



TOGETHER
for a sustainable future

OCCASION

This publication has been made available to the public on the occasion of the 50th anniversary of the United Nations Industrial Development Organisation.



TOGETHER
for a sustainable future

DISCLAIMER

This document has been produced without formal United Nations editing. The designations employed and the presentation of the material in this document do not imply the expression of any opinion whatsoever on the part of the Secretariat of the United Nations Industrial Development Organization (UNIDO) concerning the legal status of any country, territory, city or area or of its authorities, or concerning the delimitation of its frontiers or boundaries, or its economic system or degree of development. Designations such as “developed”, “industrialized” and “developing” are intended for statistical convenience and do not necessarily express a judgment about the stage reached by a particular country or area in the development process. Mention of firm names or commercial products does not constitute an endorsement by UNIDO.

FAIR USE POLICY

Any part of this publication may be quoted and referenced for educational and research purposes without additional permission from UNIDO. However, those who make use of quoting and referencing this publication are requested to follow the Fair Use Policy of giving due credit to UNIDO.

CONTACT

Please contact publications@unido.org for further information concerning UNIDO publications.

For more information about UNIDO, please visit us at www.unido.org

D00425

21



United Nations Industrial Development Organization

Interregional Petrochemical Symposium on the
Development of the Petrochemical Industries
in Developing Countries

Moscow, USSR, 20 - 31 October 1969

UNEP/WHO/UNEP/WHO/UNEP/WHO

FIN. ST. 7/1

SUMMARY

NEW POLYVINYLCHLORIDE FIBRES ^{1/}

G. Lanzetta
Montell S.p.A.
Milan, Italy

A new catalytic system for the polymerization of vinyl chloride at very low temperatures, down to -60°C , has rendered easily and economically feasible the industrial production of polyvinylchloride with a high degree of stereoregularity, i.e. more crystalline, resistant to solvents, and with improved mechanical properties at higher temperatures.

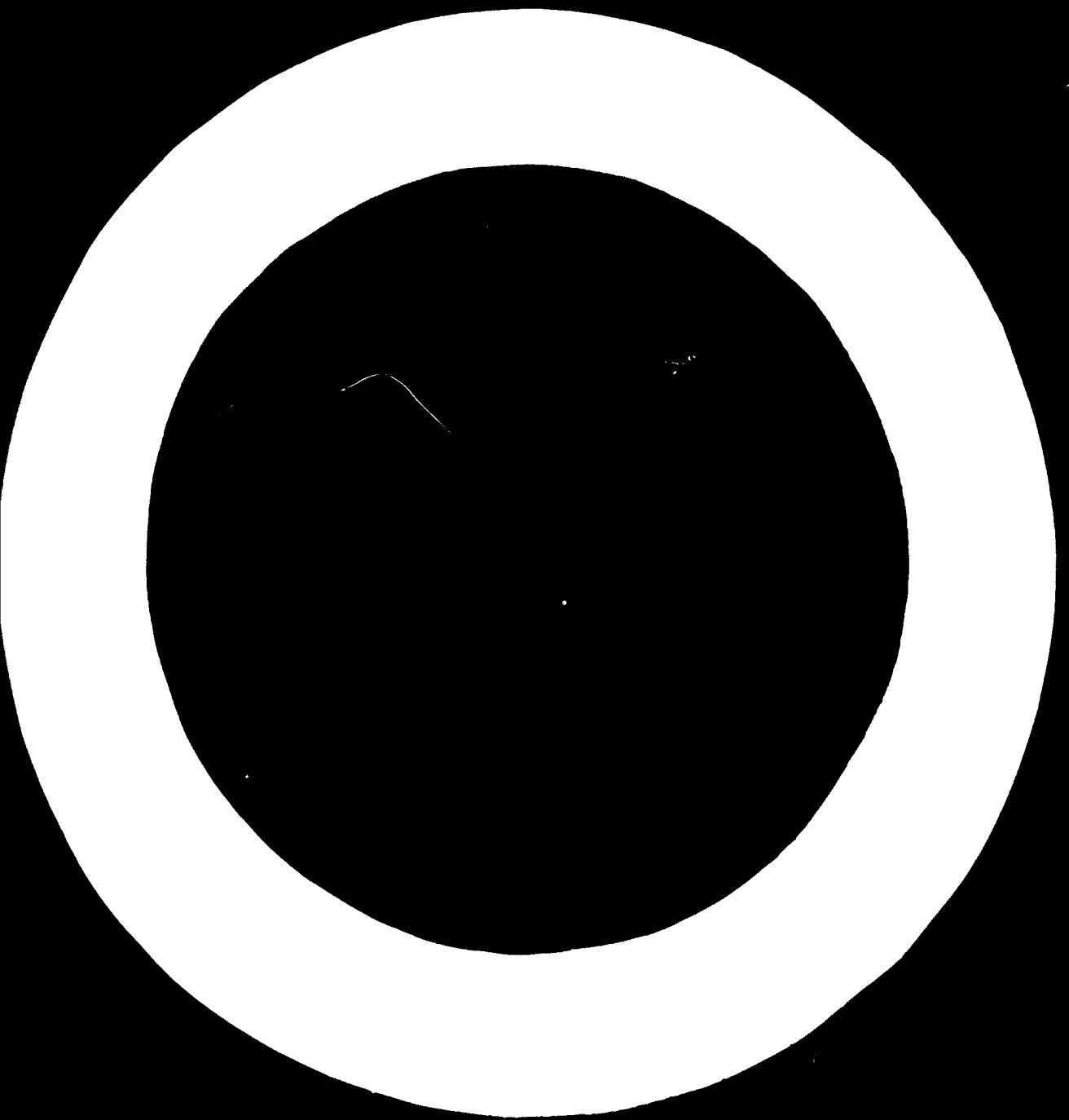
New techniques were developed to transform this polymer into fibres ranking by far above the traditional PVC fibres in their overall properties.

The initial shrinking temperature reaches 130°C , as compared to 65°C for traditional PVC fibres. The new fibre can be easily dyed at

^{1/} The views and opinions expressed in this paper are those of the author and do not necessarily reflect the views of the secretariat of UNIDO. This document has been reproduced without formal editing.

high temperature and is resistant to dry-cleaning solvents. Due to its inherent property of non-flammability, joined with good processability and a pleasant wool-like hand, the new fibre is highly suitable for a wide range of textile applications.

We regret that some of the pages in the microfiche copy of this report may not be up to the proper legibility standards, even though the best possible copy was used for preparing the master fiche.





with a low... well
...
...
...
...

...
...
...
...
...

...
...
...
...
...

...
...
...
...
...

...
...
...
...
...
...
...

...
...
...
...
...
...
...

...
...
...
...
...
...
...

... have, ... color is, ... consumption in the ...

... can, ... the ...

... stress, ...

... mechanical ...

... article ...

... and ...

... instrument ...

... of the fibers ...

... idea as to the ... elastic modulus ... fibers ... than 100.

The basic modulus or Young modulus is the ratio, or more exactly the first derivative of stress with respect to strain, $E = \frac{d\sigma}{d\epsilon}$, where σ is the stress and ϵ the strain. In the case of a linear elastic material, E is constant and is expressed in units of kg/cm^2 .

Another characteristic of the material is its Poisson's ratio, ν , which is the ratio of the lateral contraction to the longitudinal extension. It is expressed in units of cm/cm .

The elongation at break, ϵ_b , is the strain at which the material finally ruptures. It is expressed in units of cm/cm .

The yield point, σ_y , is the stress at which the material begins to deform permanently. It is expressed in units of kg/cm^2 .

The ultimate tensile strength, σ_u , is the maximum stress that the material can withstand before rupture. It is expressed in units of kg/cm^2 .

The modulus of resilience, U_r , is the energy absorbed per unit volume of the material up to the yield point. It is expressed in units of kg-cm/cm^3 .

It is clear that the same material may have different characteristics depending on the conditions of use. For example, a material may have a high modulus of elasticity but a low elongation at break.

The first step in the mechanism of the reaction is the formation of a stabilized carbon radical. This radical is formed by the reaction of the initiator with the monomer, which leads to the formation of a radical. The radical then reacts with the monomer to form a polymer chain. The reaction is exothermic and the heat released is used to maintain the reaction temperature. The reaction is also self-accelerating, as the heat released increases the rate of reaction.

The following steps in the mechanism are the propagation and termination steps. The propagation step involves the reaction of the radical with the monomer to form a new radical. This step is exothermic and the heat released is used to maintain the reaction temperature. The termination step involves the reaction of two radicals to form a stable molecule. This step is also exothermic and the heat released is used to maintain the reaction temperature. The overall reaction is exothermic and the heat released is used to maintain the reaction temperature.

The mechanism of the reaction is similar to that of other free-radical polymerizations. The reaction is initiated by a radical, which reacts with the monomer to form a polymer chain. The reaction is exothermic and the heat released is used to maintain the reaction temperature. The reaction is also self-accelerating, as the heat released increases the rate of reaction.

A particular feature of the polymerization is the fact that the reaction is exothermic and the heat released is used to maintain the reaction temperature. This is a characteristic feature of free-radical polymerizations. The reaction is also self-accelerating, as the heat released increases the rate of reaction. The reaction is also exothermic and the heat released is used to maintain the reaction temperature.

of the dielectric temperature dependence. All these trends, found by means of infrared analysis and other methods, are, in principle, corresponding to the structure of the polymer, according to the corresponding theory. The results of the present work, however, are in full agreement with the above-mentioned theory.

The infrared analysis of the polymer, which is a polyethylene, shows that the absorption bands in the region of 1000-1300 cm⁻¹ are characteristic of the polymer chain. The absorption bands in the region of 1450-1600 cm⁻¹ are characteristic of the polymer chain. The absorption bands in the region of 2800-3000 cm⁻¹ are characteristic of the polymer chain.

The infrared analysis of the polymer, which is a polyethylene, shows that the absorption bands in the region of 1000-1300 cm⁻¹ are characteristic of the polymer chain. The absorption bands in the region of 1450-1600 cm⁻¹ are characteristic of the polymer chain. The absorption bands in the region of 2800-3000 cm⁻¹ are characteristic of the polymer chain. The absorption bands in the region of 3000-3600 cm⁻¹ are characteristic of the polymer chain. The absorption bands in the region of 3600-4000 cm⁻¹ are characteristic of the polymer chain.

The infrared analysis of the polymer, which is a polyethylene, shows that the absorption bands in the region of 1000-1300 cm⁻¹ are characteristic of the polymer chain. The absorption bands in the region of 1450-1600 cm⁻¹ are characteristic of the polymer chain. The absorption bands in the region of 2800-3000 cm⁻¹ are characteristic of the polymer chain. The absorption bands in the region of 3000-3600 cm⁻¹ are characteristic of the polymer chain. The absorption bands in the region of 3600-4000 cm⁻¹ are characteristic of the polymer chain.

The infrared analysis of the polymer, which is a polyethylene, shows that the absorption bands in the region of 1000-1300 cm⁻¹ are characteristic of the polymer chain. The absorption bands in the region of 1450-1600 cm⁻¹ are characteristic of the polymer chain. The absorption bands in the region of 2800-3000 cm⁻¹ are characteristic of the polymer chain. The absorption bands in the region of 3000-3600 cm⁻¹ are characteristic of the polymer chain. The absorption bands in the region of 3600-4000 cm⁻¹ are characteristic of the polymer chain.

It is well known that the polymerization of an
oxide can be carried out in a number of ways. The most common
method is the use of a catalyst. The catalyst is usually
a metal compound, such as a metal salt, which is added to
the reaction mixture. The catalyst is usually added in
the form of a solution. The catalyst is usually added in
the form of a solution. The catalyst is usually added in
the form of a solution.

The polymerization of an oxide can be carried out in a
number of ways. The most common method is the use of a
catalyst. The catalyst is usually a metal compound, such
as a metal salt, which is added to the reaction mixture.
The catalyst is usually added in the form of a solution.
The catalyst is usually added in the form of a solution.
The catalyst is usually added in the form of a solution.

The polymerization of an oxide can be carried out in a
number of ways. The most common method is the use of a
catalyst. The catalyst is usually a metal compound, such
as a metal salt, which is added to the reaction mixture.
The catalyst is usually added in the form of a solution.
The catalyst is usually added in the form of a solution.
The catalyst is usually added in the form of a solution.

The polymerization of an oxide can be carried out in a
number of ways. The most common method is the use of a
catalyst. The catalyst is usually a metal compound, such
as a metal salt, which is added to the reaction mixture.
The catalyst is usually added in the form of a solution.
The catalyst is usually added in the form of a solution.
The catalyst is usually added in the form of a solution.

The polymerization of an oxide can be carried out in a
number of ways. The most common method is the use of a
catalyst. The catalyst is usually a metal compound, such
as a metal salt, which is added to the reaction mixture.
The catalyst is usually added in the form of a solution.
The catalyst is usually added in the form of a solution.
The catalyst is usually added in the form of a solution.
The catalyst is usually added in the form of a solution.
The catalyst is usually added in the form of a solution.
The catalyst is usually added in the form of a solution.

