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MANUFACTURING OF SYNTHETIC RUBBER IN ALEXAMDRIA PETROCHEMICAL COMPLEX 1

by

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I INTRODUCTION

One may ask why Egypt should at all consider installing a synthetic rubber plant when it can buy natural rubber from other developing countries or import synthetic rubber from other developed ones. Egypt as well as other developing countries may be justified in considering such a project for the following reasons:-

- a A synthetic rubber project helps in the economy of a petrochemical complex as the one planned to be installed in Alexandria by providing an outlet for some of the products produced from naphtha cracking and which otherwise have to be burnt and valued as fuel oil.
- b The foreign hard currency required for importing rubber natural or synthetic is reduced or saved if locally evailable raw materials can be used.
- c Producing costs for some synthetic rubbers are normally lower than those of natural rubber.
- d A synthetic rubber has a uniform specified quality and does not suffer from frequent price fluctuations and interruption of supply like the natural rubber.

It can be said that new technical advances made in the production of stereo-regular rubbers have enhanced the acceptability of synthetic rubber in areas where natural rubber has maintained a prime position. It is clear that the developing countries will mainly install synthetic stereo-regular plants in future and the cis-polyisoprene, being one of the low cost synthetic rubbers that can be used interchangeably with natural rubber in heavy-duty tires, is bound to receive increasing attention from all the countries that are planning to install new rubber plants as Egypt.

II The petrochemical complex in Alexandria

Since naphtha is available in excess than what is required for the local market requirements of motor care in Egypt, this naphtha could either be exported, burnt as fuel in furnaces or used as feedstock for petrochemicals. Both polyethylene and P.V.C. are badly required for the local market in Egypt for use in the following:

a - Bags for fertilizers produced from existing fertilizer plants amounting to about one million tone per year.

.../3

- b Making F.V.C. tubes for domestic use as well as for irrigation purposes.
- c Use for shoes manufacturing.
- d Use for sheathing of cables.
- Use in other different articles and services
 like tiles, pots, etc.

Since the required articles are either imported as finished products or raw synthetic material is imported and then fabricated to the different articles required for the local market, it was clear that the production of 45000 T/year of polyethylene and about 40000 T/year of polyethylene and about 40000 T/year of polyvinyl chloride could be easily consumed in Egypt leaving no surplus available in few years soon after the start up of the petrochemical complex when the consumption will equal to 1 kg F.V.C. per capita which is still too low compared to the consumption in Lebanon which is 6 kg/capita & in other european countries reching about 35 kg. per capita.

It was also found that the consumption of different types of rubber in Egypt in 1975 and 1980 is estimated as follows:

M. TONS IN YEAR 1975

	Natural R	S.B.R.	Polybutadiene	Butyl	
Tyres Co.	4500	1900	2200	1370	
Rubber articles Co's	750	750	•••	•	
Shees Manufact. Co's	400	800	•	•	
Cables Manufact, Co's	75	75	•	•	
Others	575	522	94		
Total	6300	4047	2200	1370	

TABLE II
M. Tons in year 1980

	Natural R	S.B.R.	Folybutadine	Butyl
Tyres Co.	6500	2650	3400	2050
Rubber articles Co's	1000	1500	•	**
Shoes Manufact, Co's	500	1000	•	•
Cables Manufact, Co's	100	105		
Others	575	522		
Total	8675	5777	3400	2050

The exising production of tires in Egypt is as follows:-

	- -
Passenger cars tires	300,000 Pieces
Trucks tires	170,000
Inner tubes	600,000 *
Bicyle tires	500,000 *

The main reasons for such low consumption of tires are the following:-

The total number of motor cars in the country (excluding the army vehicles) is 130,000 cars and 45,000 trucks which is very low for a country with 35 millions inhabitants.

- h The necessity of importing any type of rubber natural or synthetic in foreign hard currency prevents existing factories from extension and stops new factories from being erected. This naturally limits the production of new articles made from rubber or substituting other materials by rubber which is not easily imported.
- c The small number of motor cars produced by a single small car manufacturing company, at a higher cost than the imported ones. The high rate of customs dues imposed on the imported cars makes it very expensive for the majority of average persons to have their own cars.
 - d The dramatically cheap fares of public transport especially the railway fares compared to the cost of running a car or using a bus.

