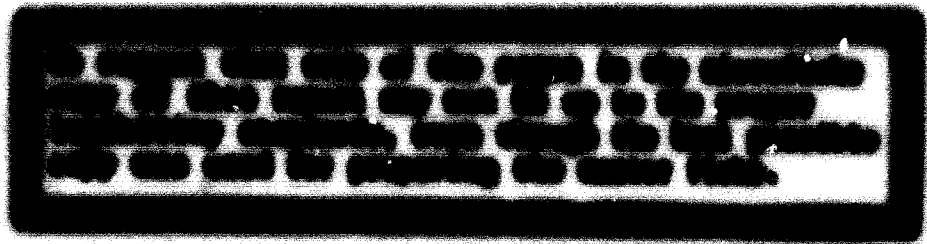
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**MEMORANDUM FOR THE BOARD OF DIRECTORS
RE: THE AVAILABILITY OF UNITED STATES
PROPERTY**

1.

Mr. Board and Mr. Director
The Board of Directors
United States

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SUMMARY

The paper begins with a survey of the present and future supply position for making coke in relation to overall resources of fossil fuels and other energy sources. This shows that optimum technology for producing will change considerably during the next half-century.

In relation to the near future, consideration is given to techniques of improving coke technology with the objects of maintaining quality standards whilst extending the range of usable coals. Classical prime making coals can be replaced or supplemented by inferior grades with the application of sophisticated techniques of blending, and coke quality and performance can be improved by charge preheating. More basic solutions involve the manufacture of artificial or formed coals.

In the immediate future, the injection of gaseous and liquid hydrocarbons into the blast-furnace tuyeres can be economically advantageous, and other possibilities are opened up by the injection of reducing gases in the top of shaft.

Proposals for direct-reduction systems based on fossil fuels, eventually supplemented by the use of nuclear heat, are examined. Finally, consideration is given to the problem of iron-oxide reduction in the absence of all fossil fuels.

World resources of coking coals are more than enough to smelt all the classical iron-ore reserves.

At first sight this statement would appear to remove any vital necessity to change ironmaking technology to a radical degree but, of course, commercial and local considerations must have an important effect on development. For this reason it seems desirable to consider the broad pattern of availability of energy sources during the next half-century.

ENERGY SOURCES

Primary sources relevant to the manufacture of iron include coking grades of hard coal, other hard coals, fuel oil, natural gas, and nuclear energy, this last including direct use of the heat of nuclear reaction. All the fossil fuels have been used for the reduction of iron ore but the situation has hitherto been completely dominated by the blast furnace using coke as prime fuel.

Fossil Fuel Reserves

Many estimates have been given of reserves and their geographical distribution, and large discrepancies are to be found.⁽¹⁻⁴⁾ Despite this, the overall picture is reasonably clear and useful conclusions can be drawn.

Hard coal reserves may be taken as about 5×10^{12} tons, of which half is in the U.S.S.R. and China and one-third in the U.S.A. Various projected rates of consumption tend to give a lifetime in excess of 1000 years. It is, of course, more difficult to obtain reliable figures for coking coal, if only on account of the problem of accurately defining coking grades; differences in this respect could introduce a factor of 2. It is probably sufficient to follow Bellanc⁽⁴⁾ and to consider a world reserve figure of 10^{12} tons. Three-quarters of this tonnage lies in U.S.A., U.S.S.R. and China and large areas of the world will have to depend on imports.

Estimates of oil reserves have varied widely and been subject to repeated reappraisal. In recent years, however, a better appreciation of the characteristics of the earth's crust has led to estimates which are more likely

to remain stable... In recent years, despite improved techniques... the estimate of known reserves has been... changes are complicated by inadequate detailed... in potential extraction rates but a working... real equivalent may be taken as a basis for... of this being located in developing countries... Authorities differ but most are agreed that... relative prices and consequent consumption... the century.

The position for... is largely similar, with substantially lower reserves and... use as a convenience fuel. Serious world wide... by 2000 A.D., certain areas having major problems...