Anybody will tell you that the polymerization of an
oxide can be carried out in a number of ways. The most
common method is the use of a catalyst. The catalyst is
usually a metal compound, such as a metal salt, which is
added to the reaction mixture. The catalyst is usually
added in the form of a solution. The catalyst is usually
added in the form of a solution. The catalyst is usually
added in the form of a solution.

The first thing I noticed when I stepped out of the car was the cold. It was a crisp, clear cold, the kind that makes you pull your coat tighter around you. The air smelled like pine and snow. I had never before.

The snow was falling in soft, gentle flakes, catching in the air like tiny white dancers. I looked up, trying to catch one in my hand, but they melted before I could. The ground was already covered in a thin layer, and the trees were coated in white.

I had never seen anything like this before. The houses were small and cozy-looking, with glowing windows. The streets were quiet, with only a few people walking. It felt like a secret world, hidden away from the rest of the world.

I walked down the street, feeling the snow under my boots. The air was so clean, so fresh. I had never before. The snow was falling in soft, gentle flakes, catching in the air like tiny white dancers. I looked up, trying to catch one in my hand, but they melted before I could. The ground was already covered in a thin layer, and the trees were coated in white.

The snow was falling in soft, gentle flakes, catching in the air like tiny white dancers. I looked up, trying to catch one in my hand, but they melted before I could. The ground was already covered in a thin layer, and the trees were coated in white. I had never before.

I had never before. The snow was falling in soft, gentle flakes, catching in the air like tiny white dancers. I looked up, trying to catch one in my hand, but they melted before I could. The ground was already covered in a thin layer, and the trees were coated in white.

I had never before. The snow was falling in soft, gentle flakes, catching in the air like tiny white dancers. I looked up, trying to catch one in my hand, but they melted before I could. The ground was already covered in a thin layer, and the trees were coated in white.

The first part of the report is a general description of the
 and is in a very general way. The second part is a description
 like an account of the work done in the field of the
 and is in a very general way. The third part is a description
 and is in a very general way. The fourth part is a description
 and is in a very general way. The fifth part is a description
 and is in a very general way. The sixth part is a description
 and is in a very general way. The seventh part is a description
 and is in a very general way. The eighth part is a description
 and is in a very general way. The ninth part is a description
 and is in a very general way. The tenth part is a description
 and is in a very general way.

The second part of the report is a description of the
 and is in a very general way. The third part is a description
 and is in a very general way. The fourth part is a description
 and is in a very general way. The fifth part is a description
 and is in a very general way. The sixth part is a description
 and is in a very general way. The seventh part is a description
 and is in a very general way. The eighth part is a description
 and is in a very general way. The ninth part is a description
 and is in a very general way. The tenth part is a description
 and is in a very general way.

The third part of the report is a description of the
 and is in a very general way. The fourth part is a description
 and is in a very general way. The fifth part is a description
 and is in a very general way. The sixth part is a description
 and is in a very general way. The seventh part is a description
 and is in a very general way. The eighth part is a description
 and is in a very general way. The ninth part is a description
 and is in a very general way. The tenth part is a description
 and is in a very general way.

The fourth part of the report is a description of the
 and is in a very general way. The fifth part is a description
 and is in a very general way. The sixth part is a description
 and is in a very general way. The seventh part is a description
 and is in a very general way. The eighth part is a description
 and is in a very general way. The ninth part is a description
 and is in a very general way. The tenth part is a description
 and is in a very general way.

1911, 1912

with a fine fiber, no or a coarse one, that it was rubbed,
and that it was rubbed with a piece of paper with negative electricity.
The effect of the rubbing was to produce a certain amount of
charge on the surface of the paper, which was then rubbed with a
piece of paper, which was rubbed, and then, a certain
amount of charge was produced.

The effect of the rubbing was to produce a certain amount of
charge on the surface of the paper, which was then rubbed with a
piece of paper, which was rubbed, and then, a certain amount of
charge was produced. The effect of the rubbing was to produce a
certain amount of charge on the surface of the paper, which was
then rubbed with a piece of paper, which was rubbed, and then,
a certain amount of charge was produced. The effect of the rubbing
was to produce a certain amount of charge on the surface of the
paper, which was then rubbed with a piece of paper, which was
rubbed, and then, a certain amount of charge was produced.

The effect of the rubbing was to produce a certain amount of
charge on the surface of the paper, which was then rubbed with a
piece of paper, which was rubbed, and then, a certain amount of
charge was produced. The effect of the rubbing was to produce a
certain amount of charge on the surface of the paper, which was
then rubbed with a piece of paper, which was rubbed, and then,
a certain amount of charge was produced.

The effect of the rubbing was to produce a certain amount of
charge on the surface of the paper, which was then rubbed with a
piece of paper, which was rubbed, and then, a certain amount of
charge was produced. The effect of the rubbing was to produce a
certain amount of charge on the surface of the paper, which was
then rubbed with a piece of paper, which was rubbed, and then,
a certain amount of charge was produced.

The effect of the rubbing was to produce a certain amount of
charge on the surface of the paper, which was then rubbed with a
piece of paper, which was rubbed, and then, a certain amount of
charge was produced. The effect of the rubbing was to produce a
certain amount of charge on the surface of the paper, which was
then rubbed with a piece of paper, which was rubbed, and then,
a certain amount of charge was produced. The effect of the rubbing
was to produce a certain amount of charge on the surface of the
paper, which was then rubbed with a piece of paper, which was
rubbed, and then, a certain amount of charge was produced.

white to yellow coloration of the wool fibers, and of various types, most often during the period of the winter season.

Table 7 reports the results of the tests of the wool fibers. In the course of the tests, the wool fibers were subjected to a number of references in order to determine the effect of the various treatments on the wool fibers. The results of the tests show that the wool fibers are very sensitive to the action of the various treatments, and that the wool fibers are very sensitive to the action of the various treatments. The results of the tests show that the wool fibers are very sensitive to the action of the various treatments, and that the wool fibers are very sensitive to the action of the various treatments.

The above table lists the results of the tests. This determination indicates the brittleness of the wool fibers, and the percentage values reported for the wool fibers are based on the results of the tests.

The duration of the tests, the wool fibers are subjected to be one of the fibers in various conditions of the wool fibers, and the wool fibers are subjected to be one of the fibers in various conditions of the wool fibers, and the wool fibers are subjected to be one of the fibers in various conditions of the wool fibers.

Table 8 reports the results of the tests of the wool fibers, and the wool fibers are subjected to be one of the fibers in various conditions of the wool fibers, and the wool fibers are subjected to be one of the fibers in various conditions of the wool fibers, and the wool fibers are subjected to be one of the fibers in various conditions of the wool fibers.

Not only does the wool fibers are subjected to be one of the fibers in various conditions of the wool fibers, and the wool fibers are subjected to be one of the fibers in various conditions of the wool fibers, and the wool fibers are subjected to be one of the fibers in various conditions of the wool fibers, and the wool fibers are subjected to be one of the fibers in various conditions of the wool fibers.

1004.34/75
1004.34

The next step (ie. 1004.34) is the addition of levil in trichloro-
ethylene, which is the most widely used degreasing solvent, as compared
with the other media. The effect of levil on the stability of the
emulsion is given in Table 1. The results show that the stability

of the emulsion is improved as the concentration of levil
increases. The effect of levil on the stability of the emulsion is
shown in Table 1. The results show that the stability of the
emulsion is improved as the concentration of levil increases. The
effect of levil on the stability of the emulsion is shown in Table 1.
The results show that the stability of the emulsion is improved as
the concentration of levil increases.

When levil is added to the emulsion in the case of levil, we
will find that the emulsion is more stable. Like other
degreasing solvents, levil is a non-polar solvent. Levil
is a non-polar solvent, and it is known that non-polar solvents
are more effective in removing oil and grease from surfaces. This
is because the non-polar solvent is more similar to the structure of the
oil and grease, and it is able to penetrate the surface more easily.
The results show that the stability of the emulsion is improved as
the concentration of levil increases.

The effect of levil on the stability of the emulsion is shown in
Table 1. The results show that the stability of the emulsion is
improved as the concentration of levil increases. The effect of
levil on the stability of the emulsion is shown in Table 1. The
results show that the stability of the emulsion is improved as the
concentration of levil increases. The effect of levil on the
stability of the emulsion is shown in Table 1. The results show
that the stability of the emulsion is improved as the concentration
of levil increases. The effect of levil on the stability of the
emulsion is shown in Table 1. The results show that the stability
of the emulsion is improved as the concentration of levil increases.
The effect of levil on the stability of the emulsion is shown in
Table 1. The results show that the stability of the emulsion is
improved as the concentration of levil increases. The effect of
levil on the stability of the emulsion is shown in Table 1. The
results show that the stability of the emulsion is improved as the
concentration of levil increases. The effect of levil on the
stability of the emulsion is shown in Table 1. The results show
that the stability of the emulsion is improved as the concentration
of levil increases.

118 It is intended to be used, not only for the production of yarns which are filamentary in character, but also for the production of non-filamentary yarns, such as those used in the manufacture of paper, etc. (General characteristics of the various types of filamentary yarns are given in the following table.) The filamentary yarns are characterized by the fact that they consist of a number of filaments, which may be of different lengths, and which are twisted together to form a yarn. The non-filamentary yarns are characterized by the fact that they consist of a number of filaments, which are of different lengths, and which are twisted together to form a yarn. The filamentary yarns are characterized by the fact that they consist of a number of filaments, which may be of different lengths, and which are twisted together to form a yarn. The non-filamentary yarns are characterized by the fact that they consist of a number of filaments, which are of different lengths, and which are twisted together to form a yarn.

The filamentary yarns are characterized by the fact that they consist of a number of filaments, which may be of different lengths, and which are twisted together to form a yarn. The non-filamentary yarns are characterized by the fact that they consist of a number of filaments, which are of different lengths, and which are twisted together to form a yarn.

The filamentary yarns are characterized by the fact that they consist of a number of filaments, which may be of different lengths, and which are twisted together to form a yarn. The non-filamentary yarns are characterized by the fact that they consist of a number of filaments, which are of different lengths, and which are twisted together to form a yarn.

The filamentary yarns are characterized by the fact that they consist of a number of filaments, which may be of different lengths, and which are twisted together to form a yarn. The non-filamentary yarns are characterized by the fact that they consist of a number of filaments, which are of different lengths, and which are twisted together to form a yarn.

The removal of the original components of the starting fibre, which is
in the case of synthetic fibres the spinning effects were obtained
by the removal of the original components of the starting fibre, which is
in the case of synthetic fibres the spinning effects were obtained

... (text is extremely faint and largely illegible due to low contrast and blurring) ...
... (text is extremely faint and largely illegible due to low contrast and blurring) ...
... (text is extremely faint and largely illegible due to low contrast and blurring) ...

... (text is extremely faint and largely illegible due to low contrast and blurring) ...

... (text is extremely faint and largely illegible due to low contrast and blurring) ...

... (text is extremely faint and largely illegible due to low contrast and blurring) ...
... (text is extremely faint and largely illegible due to low contrast and blurring) ...
... (text is extremely faint and largely illegible due to low contrast and blurring) ...
... (text is extremely faint and largely illegible due to low contrast and blurring) ...
... (text is extremely faint and largely illegible due to low contrast and blurring) ...

surround us with safety, and Leavel will be well protected for those next exposed to fire dangers. On millinery, for example, the head and children. Leavel is in some respect ideal for children's wear. It is near, in fact, considered with the finest of the modern chemical reaction or other means to protect the garment, and sweater, sweater, etc., all of which have the same advantages. The fabric is soft, comfortable, and easily washed in view of the fact that it is very quickly dried.

Combining the fibers of Leavel with the excellent properties of Uretil and other fibers, which are highly durable, and the high quality of the aesthetic appearance, which is in line with the modernity - which is particularly important for the bulky and heavy garments such as that of heavy fabric.

Many types of fabric, which are used for personal use, are equally good for use in the Leavel.

Leavel's chemical resistance is enhanced by the protective clothing, which is used in some cases in the presence of corrosive substances, and the industrial applications, such as filters and felts, which are used in high temperature conditions.

We shall conclude this report with a review of the trials of blends of Leavel with other fibers, which constitute a very successful combination of the properties of the components.

The first of such blends are the Leavel/Leavel blends, suitable for general use. The high quality of even in staple fibers is well known: they have incomparable fiber resistance, which remains over time the right, dense appearance of the garment. If Leavel is introduced in the blend to the extent of 20% - 30%, depending on the type of garment, is not only a mixture of already excellent wear characteristics, but also it gives it the aesthetic appearance; and of course it provides full security against possible flame-spreading dangers.