However, the above consumptions of the different types of rubber illustrate that the installation of a synthetic rubber plant for Egypt is an attractive and economic preposition.

It was therefore decided because of afore monthoned consumption figures that it is possible to erect a petrochemical complex in Alexandria based on cracking about 256000 T/year of naphtha produced from Amriya Refinery nearby. This will produce about 80000 T/year of ethylene which will be used for the production of 45000 T/year of polyethylene and about 40000 T/year of P.V.C. The other products that will be available from the steam cracking are the C3 cut and the C4 cut. It is estimated that 38400 T/year of propylene and 24800 T/year of G4 cut will be produced and about 10600 T/year of Butadiene could be extracted from the latter. Table 3 indicates the intake and production figures for the complex & Table 4 shows the estimated investment cost for the different units.

III Pyrclysis of S.R. Naphtha

Intake 1	M. I/year	& nt
S.R. Naphtha F.R.	256000	100
Production :		
H ₂	6864	2.7
CHy	36800	14.4
Bthylene	80000	31.2
Propane	3920	1.6
Propylene	38400	15.0
C4 cut	24800	9.7
Gasoline	52936	20,6
Fuel Oil	12280	4.8
Chlorine required for	F.V.C. 26400	

TABLE IV

ESTIMATED INVESTMENT COST FOR THE FETROCHEMICAL COMPLEX IN ALEXANDRIA

Stage A	T/year	LE. 10 ⁶
Ethylene craking	80000	13.789
Polyethylene low density	45000	22,777
Venyl chloride monomer	43000	7.975
Polyvenyl chloride	4000C	6,722
Butadiene extraction	10000	3,651
Polybutadiene	15000	5,907
Butadiene-Styrene (x)	15000	4,900
Sthylene glycol	10000	4,253
Polypropylene (x)	30000	8 .738

⁽x) Under consideration,

III CHOICE OF SYNTHETIC RUBBER SUITABLE FOR EGYPT

The choice of any particular rubber will obviously depend on considerations of strict competationess and of economic consideration resulting from raw material availability, monomer synthesis cost and processing difficulties.

Although S.B.R. is one of the cheapest versatile compounds and oil extended S.B.R. is used for passenger car tires as well as footwear, conveyer belts, cable insulation, hose and foam products yet the main difficulty with S.B.R. tire compounds is its high heat generation under dynamic conditions. These compounds also have power hot tear resistance than natural rubber compounds. With the increase of tire thickness and size, releasing tires from moulds during manufacture and dissipation of heat in the running of the tires become difficult problems. In the developing countries where the greatest use of rubber is for large bus and truck tires, it is reported that S.B.R. is not a suitable substitute for natural rubber.

Of the conventional synthetic rubbers, butyl, shloroprene have established markets and their consumption is increasing. Butyl rubber a copolymer of a small

percentage of isoprene with isobutylene (and sometimes disyclopentadiene) is relatively impermeable to air and has good resistance to heat and oxygen. Virtually all inner tubes in U.S.A. & Europe are made from butyl rubber. However, butyl rubber consumption for tires did not increase well because of excessive tread wear.

IV STEREO REGULAR RUBBERS

Three groups of stereoregular synthetic rubbers have been developed and have shown great potential for rapid growth; these are: cis-polybutadiene, cis-polyisoprene and ethylene propylene rubbers. The latter will not be discussed in this paper since its consumption is still very limited.

I - POLYBUTADIENE RUBBER

P.B. and proof of this lies in the large number of plants that are planned. Compared with 100% SBR, PB has better wear, heat build-up and groove crack resistance. The only falling off is noticed in resistance to chipping. This may be improved with more knowledge about its compounding and blending with natural rubber in 50-50 for heavy duty tires and in 25-75 blends with SBR for passanger tires.