The impact of the... of various fossil fuels on ironmaking technology will be... As a particular resource approaches exhaustion, its price will... this may lead to the abandonment of its use or its replacement by... substitute. In the case of coking coal there are... to meet all possible requirements... for an indefinite period: estimates of iron-ore reserves vary... the highest suggested being just over 0.5×10^{12} tons. ... of coal in any particular location will be determined by... of the following factors:-

- (1) social-economic... of mining the coal.
- (2) economics of... of liquid or gaseous hydrocarbons - this may be... with time. Currently replacement is favour... a state of affairs not likely to... the present century.
- (3) new technologies... sources other than fossil fuels may become... favourable at any time after the next few years.

SOLID FUEL FOR THE BLAST FURNACE

Despite the extensive resources of first-class coking coals distributed around the world, many countries have a real economic interest in manufacturing good blast-furnace fuel from relatively inferior raw material. The situation arises where coals abound of grades other than those ideal for coking. A great deal of research and development has gone into the problem and it is possible to maintain economic operation in the face of a continuously degenerating supply position.

Blending

Experience in the United Kingdom has been concerned largely with the manufacture of classical coke in normal-type coke ovens. Some coking coals of 15-25% volatile matter (dry basis) have become increasingly scarce and also more expensive to win and it has been necessary to resort more and more to the higher volatile ranges with varying degrees of caking power. In terms of the British National Coal Board classification, this corresponds to a move from 301 type to types 500 and 600 (say from A3h/53h to A6h/70h on the B.N.C.B. classification⁽⁷⁾). Under these circumstances it is possible to effect improvements in physical properties of coke by incorporating various low-volatile additives in the charge.

Detailed accounts of work in this field have been given by the British Coke Research Association and the National Coal Board.^(8,9)

This work, together with a great deal of subsequent development at coking-oven plants, showed that substantial improvements can be made in the strength of coke made from the more readily available medium caking coals. Bauxite, a cheap and readily available additive, will increase the $\text{M}40$ or stability index and raise the mean size of the product, but it has an adverse effect on the abrasion ($\text{M}10$) index and in consequence may need to be supplemented by other materials. Low-volatile coals of the semi-anthracite type are found generally to be beneficial additives, the only disadvantage being usually that of cost due to regional shortage or to high transport charges in the U.K. Somewhat similar improvements can be brought about by the use of char manufactured by partially devolatilising high-volatile coals.

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Conclusion

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... get the ... The ... strength ...

Reports have been published of furnace trials using different ... the total ... of the furnace walls.

P.M.C. gas was used in the U.S. Steel Corporation experimental blast furnace (14) and B.F.F. briquettes in the blast furnace (15).

Various trials involving different sizes of fuelled gas in a 750 m³ furnace for a week were successful but some slight loss in efficiency was reported. ... complete success in trials on a 1300 m³ furnace over the ... with a final result of a week at over 40% substitution (16). The trials using B.F.F. briquettes were of particular interest. A furnace of 70 m³ capacity was operated exclusively on carbonised briquettes for a week but in addition a trial of 7 days duration was carried out using briquettes which had not been subjected to the final carbonising stage (17). The results suggested that this last operation could be omitted, a fact which if substantiated would have a markedly favourable influence on the economics of the process.

More recently the British Steel Corporation has run full-scale trials on B.F.F. and P.M.C. fuels profited by a laboratory study of these together with the Japanese B.L.S. product.

In addition to chemical analysis and physical testing, the preliminary studies involved observation of behaviour in a tube furnace carried out in a special combustion rig. This apparatus (18) allows fuel to be burnt in front of a tube using high-velocity hot blast. A furnace is formed and this region can be explored by a series of analysis probes to show the progress of the reduction reactions.

The first trial was conducted by the authors as a function of L/E length diameter, for a limiting value. The results obtained using this test of the characteristics of the material.

The test parameters include number of cycles, L/E length diameter, for a limiting value. The results obtained using this test of the characteristics of the material.

The first trial was made of large L/E steel specimens. The stress is based on a section of maximum width, and is referred to a L/E and a relatively great cutting rate. In the L/E test, it was found that the stress state of the specimen is not uniform for increasing the frequency. The displacement of the L/E test had a tendency of potentially reliable results and permitted to achieve satisfactory test results in terms of the test specimens for the respective test. From the properties listed in table 1.