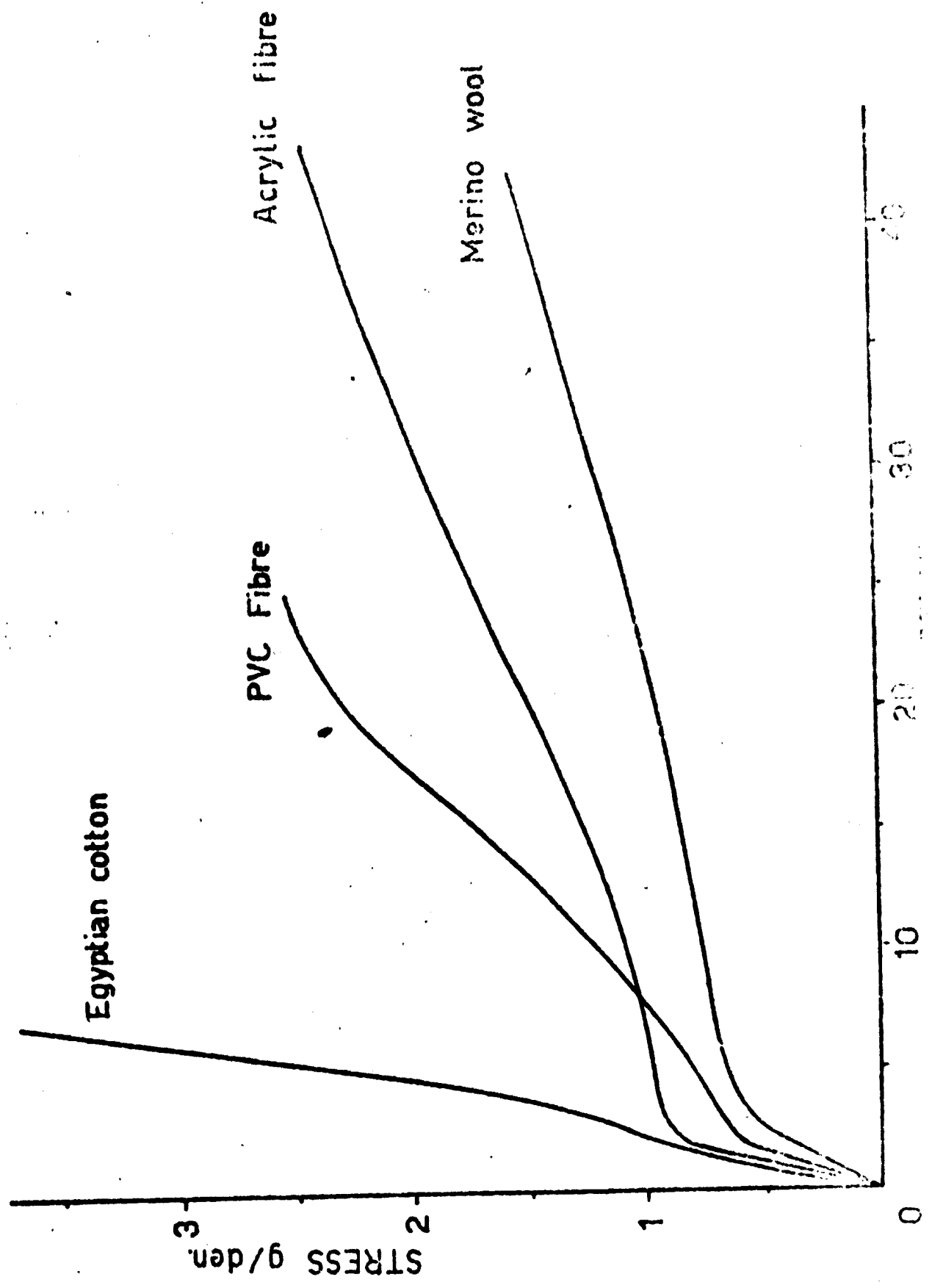
1/20/55
10-11

The use of blends has, involving wool and traditional PVC fibres, such as wool, and is of particular interest because they show how the introduction of new synthetic fibre, such as, as a general rule, over a period of time, for those already existing materials, in certain cases, presents a new way of doing things. In the production of blends of wool and synthetic fibres, the fibre, due to its different properties, undergoes a treatment which causes the shrinkage of the traditional PVC fibre, but causes the loss of fur, and increases the diameter of the fibre, and thus causes a loss of weight, thus perfectly imitating the fibre of certain animals. If fibres exist which differ in their nature and blends, it becomes possible to produce the stitching by the self surface application of wool fibres.

Wool fibres are being produced for over one year in an industrial plant of the Textile Institute of Chetillon, France, with a capacity of 100,000 spindles, and provisions for increasing the capacity are already under way.

34/55

STRESS STRAIN CURVES AT 21°C AND 65% R.H. FIG.1



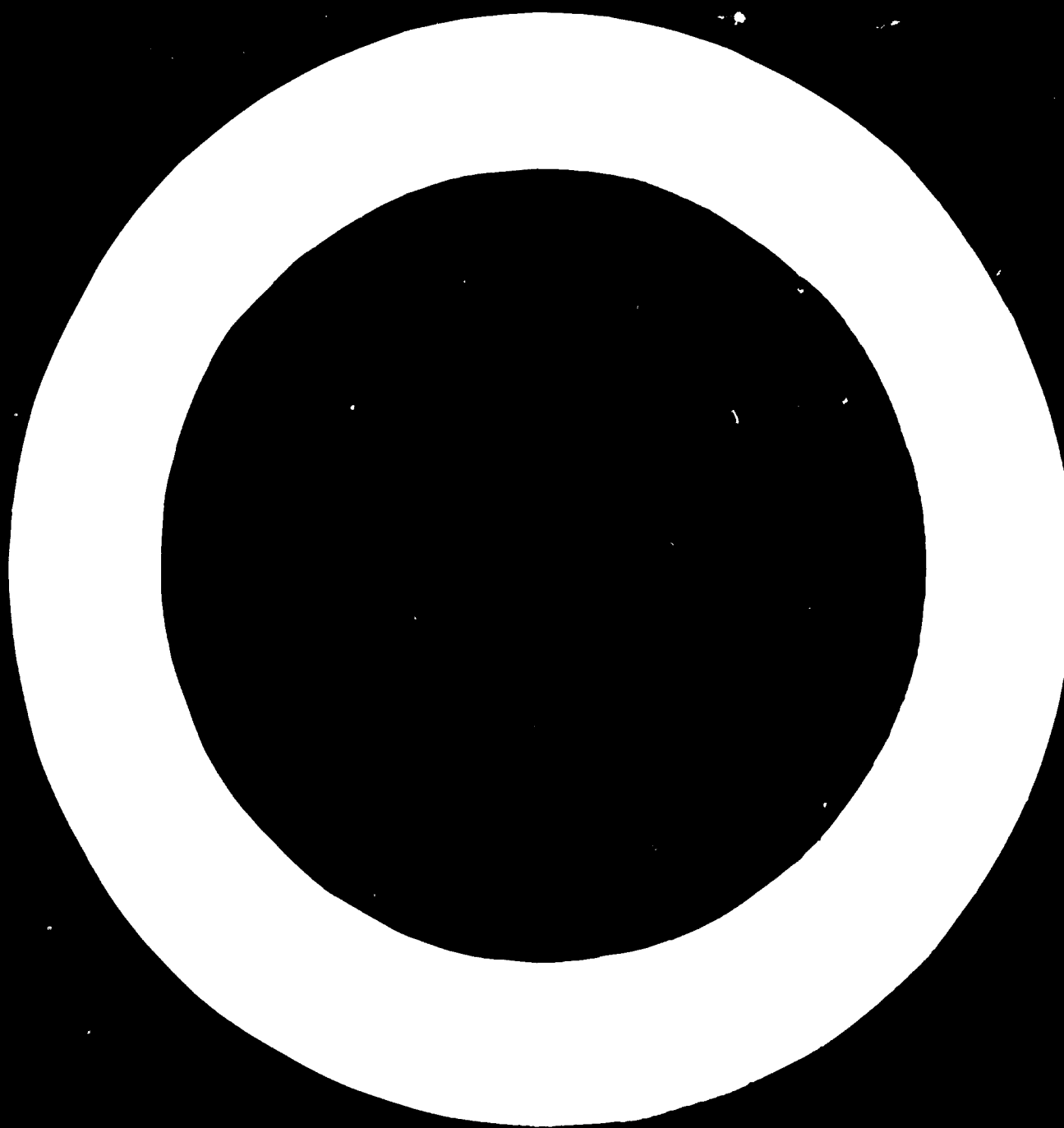
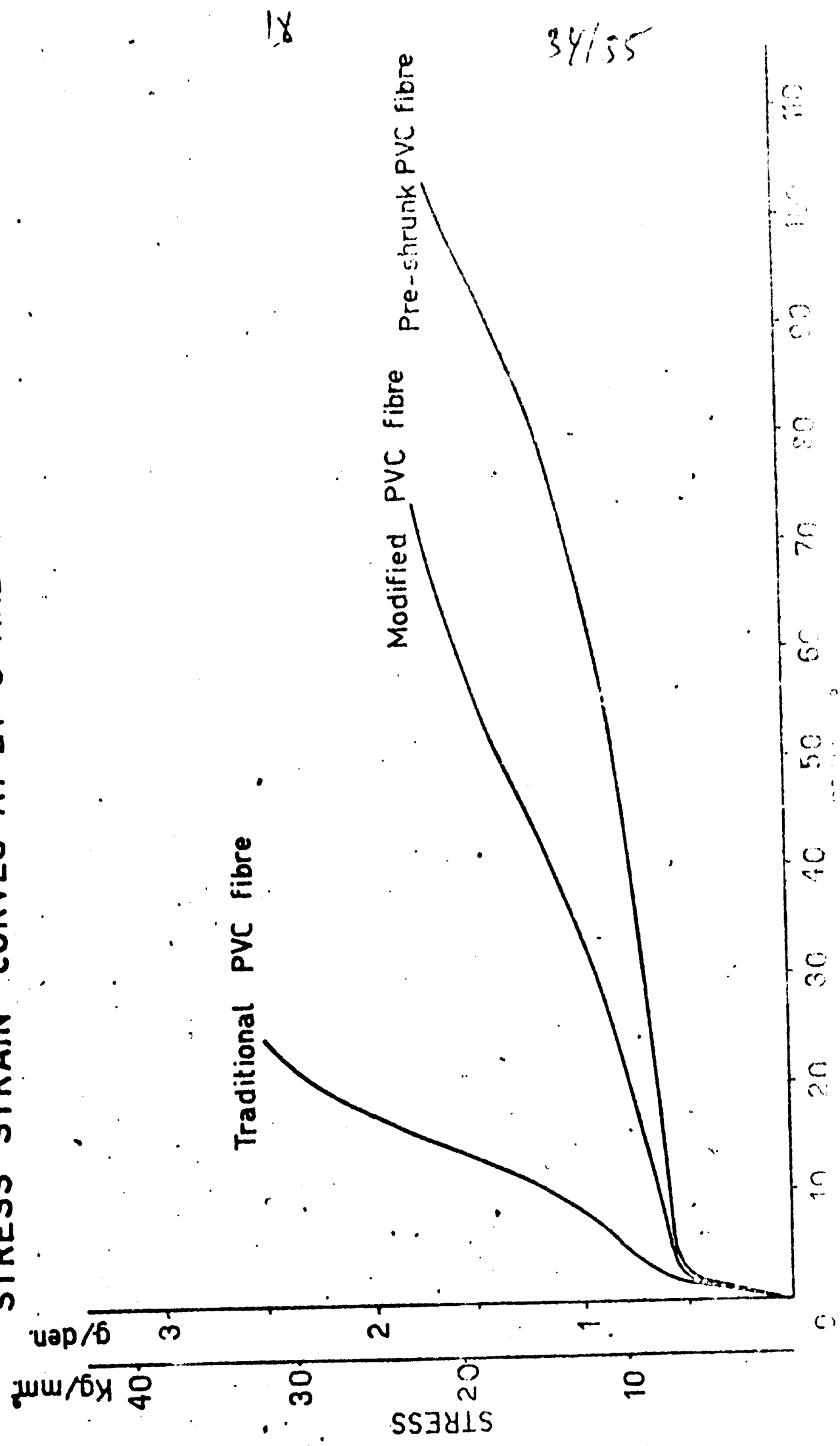
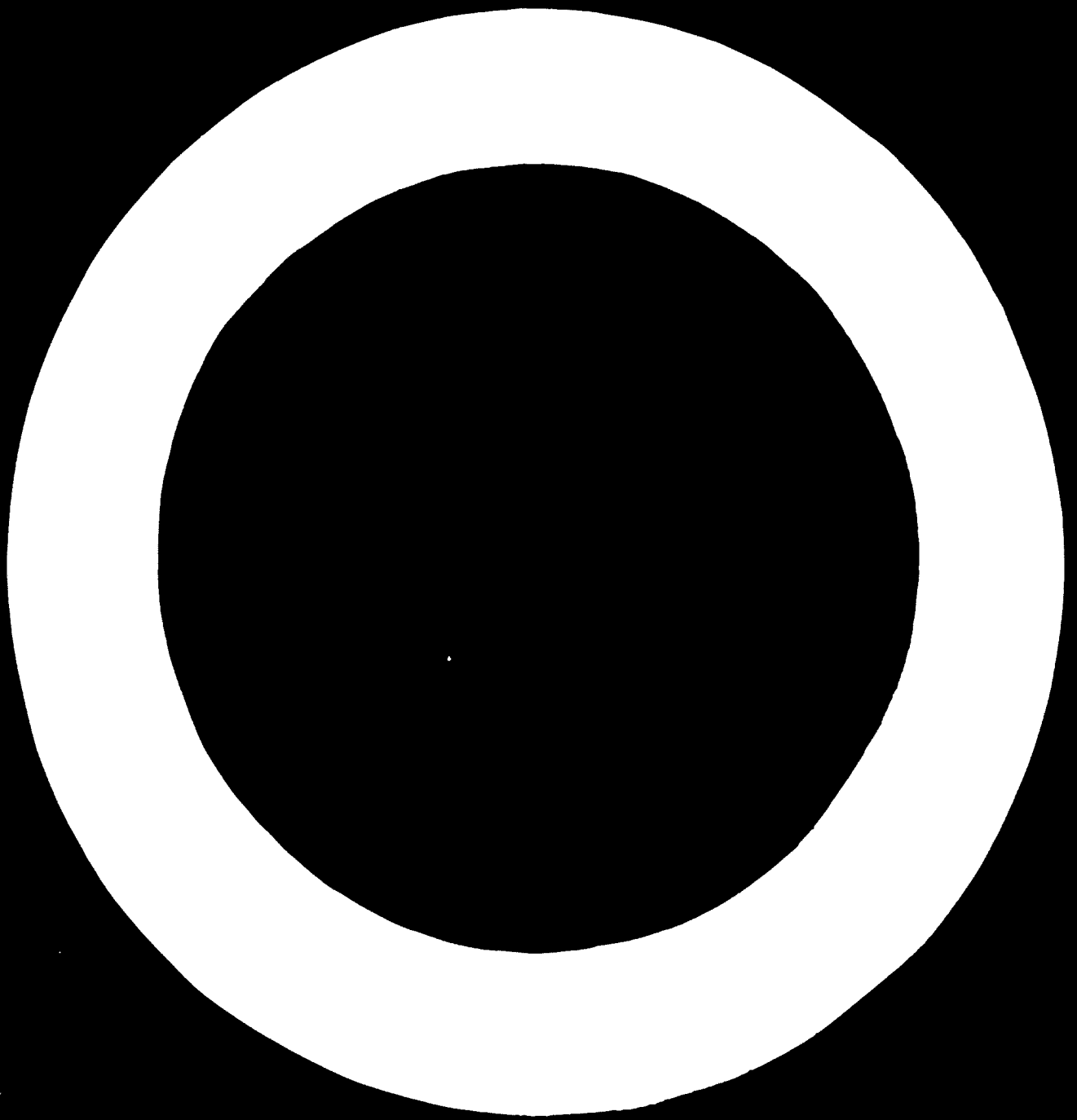