Lithium and organolithium compounds seem to be different from other alkali metals and produce polybutadiene with ca 35% cis 1-4 and 50% trans 1-4 structure. After the introduction of the Zeigler type catalyst such as Aluminium triethyltitanium tetraiodide it was possible to produce polybutadiene with more than 90% cis 1-4 structure. When this was blented with 20-25% S.B.R. it showed improved abrasion and crack resistance in tire treads. Cis 1-4 polybutadiene is characterized by high abrasion resistance, low temp, properties and good resistance to ageing, Blends with more than 50% cis 1-4 Polybutadiene with S.B.R. produce problems in tread tearing, chipping, skid resistance and general ride quality. Its consumption could therefore be estimated by assuming that about 25% of rubber in all automobile tire treads can be of cis-polybutadiene and that tire tread constitute about 60% of the weight of the rubber in a tire. Thus if 50% of the new rubber goes in automobile tires, then 8-10% of the total new rubber could be assigned to cis-polybutadiene. The solution polymerization tachnique for cis polybutadiene using Zeigler type catalyst is similar to that for cis polyisoprene.

2 - POLYISOFRENE

This rubber is the best known to use in tire side walls and footwear. Cis-polyisoprene which resembles natural rubber was obtained by polymerizing a sol. of isoprene in a solvent with either a lithium base catalyst or a Zeigler type catalyst such as aluminium trialkyl-titanium tetrachloride. In 1961 Shell produced cis-isoprene rubber with a cis content of 90-92% and in 1962 Goodyear's Natsyn produced its rubber with 95-96% cis content.

Studies have shown that polyisoprene rubber gives vulcanizates very similar to those of natural rubber. The vulcanizates have high resilience and low heat built up. Moreover it is a synthetic material of good processability and could be used after minor modifications and recipe in conventional rubber processing equipment. P.I. has the potentional of being of excellent uniformity and it is finding applications in areas previously held by natural rubber particularly where good mould flow, excellent colour and purity of polymer are of importance. Following table 5 gives the typical properties of natural rubber & polyisoprene vullanizates.

TABLE V

TYPICAL PROPERTIES OF POLYTSOPRENE & NATURAL RUBBER VULCANIZATES

	Tires I	_	Carca	
Natural rubber (No 1 smoked sheet)	100	944	100	•
Polyisoprene (natsyn 200)	-	100	••	100
ISAF black	50	50	-	**
Sulphur	2,25	2.25	3	3
Tensile strength (psi)	4300	4200	4775	4400
Blongation (%)	550	650	740	800
300% modulus (psi)	1800	1300	675	475
Shore A hardness	68	67	55	54
Rebound at 212°F (%)	76.4	74.6	84,5	83,5
Heat build up (OF) (Goodrich Flexometer)	54	54	29.0	30,5

These compounds contain conventional amounts of Zinc oxide, Stearic acid, antioxidents and accelerator.

Published tests indicate that polyisoprene is fully equivalent to natural rubber in gum rubber formulations. It is reported that disisoprene is chemically identical with natural rubber and also almost duplicate the compounded

physical properties of natural rubber; it has the advantage of being cleaner more uniform in quality than natural rubber available. It can be charged to the mill without size reduction or premastication and usually require less mixing time than premasticated natural rubber.

Oil extended polyisoprene is a good prospect on economic grounds and its latexes may well proove to be one of the most important outlets for polyisoprene.

V Methods of Isoprene production

Of the saveral methods used for the manufacture of isoprene (1, 2) two routes have been used commercially propylene dimerization (Goodyear S.D.) and dehydrogenation of isopentane (Houdry or Shell). The latter is similar to the well known dehydrogenation process for butadiene production from n-butane or n-butene. Butane or butene being structurally simpler can be dehydragenated to butadiene more selectively than isopentane or isopentene can to isoprene. Also n-butane is more plentiful than isopentane. The isopentane route is likely therefore to produce isoprene at a higher cost than the normal butane-butene route for butadiene.

In the goodyear S.D. process propylene is dimerized to 2 methyl-1 pentene in the presence of tripropyl aluminium catalyst; the former is isomerized to 2-methyl-2-pentene, which is then cracked in the presence of vapour phase catalyst to obtain isoprene and methane. This crude is then fractionated to yield a monomer of exceptionally high purity.