The first trial series of 'gauge' specimens was supported in the water and passed through a handling system to eliminate the most significant source of error before the results were compared on a number of test test specimen.

The first series trial was conducted by a test of decrease and find various divisions of L/E . The objective was to test the results already existed in the L/E test and the results and duration period of the materials. Integrated early test specimens were obtained and all weights into and out of the system directly applied decrease, further

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Section 1. The purpose of this document is to provide a comprehensive overview of the project's objectives and scope. It is intended for the use of all stakeholders involved in the project.

The project is designed to address the current challenges faced by the organization and to provide a long-term solution. The primary goal is to improve operational efficiency and reduce costs. This will be achieved through the implementation of a new system that integrates all business processes.

The project will be managed in accordance with the principles of project management, ensuring that all tasks are completed on time and within budget.

The project team consists of members from various departments, including IT, Finance, and Operations. Each team member has specific responsibilities and is committed to the success of the project. Regular communication and collaboration are essential for the project's progress.

The project is currently in the planning phase, and the next steps will be to develop a detailed project plan and to begin implementation.

The project is expected to be completed by the end of the year. The results of the project will be evaluated against the initial objectives to ensure that the project has met its goals and provided the expected benefits to the organization.

Injection of Hydrocarbons

Injection at Taper Level

Taper injection of hydrocarbons into the blast furnace gas, to improve the blast, has been practiced for many years, and has become a well established development. Recent developments have been the use of natural gas in the U.S.A. and the U.S.S.R. and the injection of oil in Europe and Japan. It has been estimated (20) that injection practices are used in the production of over 80% of the world's iron. The quantities of injection vary widely and also the replacement ratio, the latter being apparently mainly determined by the blast temperature available or, in other words, the ability to maintain necessary temperatures in spite of the chilling effect of the injections. In addition, difficulties may be experienced in securing complete conversion of the hydrocarbons to reducing gas. Two recent developments show much promise in overcoming this problem. In one system oil is injected into the oil upstream of the lance. This has proved successful at the Fort Hindustan and other plants of the British Steel Corporation. An alternative approach which is used successfully in Germany (21) is the use of a water-in-oil emulsion with 5-10% water. Both systems would appear to promote the formation of fine oil droplets by the explosive evaporation of contained water drops.

Injection in the Ash/Slag

An interesting alternative to the injection of hydrocarbons with the blast is the technique of supplementing the reducing gases of the blast-furnace stack by injecting essentially carbon monoxide/hydrogen at a level above which solution loss is not likely to occur. If the gas is preheated to the appropriate temperature, say 900 - 1000°C, and can be adequately distributed and mixed with the normal stack gases, considerable improvement in gaseous reduction can be achieved with consequent savings in coke. Calculations suggest potential economic benefits, the advantages being largely additional to those of taper level injection. Very low requirements of solid fuel are implied, perhaps of the order of 150 kg/tonne, or less.

Technical problems center on the problem of injecting gas with minimum oxidizing constituents and on the control of temperature and on the effective introduction to and distribution in the furnace stack. In partial oxidation systems such formation can be serious. If this is overcome by steam addition the gas becomes too oxidizing unless the excess water is condensed out - this involves loss of the sensible heat. Operation of the reformer at high temperature (1400°C) is helpful but leads to engineering problems of cooling to the desirable injection temperature. Catalytic reforming is an alternative but this demands extreme physical properties in the heat-exchanger material and also restricts the choice of feedstock to natural gas or light petroleum fractions.

Distribution and mixing of the injected material is evidently important and it seems clear that the method of injection can have little influence on it. Theoretical studies are of a certain limited value but practical trials are necessary; a start has been made in this direction. Inevitably some of the early work was done in small experimental furnaces in which the mixing problem is unlikely to be significant.

In the U.S. Bureau of Mines furnace⁽²²⁾ stack injection was based on a mixture of natural gas reformed by oxygen with dry top gas used as a coolant. Injection was at 900 - 1000°C and the N_2/O ratio $(O_2 + H_2)/ (CO_2 + H_2O)$ was only $5\frac{1}{2}$ to 6 and replacement ratios were in the range 0.5/0.7 kg coke/m³ natural gas.