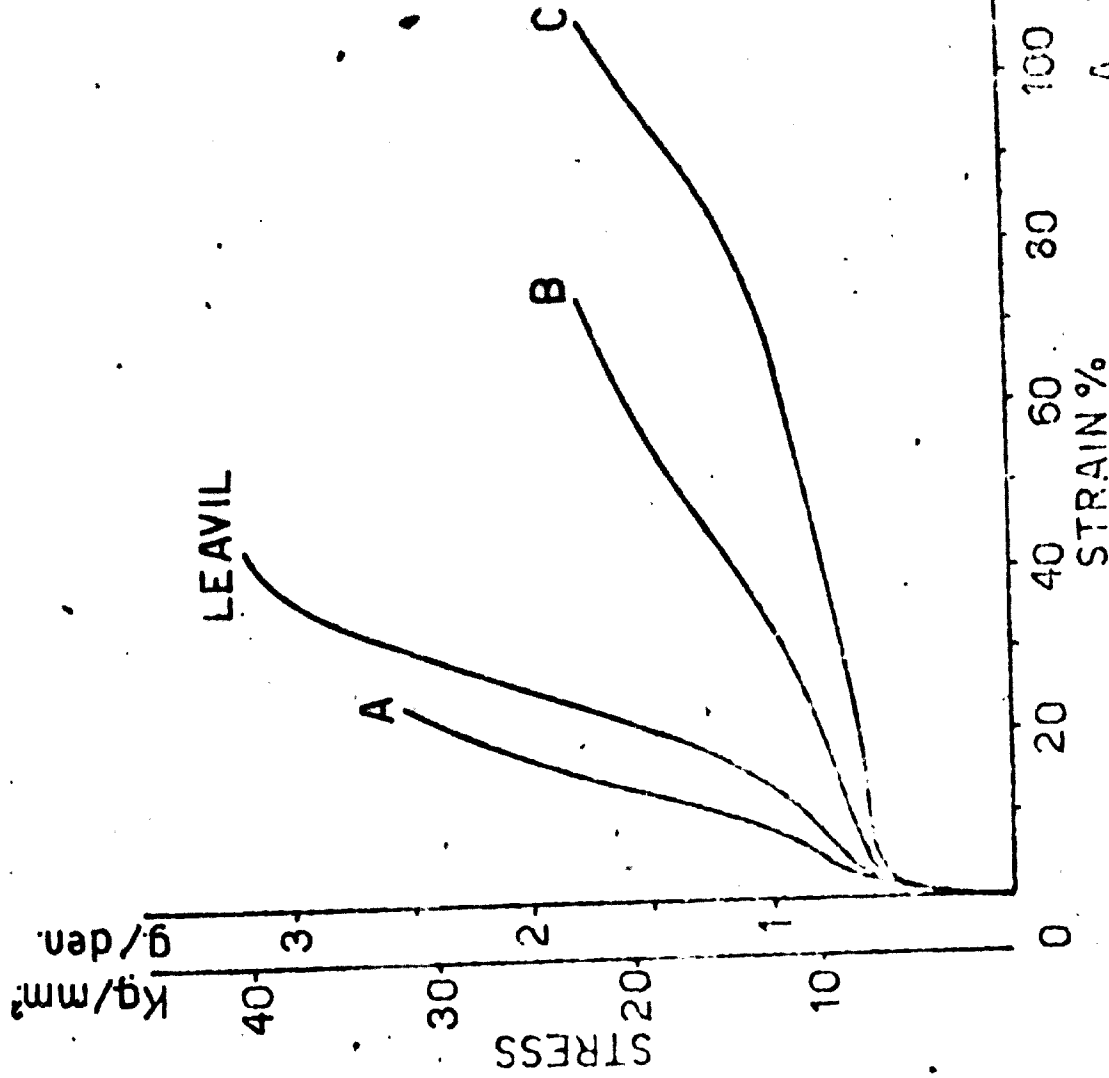
FIG. 2

STRESS STRAIN CURVES AT 21°C AND 65% R.H.

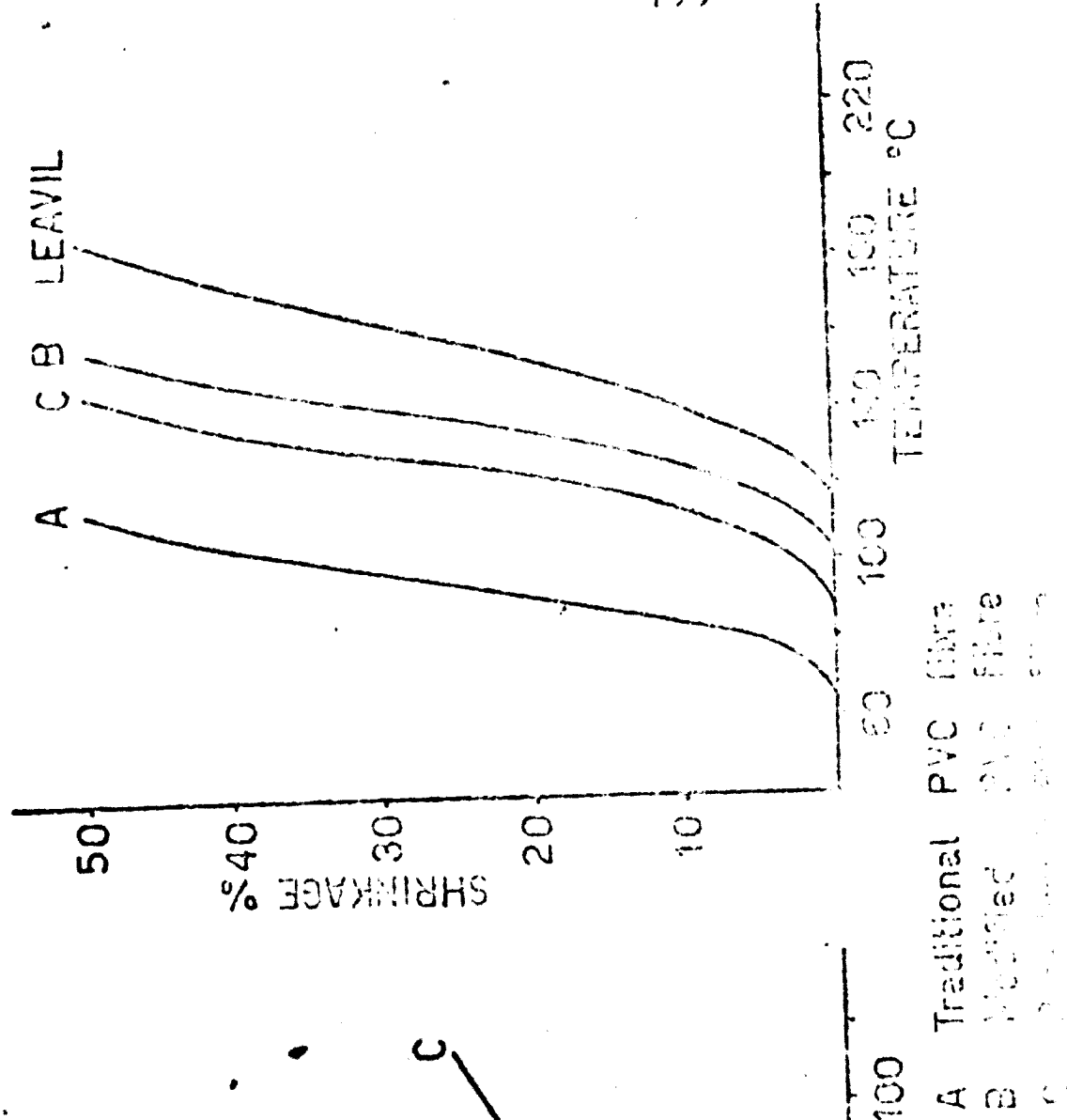


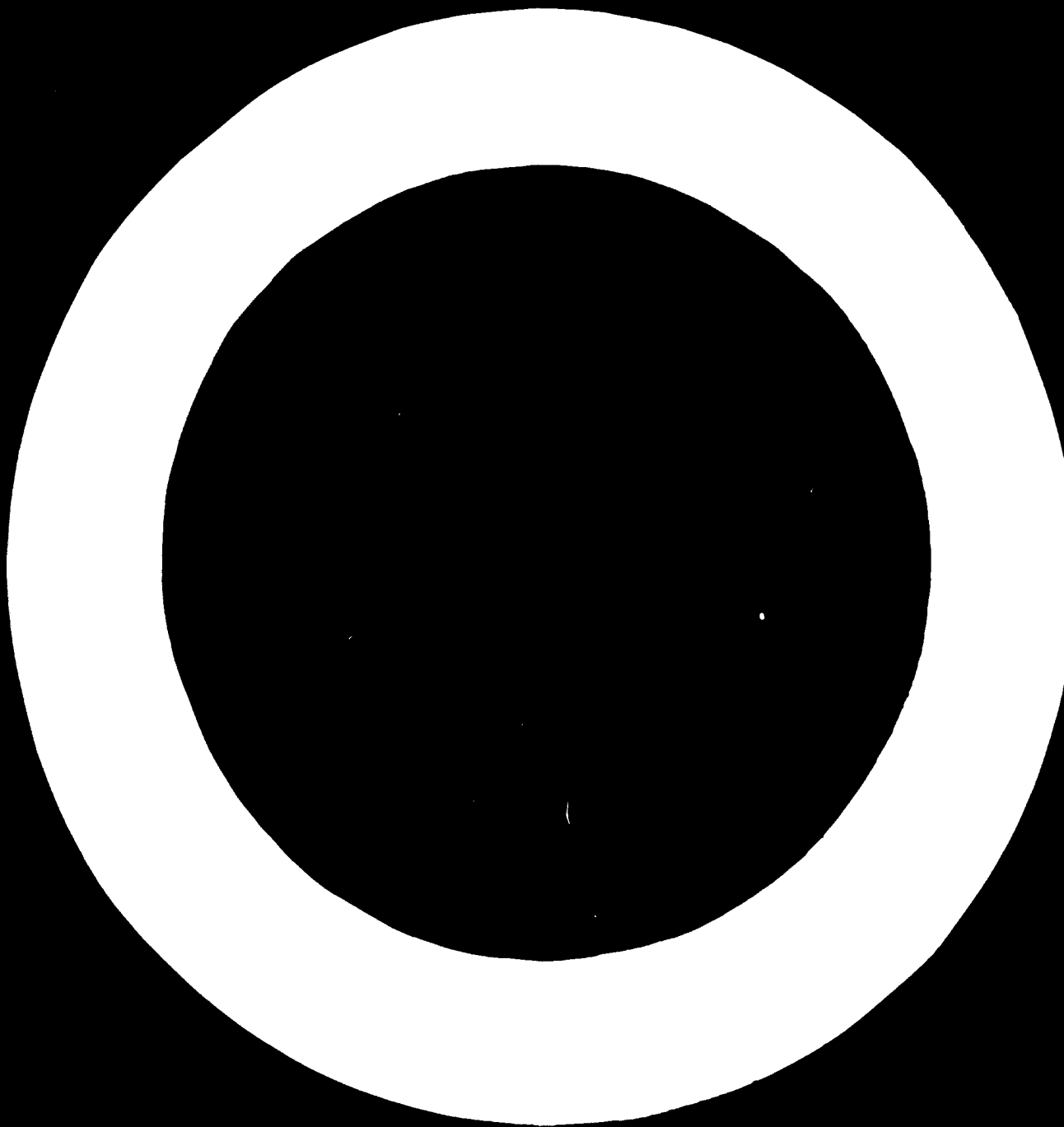


STRESS STRAIN CURVES
AT 21°C AND 65% R.H.