VI CHEMISTRY OF ISOPRENE

It is obtained from propylene 99.9% pure free from water & sulphur & other reactive substances in 3 reaction steps:-

- 1 Demerization of propylene to 2-methyl pentene-1.
- 2 Isomerization of 2-methyl pentene-1 to 2-methyl pentene-2.
- 3 Pyrolysis of 2-methyl pentene-2 to isopreme & methane.
- 1 2 CH₂ = CH-CH₃ 3000 p.s.i CH₂ = C CH₂-CH₂-CH₃

 propylene 2 methyl pentene-1.

 In the liquid phase & catalyst is tri-n-propyl alumium.

 It is an exothermic reaction = -23 Kcal. Selectivity

 is 90%. Trimer is the principal by product.

Isomerization is in vapour phase at atmospheric pressure & 100°C, over special silica alumina catalyst. $CH_2 = C - CH_2 - CH_2 - CH_3 - CH_3 - CH_3 - CH_2 - CH_3$ 2 methyl pentene-1 2 methyl pentene-2

= 1.5 Kcal exothermic

equilibrium coversion is = 85%

3 -
$$CH_2 = C - CH_2 - CH_3 - CH_2 = C - CH_3 - CH_2 + CH_4$$

2 methyl pentene-2 isoprene

= 15 Kcal endothermic

It is conducted at 660°C & atmospheric pressure, steam as diluent & HBR as a catalyst.

VII PRODUCTION COST OF ISOFRENE

Production cost estimates based on propylene price of 1.5-2.5 \$/1b show that isoprene can be produced from propylene to complete with butadiene, which is a well established chemical selling at 11-13 g/lb.

The following tables illustrate the elements of operating cost for a 20,000 T/year plant for isoprene production from Isobutylene or propylene. The use of propylene for the manufactures of synthetic rubber in Egypt will eleminate the necessity of installing a polypropylene unit whose product is not essential for Egypt and thus will save the polypropylene plant investment which is estimated to be 8.738 million pounds for a capacity of 30000 T/year. (Refer Table 4).

TABLE 1

BASED ON ISOBUTYLENE & PROFYLENE

MANUFACTURING COST AND SELLING FRICE

Variable charge	Unit cost \$	<u>Isobutyl</u> Annual	Annua 1	Propylene Annual Annual
		Quantity	cost (103g)	Quantity cost (103g)
Rew material			(40)	(10 p)
Isobutylene (tons)	24	21800	· 520	• -
Methynol (tons)	50	19800	990	
Propylene (tons)	50	-	•	31000 1550
Byproducts Residols(tons) 10	(9200)	(-921	
Investments in 10 ⁶ g	•	·	5,6	8,6
Utilities			,	
Electricity (10 ⁶ Kwh) 10	6350	63)
Steam (tons)	1.6	208000	333	}
Cooling water (10 ³ M ³) 6	2780	17	# 135000/y.
Process water (M3)	5.2	64000	13	}
Fuel produced (10 ⁶ K	cal) 1.5	(1600)	(-2)	}
Catalyst and chemica	ls		150	90
Labour men/shift			7	7
Seberies & wages (men/year	500		140	

.../20

Supervision	at Zox	or	8alaries	
•	and wan	A 6		35

and wages	35	_35
Total variable charges	2167	1950
Fixed charges	1290	1970
Manufacturing cost	3457	3920

SELLING FRICE	Isobutylen Annual cos	
Net cash flow at 20% of		
total investment	1120	1720
Depreciation provides	700	1070
Net income after taxes	420	650
Net income before taxes at 50%	840	1300
Manufacturing cost	3457	3920
Isoprene sales	4297	5220
Isoprene selling price \$ /ton	215	260
· · · · · · · · · · · · · · · · · · ·	or #/1b 9.8	or #/1b 11.8

VII ISOPRENE TOTAL CAPITAL INVESTMENT

	ø
Process units investment	5613000
Tankage & utilities	1293000
Total	6906000
Dimerization section	1501000
Isomerization section	1521000 707000
Pyrolysis section	1955000
Recovery section	1064000
Purification section	1659000
Total process & Utilities cost	6906000
General service facilities at 15% of above	1036000
Total fixed capital	7942000
Interest on constmetion loan at 7% year (x)	556000
Start up cost 10% of total process & utilities cost	794000
Working capital	750000
Total capital investment not including land	10042000
(x) Construction period of 24 months finance	d loan of
\$7.942 million for average of half of co	
period assumed.	