Nippon Kasei Kisha⁽²³⁾ worked with a furnace of 0.6 m inside diameter, 3.2 m³ volume, and injected the products of partial oxidation of oil at 1200 - 1300°C. The N_2/O ratio was about $8\frac{1}{2}$ and there was 1/3 of coke. Replacement ratios of about 0.8 (coke/oil) were obtained; indirect reduction in the stack, initially 65%, went up to 85%.

AIRED using the experimental furnace at Lidge⁽²⁴⁾ studied the use of partial oxidation natural gas-oxygen with an N_2/O about 5. The fact that the gas was injected at 1400°C encouraged reaction of the oxidizing constituents with the coke. The resultant replacement ratio is given as 0.45 kg coke per 1 m³ natural gas. This work also showed that stack

The results of the tests conducted on the furnace are shown in the following table:

In general, the furnace is very efficient, converting a large portion of the total energy input into a fuel gas of low calorific value. The furnace is capable of operating at a pressure of 100 lb./sq. in. and is capable of handling a wide range of feedstocks. The furnace is capable of operating at a pressure of 100 lb./sq. in. and is capable of handling a wide range of feedstocks. The furnace is capable of operating at a pressure of 100 lb./sq. in. and is capable of handling a wide range of feedstocks.

Other work on a prototype unit has been completed by the company and the results are summarized in the following table. The furnace has a length diameter of 4:1 and the gas flow rate was taken from an existing reforming plant and preheated before injection. The gas was manufactured by steam reforming of natural gas. Heating was made with an excess of steam, which was subsequently removed, to give a steam to gas ratio of 10:1. Horizontal probe traverses suggested that production across the stack was good and a replacement ratio of 1.0 (lb. reformer gas / 0.6 lb. natural gas) was obtained, this being independent of simultaneous injection of 75 cc/lb. of oil through the burner.

The Hot Insulation

The plant furnace is an extremely efficient device, having very small thermal losses, but it does convert a large proportion of the total energy input into a fuel gas of low calorific value. A useful proportion of this heat can be got back into the system via heat preheating; much of the attraction of fuel injection stems from the fact that it increases the amount of heat in heat which the furnace will accept. There are other possibilities of recycling low-grade thermal energy. In particular the oxidizing constituents can be utilized in the reforming of hydrocarbons, the reducing components then providing part of the injection stream. Analogous operations can form part of the reforming and gasification in the Midrex and Fischer process (see previous) so it should eventually be possible

Further studies are possible by the injection of top gas, possibly
in the form of CO₂, in the furnace. The small amount of top gas collected
was immediately sent through a series of tests. Some of these have
analyzed both a mixture of top gas with water and the top gas at low flow
the injection of oxidized top gas (CO₂) at 100% together with top
gas (CO) and its subsequent use.

The entire field of high-carbon processing and reducing gas injection
presents an interesting world market for increasing production as well as for
saving the cost of current prices, resulting economies.

CONCLUSIONS

In recent years many processes designed to replace the blast furnace
have been the subject of special articles, reviews, and international
conferences. For of the hundreds of processes proposed have survived
to reach the stage of industrial operation and of these the majority show
little promise of widespread application.

The consensus of opinion at present is that direct-reduction processes
are applicable in situations where demand is insufficient to support economic
blast-furnace operation, where natural gas is cheap or where coke is
particularly expensive. Otherwise the blast furnace may be expected to
dominate the field.

Of direct-reduction plants installed in recent years, some (for example,
Kilbuck, Ill.) depend on relatively cheap natural gas to produce a sponge iron
product, and their fortunes are likely to be bound up with its future avail-
ability under conditions of world-wide demand for 'conventional fuels'.

Systems using solid fuel, particularly non-coking coal, include the
B/F process and the electric iron-shaft furnace, the latter having the
particular advantage of producing a sinter product, the impurities being
removed in a slag. With the development of cheap electrical energy supplies
and an increasing difficulty in obtaining fossil fuels, this process could
become important in the long run.