EMPIRICAL RELATIONSHIPS
OF PVC FIBRES IN AN INERT
MEDIUM (SILICON OIL)



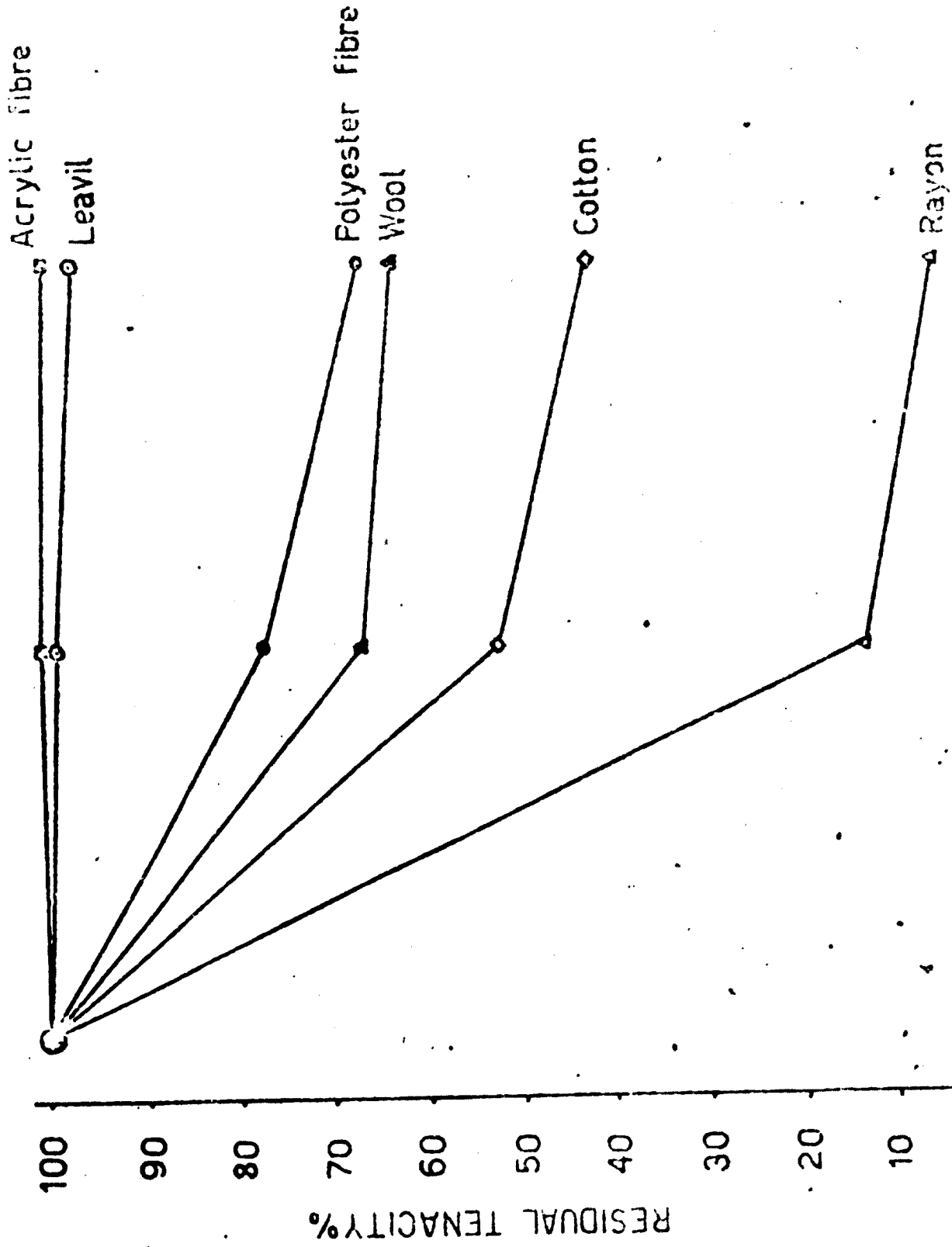


SUNLIGHT FASINESS

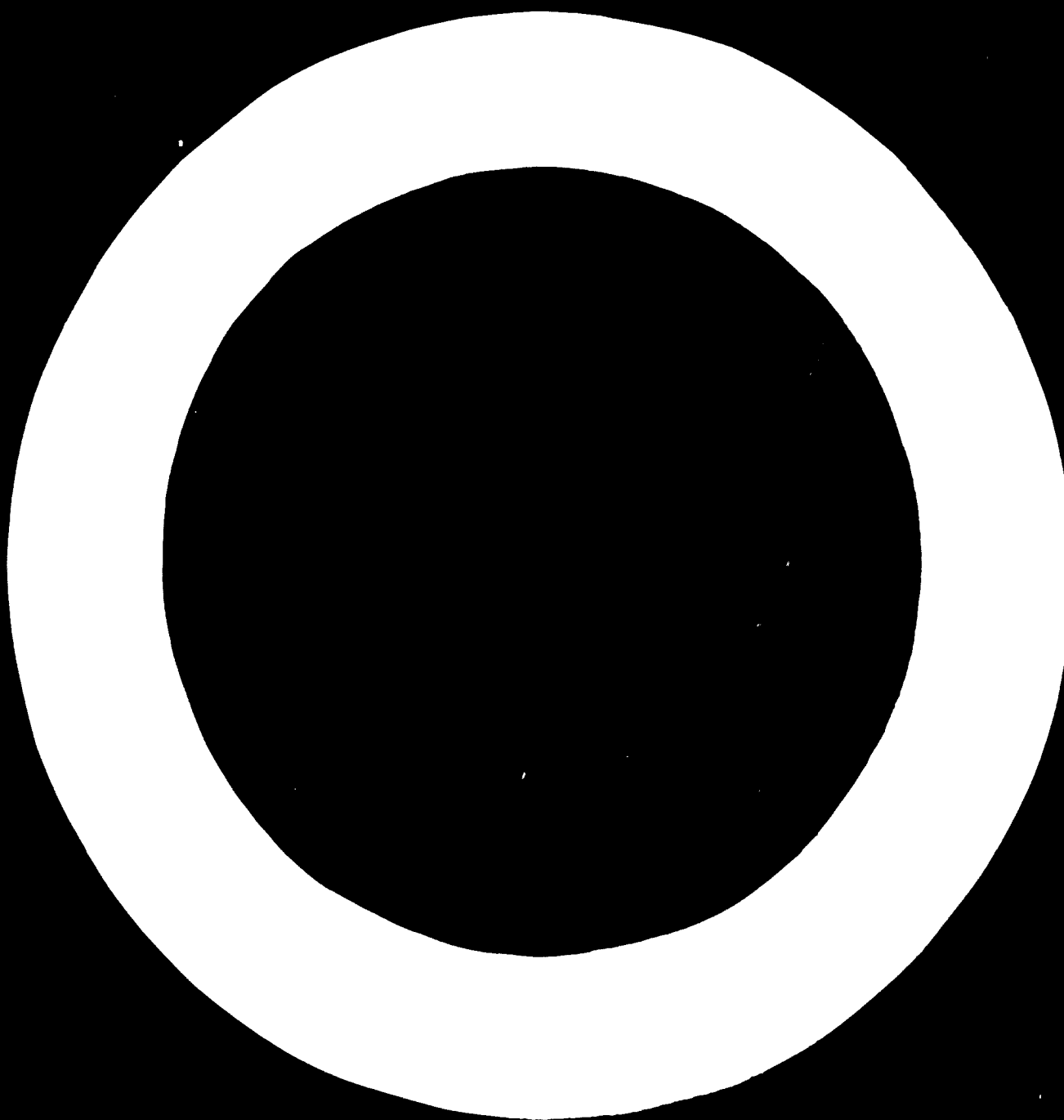
10000

20

34/55



MONTHS OF SUNLIGHT

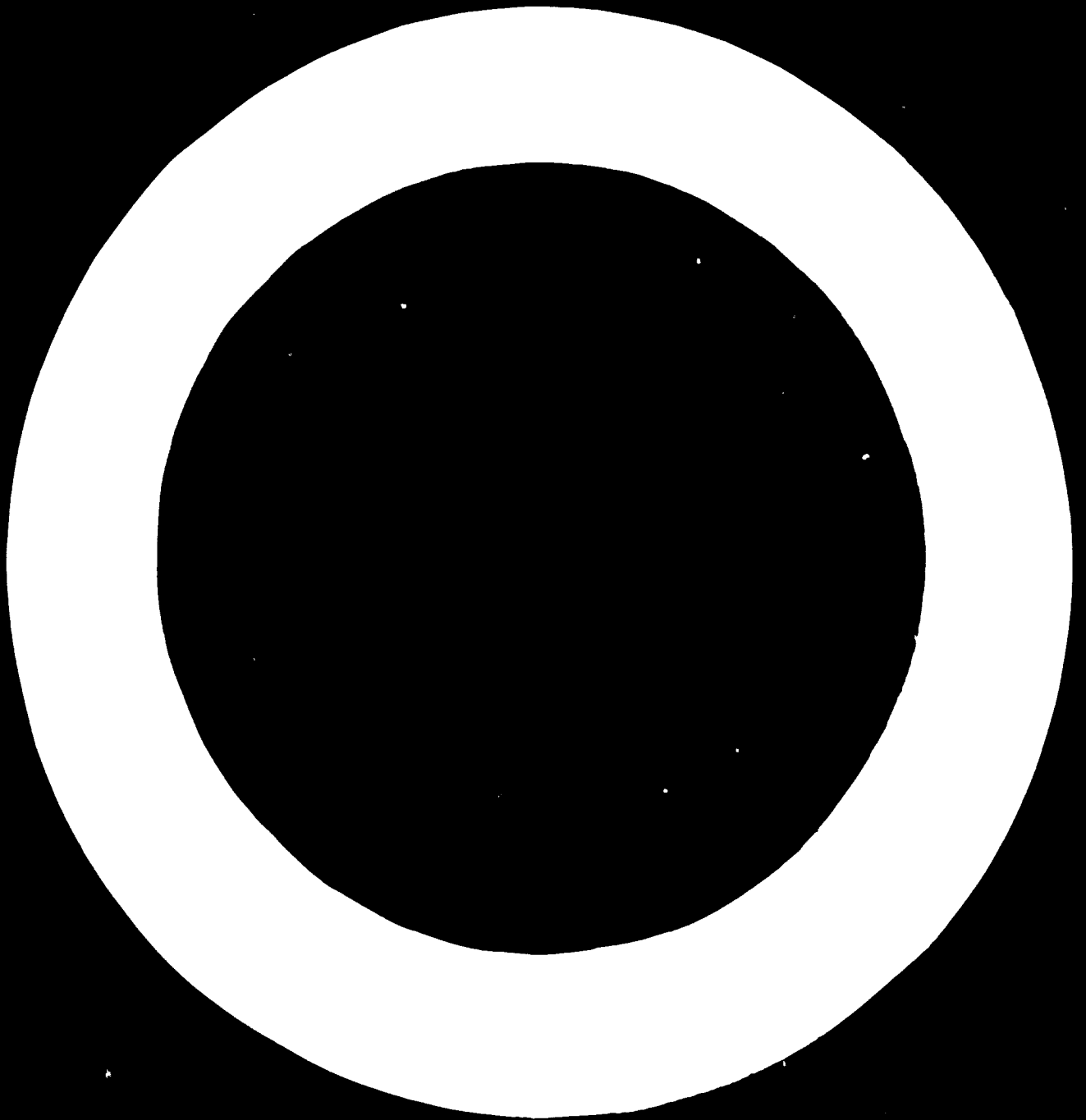


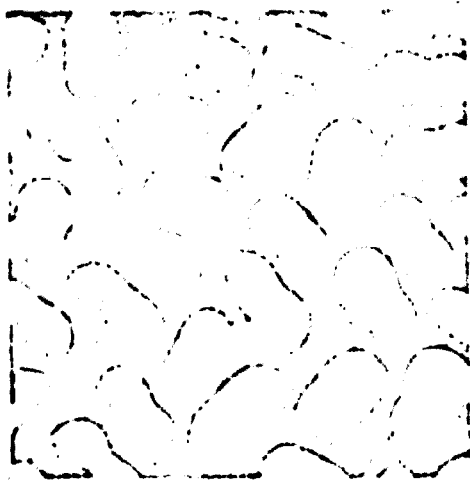
34/55

21

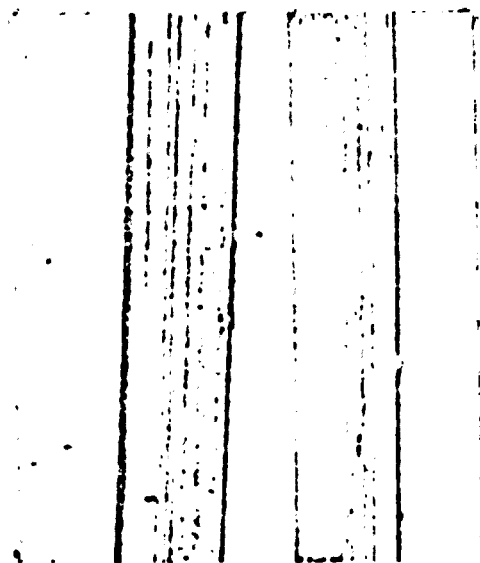
Fig. 5 - HEAVN resistance to chemical agents