VIII PRODUCTION COST FOR 80 MILLION LB/YR AT 0.9 STREAM FACTOR

Direct ope		r unit cost	\$/1b 0.29
Labour	Operating 6 men/s		0.26
		process unit invest.	
	Control Lab. 20% of	operating labour	0.05
•	Total labour	•	0.60
Material	Raw & process		
	Propylene 3.0 £/1	Lb	6,85
	Tri-n-propyl alumin		0.93
	HR	65.0 \$/1b.	0,02
	N&OH .	3.0 d/1b.	
	Maintenance	3,8% of process unit invest.	0,25
	Operating	10% of operating labour	0,03
Total Mat	erial		8.09
Total uti	.lities	•	0.95
Total dis	ect cost		9,64
Plant ove	Main (* 1994) Phead Main Hally Said (1994)		0,48
Taxes &	neurance		0,20

Net producti	ion cost	10.47
Other	1.5 ½/1b.	(0.67)
Methane & others	0.7 \$/1b.	(0.56)
Propane-propylene	2.5 \$/1b.	(-0.06)
By-product credit		
Total production cost		11.76
Depriciation Interest on working capital		0.06
		0.99
Cash expendition		10,71
C & 4		O _• 39
Plant cost		10.32

.../24

IX WILLIES SHOMERY

	Dimerz.	1scmery Sect.	Pyrol. Sect.	Recov. Sect.	Purif Sect.	Total
Water (1000 gal/hr) cooling	69.2	9 9.5	241.0	55.8 (x)	142,5	608,0
Make up (steam plant)	0.8	3,3	3.4	1.4	4.7	13,6
Make up (cooling water)	1.4	2.0	4.8	1.1	2,8	12,1
Steam 1b/hr	5970	25600	26800	11350	36500	106220
Elect, power (KW) process	500	8	17	590 (x)	3	1118
Utilities	28	62	104	31	88	313
Fuel (10000 cu.ft/nr	20.4		55.3			75.7

(x) Includes requirements for refrigeration.

VII STEREOSFECIFIC POLYMERIZATION COST

Polymerization manufacturing cost is quite similar for the 3 stereo regular rubber : FB, PI & EM.

Next table 7 gives the elements for a 20000 T/year plant starting from Butadiene or isoprene or ethylene and propylene. The manufacturing cost for such a plant, including profits and taxes is given in table 4.

Polyisoprene and Folybutadiene selling prices may be estimated as follows:-

Monomers selling price (Isoprene & Butaliene) 220 \$/ton or 10 £/1b.

Polymerization manufacturing cost including profits and taxes at 50% 265 \$/ton or 12 \$/lb. Polymer selling price 485 \$/ton or 22 \$/lb.

TABLE X

20,000 Tons/Year Stereospecific folymerization plant Elements of Operating Cost

Investment (MMS)

8

(including start up expenses, laboratories, inital catalyst & chemical charges and paid up royalities).

Utilities requirements:

Electricity (106 kwh/year)	25300	
Steam (tons/jear)	136000	
Cooling water (103 M3/year)	7300	
Process Water (M3/year)	450000	
Nitrogen (103 M3)	8600	
Chemicals and catalysts (%)consumption	1,260,000	
Labour men per shift	20	

TABLE XI

MANUFACTURING COST INCLUDING PROFITS AND TAXES

<u>Unit</u>	Unit cost	Annual Quantity	Angual cost
[Hilities			
Electricity (10 ⁶ Kwh)	10	25,300	253
Steam (tons)	1.6	136,000	218
Cooling water (10 ³ M ³)	6	7,300	44
Process water (M3)	0.2	450,000	90
Nitrogen (10 ³ M ³)	2	8,600	17
Chemicals and catalyst			1,260
Labour	1		
Salaries and wages (men/shift)	5,000	80	400
Supervision at 25 percen of salaries & wages	t		100
Total variable charges		•	2,402
Fixed charges			1,840
Total manufacturing cost			4,242
Net cash flow at 20 perce	ent of total i	investment	1,600
Depreciation provides			1,000

Net income after taxes	60 0
Net income before taxes at 50%	1,200
Total manufacturing cost	4,242
Manufacturing cost including profits after taxes at 50 percent	5,442
Manufacturing cost including profits (\$\mathbb{Z}\/\text{ton})	272 (or \$/1b)12.3

N.B.: Manufacturing cost for a 40,000 tons/year plant is *estimated at 11 \$/1b.