NUCLEAR ENERGY APPLICATIONS

With the development of fusion energy plants much thought has been given to the possibilities of application to iron and steel technology. The potential availability of cheaper electrical energy from this source gives up a number of interesting prospects but in general few technical problems are involved. Of much more interest, however, is the question of the use of nuclear heat without the loss of efficiency consequent on its conversion to electricity.

High-temperature reactors exist in the U.K. (Dragon), in the U.S.A. (Fueh lotion), and in the Federal Republic of Germany (Jülich); it seems likely that this type of device can operate with coolant helium outlet temperatures up to 1000°C . Before passing to a power-generation system, the coolant can be used to provide heat to sustain endothermic reforming or reduction reactions.

Extensive studies have been carried out under sponsorship of the Commission of the European Communities^(32,33). In this co-operative work, various groups examined reduction processes based on coal or on reformed reducing gas and analysed the overall energetics of various systems⁽³⁴⁻³⁶⁾. Similar studies have been carried out by the British Steel Corporation in association with the Dragon project team⁽³⁷⁾ and by the Japanese Iron and Steel Institute⁽³⁸⁾.

Most attention has been directed to sponge-iron plants based on a reducing gas, the reactor heat being used to supply the heat of the reduction reactions and sensible heat for the incoming gas and, in some cases, the thermal requirements of a reforming process. In the present state of technology, temperature levels in the various schemes are somewhat critical and much reliance has to be placed on future improvements in performance of materials, particularly heat-exchanger elements. Processes studied in some depth are basically analogous to the existing sponge plants such as H-iron or FIOR in that they use hydrogen or CO/H_2 mixtures which are derived from fossil fuels. This generally demands a source of heat effectively at 900°C or more. It is true that high-pressure hydrogen systems operate at much lower temperatures but the reforming units designed

to convert natural gas to hydrogen have thermal demands in the higher-temperature range. Moreover, the primary helium coolant circuit must operate at elevated pressure - say, 40 kg/cm² - and must be protected against cross-contamination with the reacting gases by secondary heat exchange. This means that materials must be employed capable of withstanding pressure differentials of many atmospheres at a temperature of 1000°C and in presence of CO, H₂, and He. These are formidable requirements but they will no doubt be met. The association of ore reduction reactors with atomic piles also presents problems arising from the need to bring handling facilities for ores and products into close proximity to highly sophisticated systems involving complex safety devices.

Another problem is involved in the necessary scale of operation. A reactor of 2000 M W (Thermal) is probably the smallest that can be contemplated on economic grounds and this would provide heat for iron production appropriate to an ingot capacity of 5,000,000 t.p.a. together with appreciably more electrical energy than the works could use⁽³⁷⁾. Suggestions have been that a large reactor system be used to produce hydrogen which would then be transmitted through pipe-lines to the point of use. This would mean that the reactor could not supply the sensible heat required by the iron reduction process.

Insofar as schemes currently under discussion are based on fossil fuels they could have a beneficial effect on the requirements of the iron and steel industry. In comparison with gaseous reduction processes which rely on natural gas both as chemical feedstock and source of reaction heat, a combined nuclear/fossil scheme would save 50% of the hydrocarbon demand. This in itself is unlikely to influence the date of effective exhaustion of reserves but it could maintain a favourable economic balance for a longer period. Moreover, the conversion of solid fuels to gases, including hydrogen, is likely to benefit from cheap nuclear energy also.

On the whole it seems that the direct effects of nuclear energy on iron and steel manufacture are likely to be long-term ones and, of course, the possibilities mentioned so far imply the continued availability of fossil fuels. For the 21st century in fact they probably require coal which, at least, could be used more directly via formed coke.

It is perhaps appropriate therefore to consider the prospects for ironmaking when solid fuels are no longer available. Such a state of affairs is likely to arise through complete exhaustion of reserves on one of the following grounds: (a) the inability of persuading anyone to mine the coal, (b) a general prohibition of 'dirty' operations associated with coal mining and processing or (c) concern with the rising level of CO₂ in the atmosphere resulting from combustion of fossil fuels.