Solution	Test conditions	Weight loss %	Tensile strength p/den	Elong. %	Warpage (inches)	Color
-	as such	-	3.3	40	30	-
2% HNO ₃	168 hours at 20°C	0.45	2.85	50	30	-
2% H ₂ SO ₄	168 " " 20°C	0.20	3.3	50	29	-
2% HCl	168 " " 20°C	0.00	3.3	47	30	-
2% NaOH	168 " " 20°C	0.20	3.2	49	27	-
2% H ₂ SO ₄	1 hour to boil	1.5	3.3	49.5	28	-
30% H ₂ SO ₄	1 " "	1.5	3.25	50	30	-
2% NaOH	1 " "	0.8	3.2	50	27	Yellow
30% NaOH	1 " "	0.8	2.8	50	30	Yellow





LEAVII. fibre cross-section



LEAVII. fibre surface

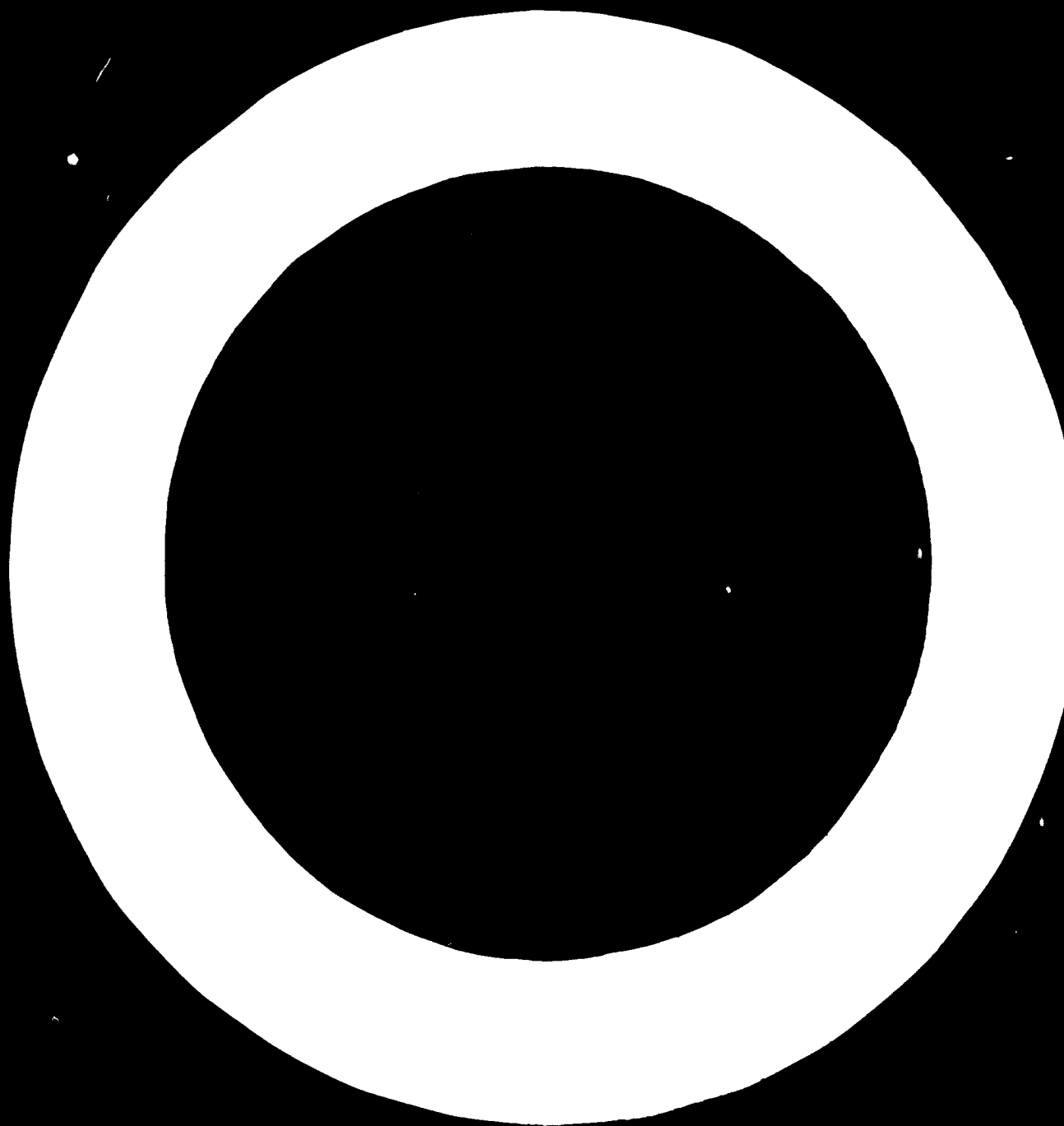
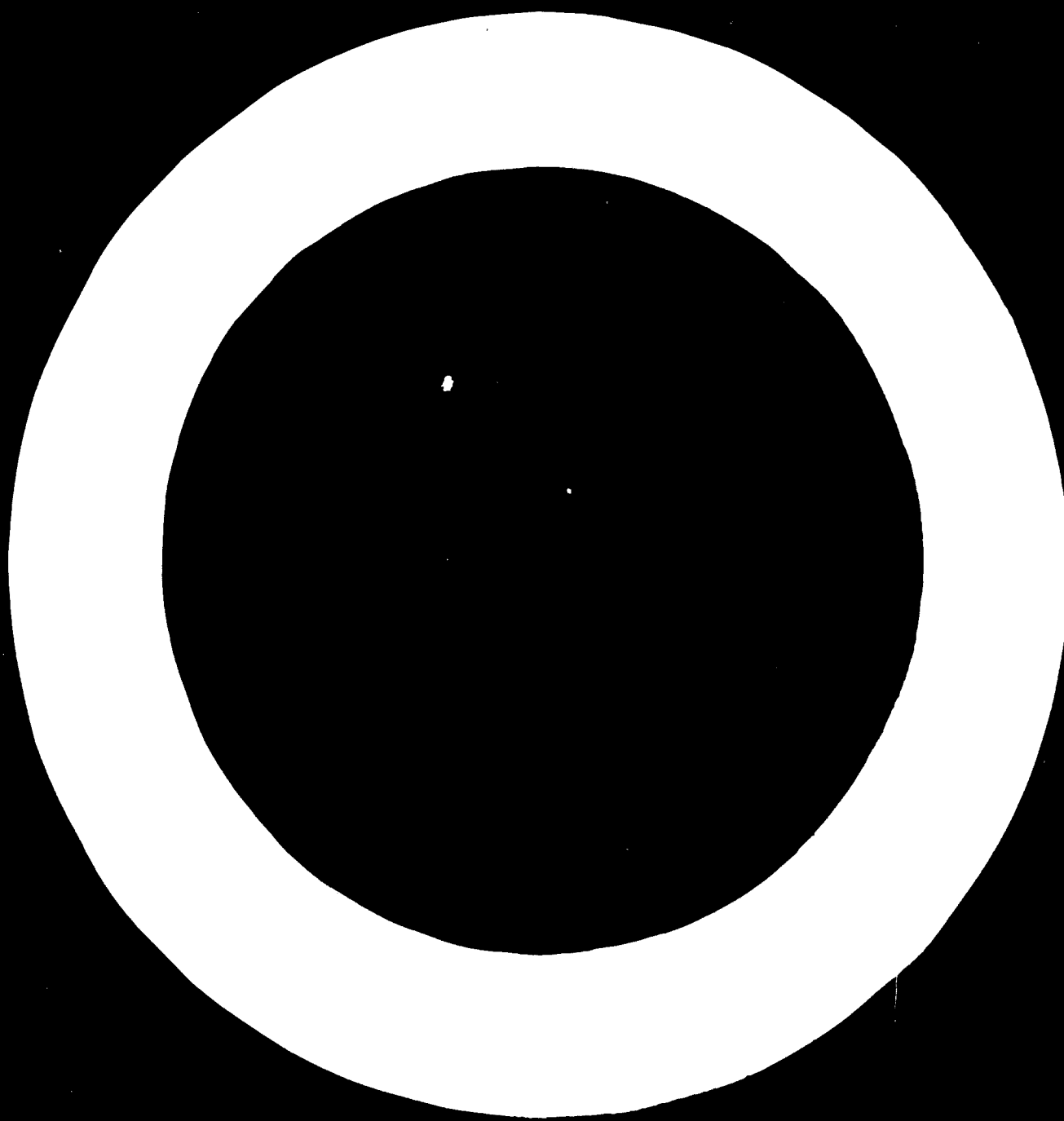


Fig. 7 - LENVIN fibre properties (commercial deniers)

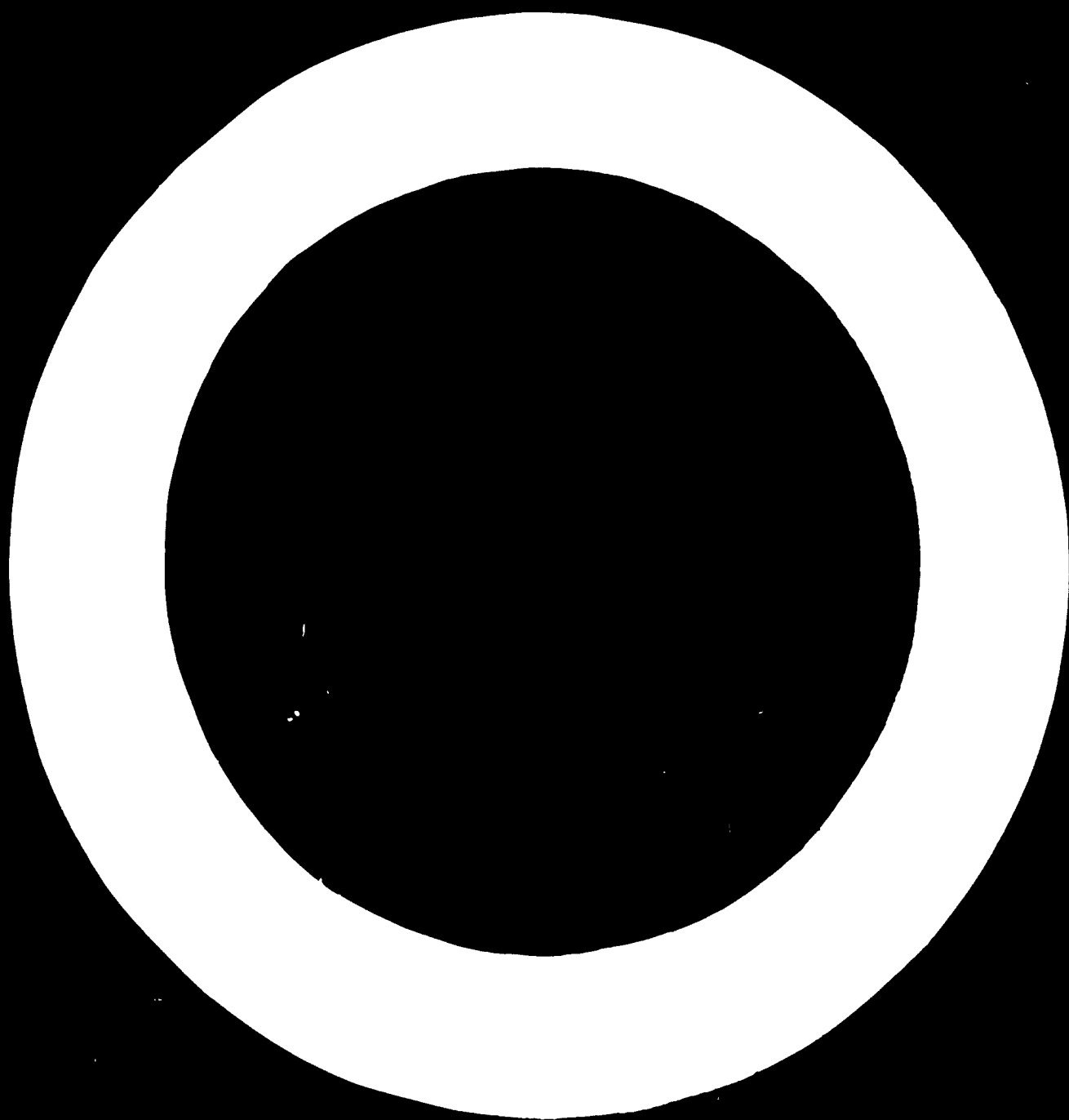
Denier	Tenacity g/den	Elongation %	Elastic Modulus g/den	Yield Point g/ten	Yield Tenacity g/den	Stress No.
1.5	3.30	36	30	0.60	2.8	85
2.0	3.20	38	29	0.55	2.7	80
3.0	3.10	40	28	0.55	2.5	75
5.0	2.9	46	27	0.50	2.1	70
8.0	2.8	50	27	0.50	1.9	70
15.0	2.6	60	27	0.50	1.7	65
25.0	2.3	70	24	0.50	1.4	60