IX- STYRENE - BUTADIENE RUBSER

This is the main synthetic rubber produced all over the world and it is made by one of the following processes.

- A) Emulsion process at 122 of (hot process)
- E) Emulsion process at 41 of (cold process)
- C) Solution process using alkyllithium catalyst
- D) Alfin (Alcohol & olefin) Solution process using complex sodium salt.

It is reported that in U.S.A. about 90% of all solid S.B.R. is made by a cold emulsion polymerization process.

Solution process using lithium catalysts and producing random copolymers is in commercial use. Alfin process is not widely used yet. Compared to emulsion S.B.R., solution S.B.R. has a narrower molecular distribution and a higher cisdiene content.

Solution S.B.R. requires lower amounts of an accelerator to obtain a given modulus and hardness than emulsion S.B.R. does, therefore when used in tires gives lower heat build up, higher resilience and better blowout resistance. Solution S.B.R. gives poorer mill binding, rougher extrusions and generally more difficult to process. By proper balancing however solution S.B.R. as presently made has significantly better properties than emulsion S.B.R. and is still not too difficult to process.

Alfin S.B.R. has a narrow molecular weight distribution, It is specially free from the low molecular weight portion and it has

a diene configuration of 69 % trans and 31% vinyl. Dut to high trans contact, it requires a higher milling temperature than other S.S.R. . Performance of alfin S.B.R. should be between that of emulsion and that of solution S.B.R.

It is reported that in 1960 world market will contain more solution S.B.R. than emulsion S.B.R. by about 50%.

TA W. AII

CAPITAL 1. VESTMELT IN 1000 DOLLARS FOR STYRENE - 3UTADJENE RUBBER BY COLD EMULSION PROCESS

Plant capacity

77.4 Million pounds/ year

Stream factor

0.9

			•••		
	Raw Material preparation	Polymeriza- tion	Monomer re- covery	Co- agulation & drying	
Battary limit invest- ment Utilities & tankage	5-65	1,914	2436	2249	73.84
investment Total Gereral service facilit at 15% of above	369 954 Jes	65 6 2 5 70	158 25 94	1511 3760	26 94 98 78
Total fixed capital Interest on construction loan at 9.5% **	'n				1482 11360 1079
Start up costs Total depreciable capit Working capital Total capital investmen not including land					800 1 323 9
, and the second					13739

x For one half the assumed construction period of 24 months.