This essentially leaves hydrogen as reducing agent and water as the only source of hydrogen. A number of authors have studied the characteristics of an energy economy involving hydrogen as an important secondary energy carrier⁽³⁹⁻⁴⁰⁾. Electrolytic hydrogen suffers the disadvantage of efficiency factors in the generation of electricity from nuclear heat and also in the process of electrolysis but there are possibilities of obtaining hydrogen by a sequence of chemical reactions sustained by process heat⁽⁴¹⁾ and these, or other techniques of converting water into hydrogen may in the very long term provide the basis for the manufacture of iron and steel.

APPLICATION PROSPECTS

The future pattern of fuel/energy usage in ironmaking will be determined primarily by the cost and availability of the various alternatives.

At the present time the position appears to be dominated by problems of securing adequate supplies of coking coal at reasonable cost. Cost price structures are perhaps somewhat distorted at present in that demands have risen rapidly in relation to the practicable possibilities of stepping up extraction rates by new investment. A further factor, which also has long-term implications, is the increasing attention which is having to be paid to safety in mining. This apart, running costs are likely to rise and there may be increasing difficulties, reflected in prices, in securing sufficient labour to work in this field.

The position is likely to continue that coal will be available in any quantity required for ironmaking over the next half-century at least but at prices which will encourage the use of alternative energy supplies. Extension of the range of coals which can be used in the classical blast furnace, whether by blending or core fundamental techniques such as the

manufacture of formed coke, will tend to curb the extent to which such developments may be expected.

The use of gaseous and liquid hydrocarbons in the blast furnace is economically attractive at the present time and incentives for intensifying their use will be pressed. In the case of injection of the hydrocarbons, relatively little capital investment is involved and the practice can readily be changed in the face of varying fuel availability. The more elaborate systems based on reforming hydrocarbons will undoubtedly require considerable investment and it would seem that the time required to pay off such schemes is of similar order to the time during which such fuels will remain cheap in the face of approaching exhaustion of reserves. If such techniques are successfully applied it may be doubted whether 'second generation' schemes will equally be economically advantageous and it is possible that solid-fuel-fired blast furnaces may be built at the end of the present century without benefit of hydrocarbon injection. Top-gas recirculation schemes which have made little progress to date may be then more attractive as they conserve fossil fuel.

Direct-reduction processes based on natural gas or oil seem likely to have a limited life and to be restricted to specific regions where such fuels are abnormally cheap. Solid fuel processes are in principle much more attractive and of course the most effective and reliable one is the blast furnace. The electric low-shaft furnaces could well become more important as the cost of electricity declines relative to other energy sources.

In the longer term it may be expected that the iron industry will be brought to depend increasingly on nuclear energy. Present ideas for employing thermal energy from atomic reactors to support processes otherwise based on gaseous and liquid fuels are extremely difficult to evaluate. By the time they could be in a state to make a serious impact on increasing the natural fuels could well be approaching exhaustion necessitating their replacement by substitutes from coal. The economics of such a situation are complex but it is not likely that they will be attractive vis-à-vis the blast furnace.

In the really long term various factors such as the unattractive nature of the coal mining industry, atmospheric pollution by combustion and perhaps exhaustion of fossil fuel resources would be expected to lead to a technology based on reduction of iron oxide by hydrogen derived from water with the aid of nuclear power.