34/55

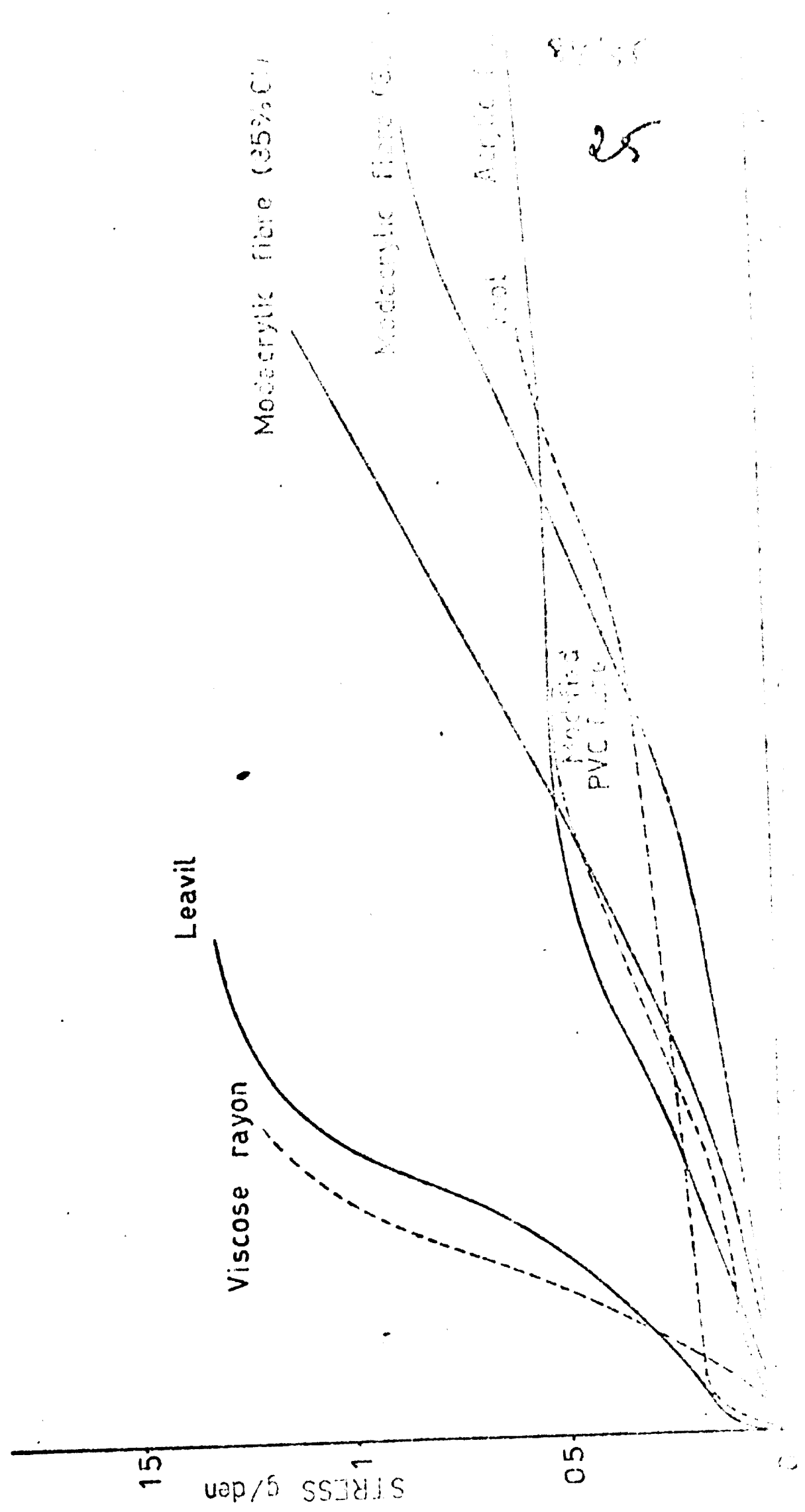
Fig. 8 - Properties of LEAVIL and other fibres under standard conditions and in water.

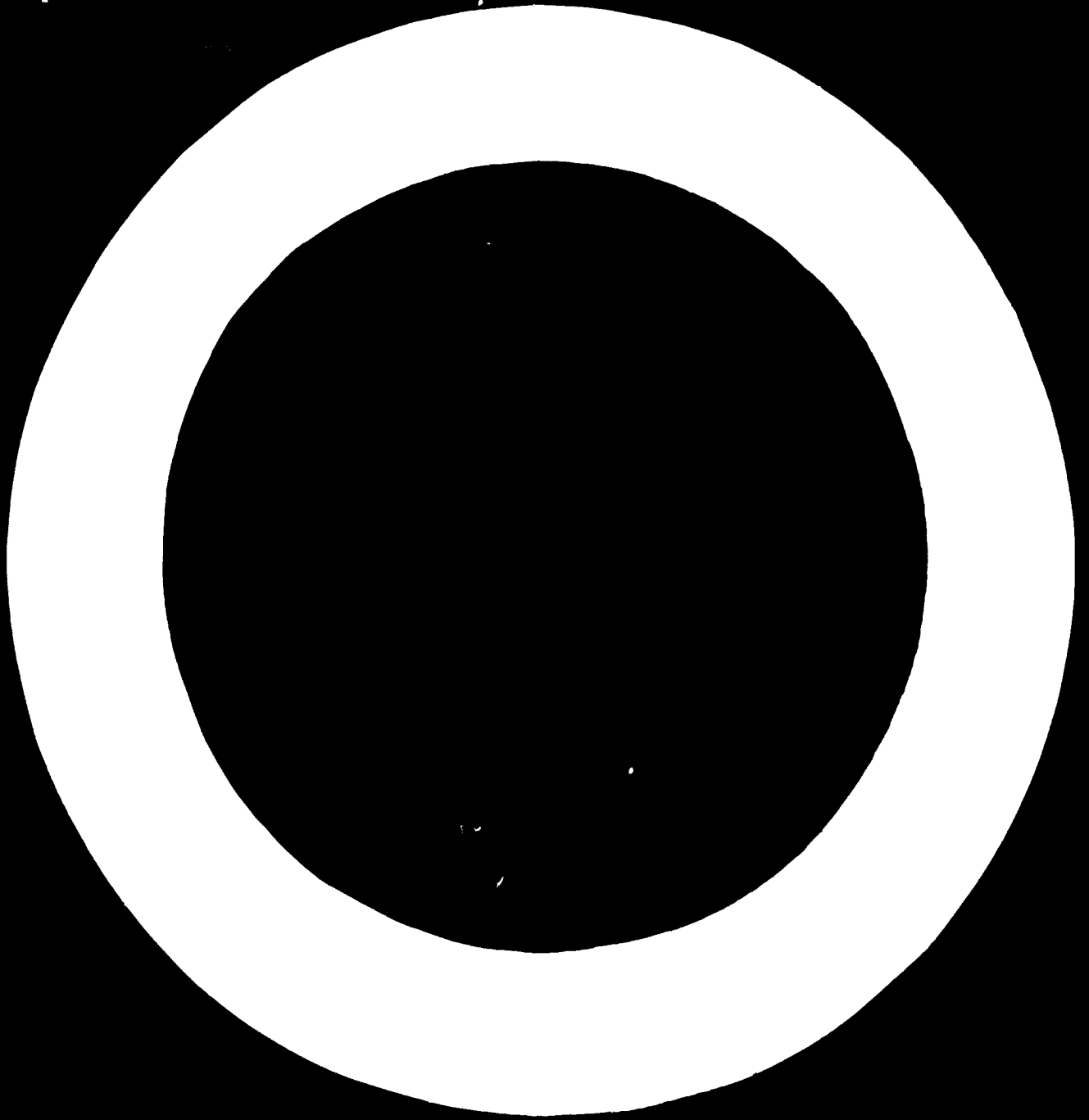
FIBRE	21°C, 65% R.H.		21°C in water		90°C in
	Tenacity g/den	Elong. %	Tenacity g/den	Elong. %	Tenacity g/den
Leavil	3.15	41	3.1	45	1.25
Traditional IVC fibre	2.5	24	2.3	27	0.8
Pre-shrunk IVC fibre	1.7	105	1.6	105	0.7
Modified IVC fibre	1.7	72	1.6	74	0.5
Modacrylic fibre (35% chlorine)	2.5	50	2.3	60	1.1
Modacrylic fibre (30% chlorine)	2.9	44	2.3	47	0.8
Acrylic fibre	2.4	43	1.8	46	0.6
Viscose rayon	2.5	18	1.4	24	1.2
Wool	1.3	39	1.2	57	0.5