TABLE XITI

STYRENE BUTADIEN UNITS OPERATING COST IN 1000 \$ / YEAR

105.3 7760.8	157.4 142.8	415.7
	7	· ·
I	, ·	
104.0	44.8	131.0
7970.5	328.7	1596.0
	7970.5	7970.5 328.7

TAILE XIV

PRODUCTION COST FOR STYRENE - BUTADIENE RUBBER BY COLD EMULSION PROCESS

	Basis or Unit cost	Total cost	1000 S/year
• • •		6/16	10003/year
Labor operating	12 men/shift, 4.90 S/man har.		513.6
Maintenance	2.5% yr of battery limit cost plus 1.5% of refrigeration cost	0.24	185.7
Control lab.	20% of operating cost	0.13	102.7
Total labor		1.03	802
Material (x) chemical	s etc.	11.30	8739.6
Material Maintenance	2.5% yr of buttery limit cost	0.24	185.7
Materials operating	10% of operating labour	0.07	51.4
Total Materials		11.61	8976.7
Ttilities			• 100 - 100 - 100 - 100 - 100 - 100 - 100 - 100 - 100 - 100 - 100 - 100 - 100 - 100 - 100 - 100 - 100 - 100 - 1
Cooling water	2\$/1000 Gal	0.02	12.9
Steam	55\$/1000 1bs	0.08	60.2
Power	1¢/Kw1	0.27	207.6
Inert gas	15\$/1000 sof	Name:	1.8
Deionized water	35\$/1000 gal	0.02	17•5
Total Utilities		0.39	300+0
Total direct op cost		13.03	10,073.0
Plant overhead	80% of total labour	0.82	641.6
Taxes & Insurance	20% yr of fixed capital	0.29	227.2
Plant cost		14.14	10,941.8
Research etc.	10% of Sales price	2.00	1,548.0
Cash expenditure		16.14	12,489.8
Depreciation :	10% yr of fixed capital	1.47	1,136.0
Interest on working	ospital *		
	6% / 30er	0.06	47.5
Total Production o	08 \$	17,67	13673,3
N.B. Plant capacit	y 77.4 million lbs/year & O	.9 stream fo	etor,

TABLE X V

COMPANISON BETWEEN DIFFERENT
TYPEC OF SYNTHETIC RUBBER

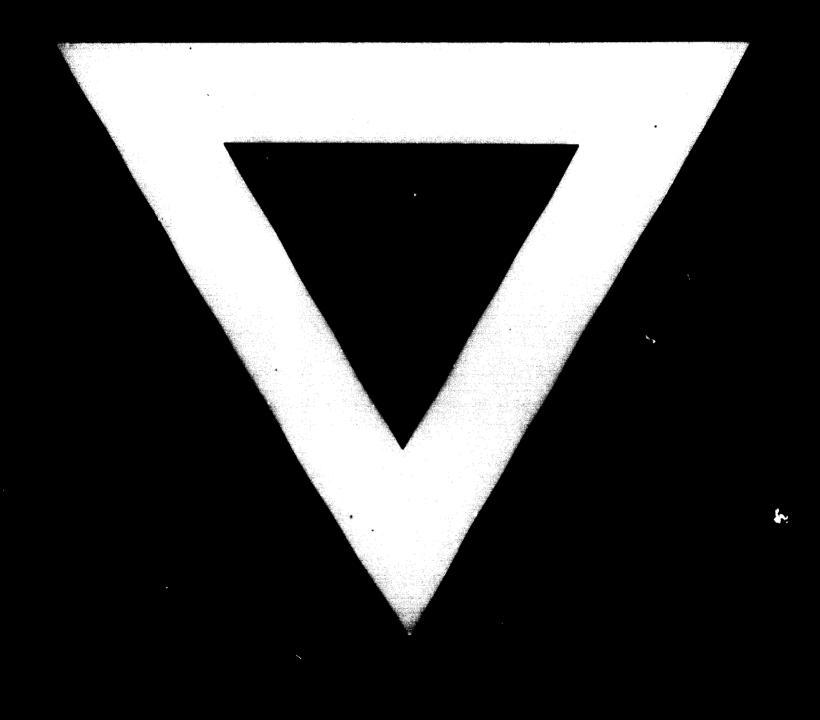
	% Used in Tires year 69 75	10 ³ Ton s in	Rate of market increase #	Price 6/1b. as in 1971
S.B.R.	58 5 5	i.135	4.5	13.515.6
Polybutadiene	72 70	260	7.7	20.522.0
Polyasoprene E.P.T.	89 66 10 5	230 95	22.3 31.0	2323.9
Batyl	84 81	153	6.3	25.327.2
Polychlorprene	1.21.2	126	7.7 () () () () () () () () () (

CONCLUSION

The choice of type of synthetic rubber plant in a developing country like Egypt is a difficult problem. The 2 alternatives of either building a Styrene Butadiene plant which can produce at the sametime polybutadiene or building a polyisoprene plant using available propylene for isoprene production. The latter alternative will make use of the available propylene produced from the naphtha cracking process istead of building a polypropylene plant to make use of such product. A high density polyethylene plant could be made to give the material required for the articles need to be made from the polypropylene unit especially when it is known that the low density polyethylene produced is much higher than what is required for the local market.

No doubt that the final judgement will depend on the outcome of economic, commercial and technical factors which will become more clear when the different offers for the different units are received by us in the second half of the current year.





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