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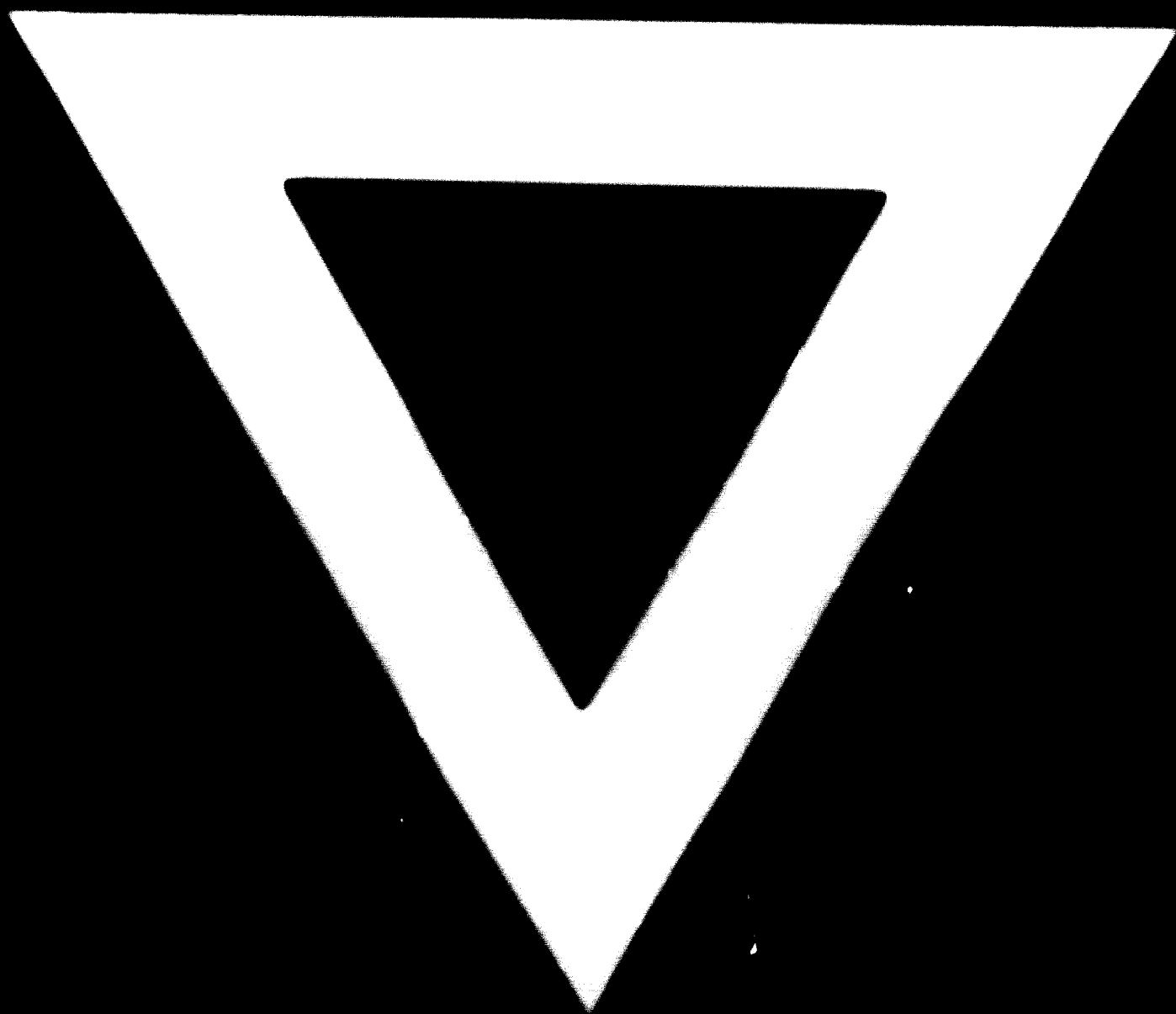
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Type of briquette	D.S.P.	D.S.P.	P.S.P.	D.S.P.
Coal source	Gravel	Gravel	S.S.	Gravel
Analysis H_2O or moisture	1.0	1.0	1.7	1.1
FD dry	81.5	81.2	80.9	82.7
ash	5.6	12.1	5.5	12.6
S	9.1	6.0	1.9	6.5
S	12.9	1.0	0.7	0.5
Ball density kg/m^3	578	602	585	779
Flow - 60	0	0		
Flow - 30			0	
Flow - 20				0
Flow - 10	9.3	10.9	5.1	5.6
Carbon loss (see test) %	11	1.7	0.9	0.6

TABLE II

	Coal Source	Calorific Qualities
Name	Allerton	Cartwood Millstone
Rank (H.C.B.)	638	601
(E.E.C.)	711	635
Moisture % (as received)	14.6	7.5
Ash % (dry)	9.3	5.1
Sulphur % (dry)	1.12	1.98
Volatile Matter (dry)	38.2	33.4
(daf)	35.5	35.2
Fixed Carbon (dry)	58.5	61.5
Calorific Value (as rec'd) cal/g.	6070	7680
Swelling Number	1.0	9.0
Proportions	75%	15%



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