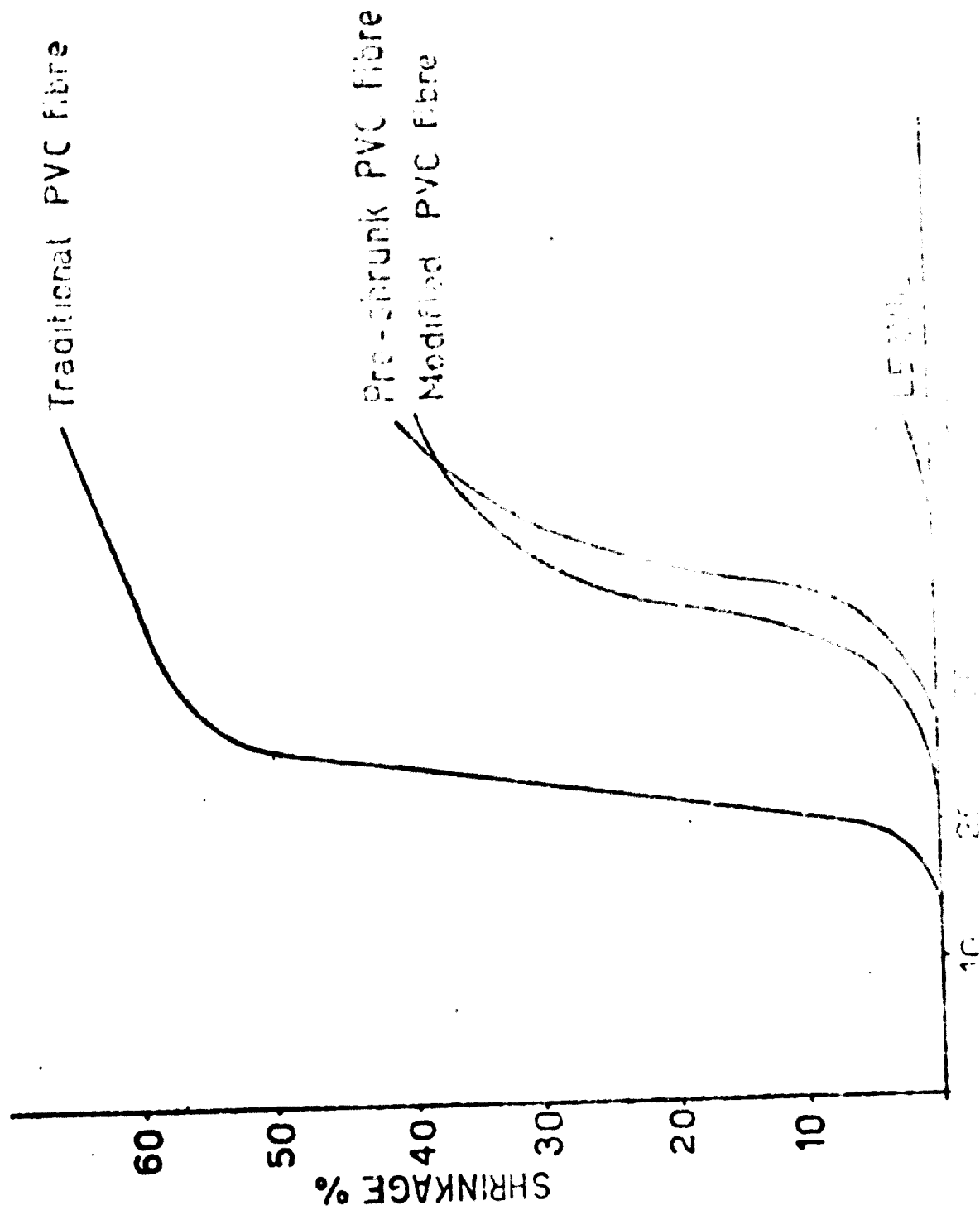
STRESS STRAIN CURVES IN WATER AT 20°C

FIG. 9





DIMENSIONAL STABILITY OF PVC FIBRES IN TRICHLOROETHYLENE FIG.10



34/55

25

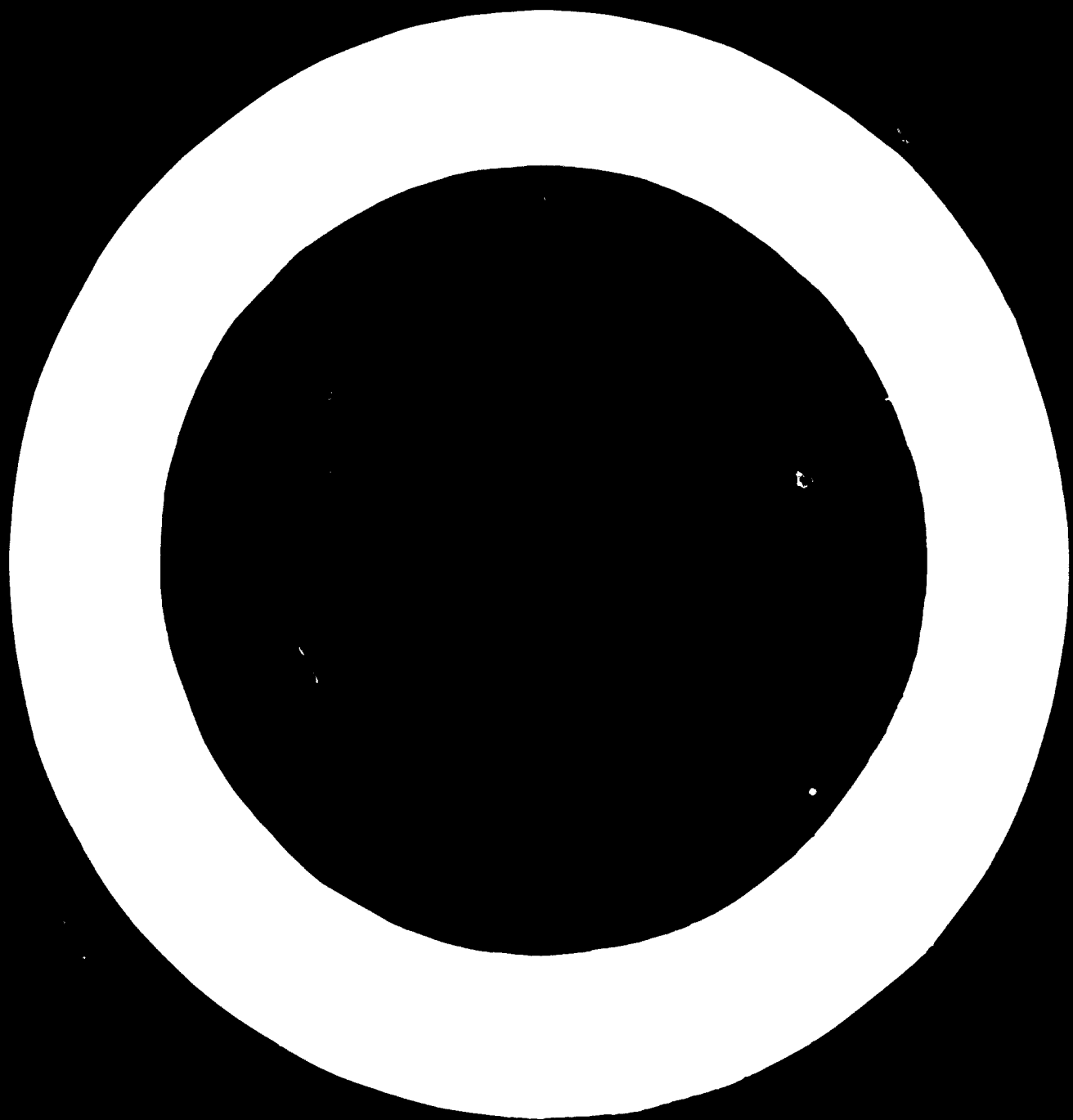


Fig. 11. III. II. receipt no. 101454

Material	Weight (g)	Volume (ml)	Concentration (%)
Acrylic acid	10	10	10
Styrene	10	10	10
	20	20	20
Diethyl malonate	10	10	10
Hexamethylenediamine	10	10	10
	10	10	10
Water	10	10	10
Triethylamine	10	10	10
	10	10	10
Acrylic acid	10	10	10
	10	10	10

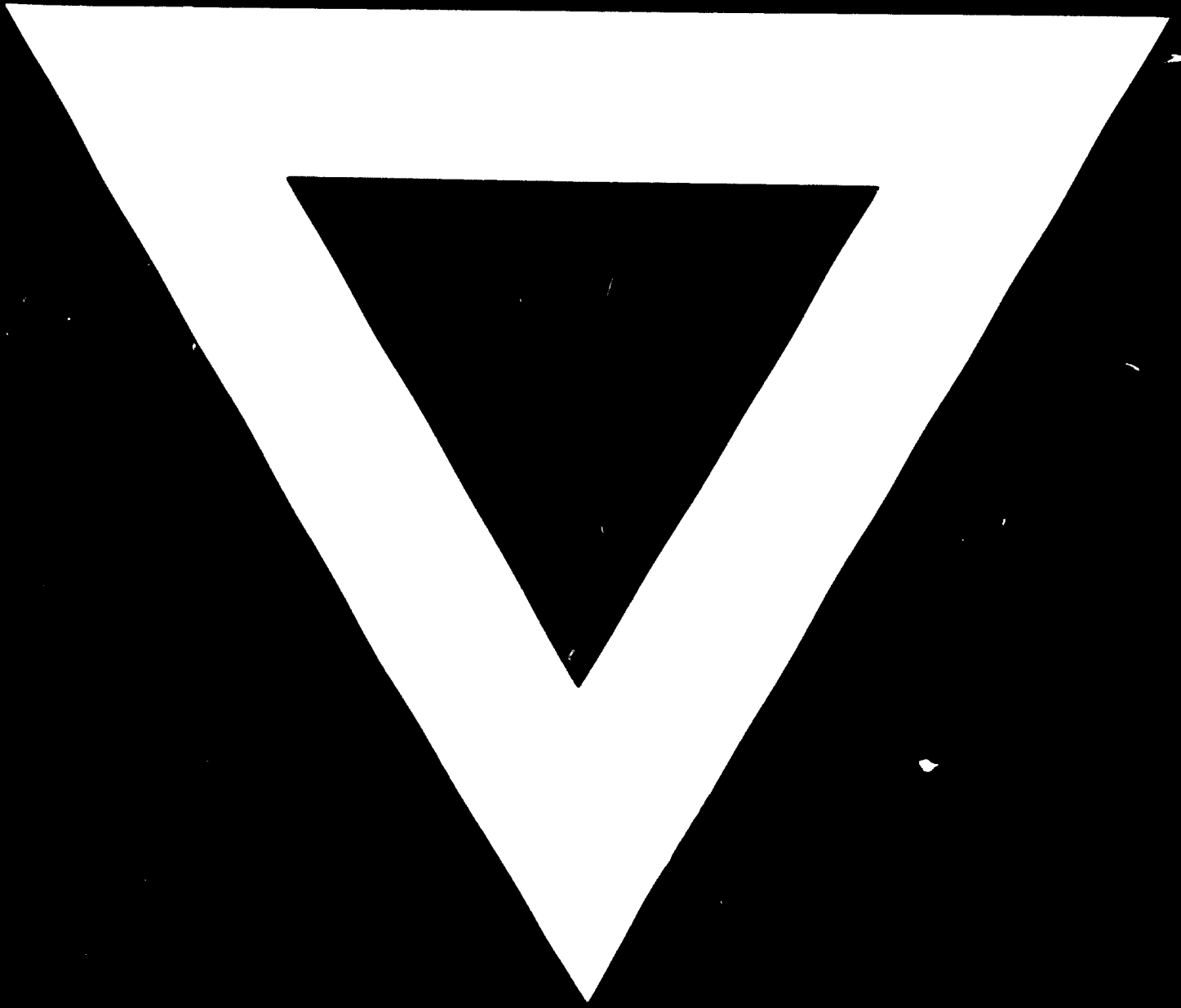
III. III. Other film schedules - specific solvents

Film	Material	Concentration (%)	Notes
III. III	Acrylic acid	10	Swells
	Styrene	10	Swells
	Diethyl malonate	10	Swells
Traditional PMMA film	PMMA	100	Swells
Pre-stretch PMMA film	PMMA	100	Swells
Modified PMMA film	PMMA	100	Swells
Polystyrene film (30% DI)	Polystyrene	30	Swells
Polystyrene film (50% DI)	Polystyrene	50	Swells
Acrylic films	Acrylic acid	100	Swells

Fig. 12 - Dispersed dyes selected for dyeing UPMIL fibre

Acetaminon Light Yellow 37	(FRANCOLOR)	Resolin Red 33	(BAYCO)
Artisil Yellow 200 Peron 22-201	(SANDOZ)	Resolin Scarlet 301	(BAYCO)
Artisil Brilliant Yellow 201 Peron 27-6001	(SANDOZ)	Tersotile Redine 71	(AGNA)
Artisil Yellow 21 Peron 27-71	(SANDOZ)	Artisil Violet 200 Peron Brilliant 7-2001	(SANDOZ)
Palanil Yellow 30	(BASF)	Palanil Brilliant Violet 4701	(BASF)
Palanil Yellow 37	(BASF)	Samaron Violet 4001	(HOECHST)
Palanil Blue Yellow 20	(BASF)	Resolin Blue-Violet 201	(BAYCO)
Samaron Yellow 4001	(HOECHST)	Peron Brilliant Violet 21	(SANDOZ)
Samaron Brilliant Yellow 401	(HOECHST)	Artisil Violet 71	(SANDOZ)
Resolin Yellow 20	(BAYER)	Tersotil Violet 71	(AGNA)
Resolin Yellow 21	(BAYER)	Esterequinon Light Blue 31	(FRANCOLOR)
Resolin Yellow 21	(BAYER)	Peron Blue 2-11	(SANDOZ)
Spectyl Yellow P-501	(GHEG)	Palanil Brilliant Blue 301	(BASF)
Microsetile Yellow 37	(AGNA)	Palanil Blue 7	(BASF)
Samaron Blue Yellow 20	(GHEG)	Palanil Blue 701	(BASF)
Samaron Brilliant Orange 201	(HOECHST)	Samaron Blue 401	(HOECHST)
Nibcol Orange 40	(GIBA)	Samaron Blue 200	(HOECHST)
Tersotil Light Orange 301 Extra	(FRANCOLOR)	Peron Dark Blue 7-201	(SANDOZ)
Tersotil Light Orange 301 Extra	(FRANCOLOR)	Peron Dark Blue 7-201	(SANDOZ)
Palanil Orange 3	(BASF)	Resolin Blue 201	(BAYCO)
Samaron Orange 200	(HOECHST)	Resolin Blue 201	(BAYCO)
Samaron Orange 20	(HOECHST)	Spectyl Blue P-50	(GHEG)
Samaron Brilliant Orange 200	(HOECHST)	Spectyl Blue P-5 conc.	(GHEG)
Acetaminon Light Pink 111	(FRANCOLOR)	Microsetile Blue 20	(AGNA)
Palanil Brilliant Pink 201	(BASF)	Peron Yellow-Brown S-201	(SANDOZ)
Samaron Pink 201	(HOECHST)	Palanil Yellow-Brown 201	(BASF)
Peron Brilliant Red 3-01	(SANDOZ)	Tersotile Yellow-Brown 201	(AGNA)
Palanil Brilliant Red 301	(BASF)	Tersotile Brown 301	(AGNA)
Palanil Red 37	(BASF)	Peron Brown S-301	(SANDOZ)
Peron Red 1-201	(SANDOZ)	Peron Brown S-201	(SANDOZ)
Peron Brilliant Red 2-201	(SANDOZ)	Peron Black S-201	(SANDOZ)
Resolin Brilliant Red 33	(BAYCO)	Peron Black S-201	(SANDOZ)
		Tersotile Black 21	(AGNA)





30 . 5. 72